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4 **Geochemical evidence for seabed fluid flow linked to the subsea permafrost**  
5 **outer border in the South Kara Sea.**  
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34  
35 **ABSTRACT**  
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37 Thawing subsea permafrost driven by rising bottom water temperatures leads to a drastic  
38 increase in fluid flow intensity in shallow marine sediments, and consequently emission of methane  
39 into the water column. Dissociation of permafrost-related gas hydrates, as well as mobilization of  
40 permafrost-sequestered methane into the global carbon cycling are of primary importance for future  
41 Arctic Ocean acidification. Previous studies in the South Kara Sea showed that abundant hydro-  
42 acoustic anomalies (gas flares) induced by seafloor gas discharge into the water column are limited to  
43 the water depths  $\geq 20$  m. Such distribution of gas flares may indicate an outer extent of continuous  
44 permafrost spatially restricting the upward fluid flow. This study focuses on geochemical analysis of  
45 the 1.1 m long sediment core, acquired offshore Yamal Peninsula (South Kara Sea) in the area of  
46 shallow fluid flow recorded in the high-resolution seismic data. Our results show narrow zone of  
47 Anaerobic Oxidation of Methane (AOM) with sharp and shallow sulfate-methane transition (SMT)  
48 located at 0.3 m subbottom depth and significant temporal variations in methane discharge confirmed  
49 by pyrite (FeS<sub>2</sub>) distribution. Concave up pore water chloride profile depicts an upward fresh/brackish  
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60 water advection in subsurface sediments. Terrestrial/fresh water genesis of interstitial methane from  
61 the sampled core is inferred from stable isotopic signatures ( $\delta^{13}\text{C}$  and  $\delta\text{D}$ ). We propose two possible  
62 mechanisms for the observed fluid flow: i) convection of thaw water from subsea permafrost, and/or  
63 ii) lateral sub-permafrost ground water discharge marking the outer limit of continuous permafrost  
64 extent offshore central Yamal Peninsula at ~45 m water depth.  
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69 **KEY WORDS:** Kara Sea, methane, permafrost, pore water, anaerobic oxidation of methane

## 70 INTRODUCTION

71  
72 Most existing field observations and numerical models on subsea permafrost focus on the East  
73 Siberian (Shakhova et al., 2010; 2015), Laptev (Koch et al., 2009; Overduin et al., 2015) and  
74 Beaufort Sea shelves (Ruppel, 2011; Brothers et al., 2012; Frederick and Buffet, 2015; Dimova et al.,  
75 2015). The Kara Sea shelf is significantly less studied, however existing data show that continuous  
76 permafrost does not extend beyond the 20 m isobaths offshore Yamal Peninsula (Rekant and  
77 Vasiliev, 2011; Portnov et al., 2013; Portnov et al., 2014; Serov et al., 2015). This has been  
78 evidenced from the distribution of gas-related hydro-acoustic anomalies limited to water depth  $\geq 20$   
79 m, which was hypothesized to mark the further extent of continuous permafrost spatially restricting  
80 upward gas migration. Offshore extent of subsea permafrost on the West Yamal Shelf is similar to  
81 the predicted Beaufort Sea shelf permafrost distribution (Brothers et al., 2012) suggesting similar  
82 scenarios of subsea permafrost evolution in these two regions.  
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90 Methane transport initiated by permafrost-induced fluid migration is of particular importance  
91 due to its contribution to atmospheric methane budget and its role in providing carbon source for  
92 local biogeochemical environment (McGuire et al., 2009). One of the important processes, which  
93 modulates methane transport through diagenetic zone is microbial-mediated anaerobic methane  
94 oxidation (AOM) coupled with sulfate reduction (SR) described by the following net reaction:  $\text{SO}_4^{2-}$   
95 +  $\text{CH}_4 = \text{HS}^- + \text{HCO}_3^- + \text{H}_2\text{O}$ . AOM is responsible for removal of interstitial methane from marine  
96 sediments, providing a geochemical barrier for methane transport into the water column and the  
97 atmosphere. In the diffusion-based systems, steady-state AOM implies the inverse downcore profiles  
98 of sulfate and methane clearly indicating the sulfate methane transition (SMT), a depth interval where  
99 the reactants are entirely consumed with 1:1 stoichiometry (Iversen and Jørgensen, 1985; Jørgensen,  
100 2006). On the contrary, elevated upward methane flux in advection-based systems shifts the SMT  
101 closer to the seabed indicating active methane seepage (Borowski et al., 1996; Coffin et al., 2013).  
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109 In this study, we present the results of geochemical analyses on the sediment core recovered  
110 within the area of focused fluid flow offshore Yamal Peninsula in the South Kara Sea (Fig. 1). We  
111 interpret the results of geochemical and biogeochemical analyses of near-surface sediment section  
112 affected by methane discharge in context of subsea permafrost degradation as the most significant  
113 prerequisite for fluid migration on the South Kara Sea shelf. We infer that the effect of permafrost on  
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119 subsurface fluid migration in our study area may be explained by two fundamental mechanisms: i)  
120 convection of thawing water transporting the syngeneic constituents and ii) migration from remote  
121 non-permafrost sources initiated by subsea permafrost degradation.  
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## 125 **Data and Methods**

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127 The data used in the current study were collected during a cruise conducted by All-Russia  
128 Institute for Geology and Mineral Resources of the World Ocean (“VNIIOkeangeologia”) in 2012 on  
129 board RV “Neotrazimy” in the framework of the state program “Perennial Comprehensive Seabed  
130 Geological Survey of Coastal Yamal Shelf”. During the cruise, we acquired 127 bottom coring  
131 stations, including station T-04, which was chosen for detailed analysis (Fig.1a).  
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### 134 **High-resolution seismic (HRS/chirp)**

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136 HRS/Chirp data were acquired with EdgeTech 3100 SB-216S sub-bottom chirp profiler.  
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138 Frequency range of 2-14 kHz, 20 ms pulse duration and 0.25 sec pulse range provided ~0.5-1 m  
139 vertical resolution and at best 20 m penetration. Swell filter and automatic gain control were applied  
140 to raw chirp data, allowing for better visualization of the upper sediment section.  
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### 143 **Sediment sampling**

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145 Bottom sampling station T-04 (71° 50' 56.7579" N; 67° 11' 49.4381") is located offshore  
146 Yamal Peninsula in the South Kara Sea in the water depth of 45 m. Sediment core (110 cm long) was  
147 recovered using the conventional gravity corer with 127 mm diameter and 1.5 m length, containing  
148 the plastic insert. Subsea positioning of the gravity corer during the deployment was provided by the  
149 calibrated hydroacoustic system HiPAP 350 P-5 (Kongsberg) with ultra short baseline configuration  
150 (USBL) providing accuracy  $\pm 0.5$  m.  
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154 After obtaining the gravity corer on deck, the plastic insert containing the core was extracted  
155 from the pipe. In order to minimize the loss of interstitial gases and to protect the reduced porewater  
156 compounds from oxidation, the core was cut into two halves longitudinally. One part was  
157 immediately subsampled for geochemical analysis, while the other part was placed in the onboard  
158 geological lab for visual description. Geochemical sampling was performed from seven horizons: 5,  
159 10, 30, 45, 60, 90, 110 cm below the seafloor (cmbsf).  
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### 162 **Grain-size analysis**

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164 The grain size analysis was performed by a combination of sieve (sand and silt fraction) and  
165 pipette (settling of clay particles according to the Stokes law) methods (e.g. Krylov et al., 2005). The  
166 amount of sand (1-0.05 mm), silt (0.05-0.005 mm) and clay (<0.005 mm) fractions (in accordance  
167 with the classification by Rukhin (1961)) were estimated (Lapina, 1977).  
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### 170 **Mineralogical analysis**

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172 Aleurite fraction (0.05-0.1) was used for determination of sedimentary mineral composition.  
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174 Light and heavy minerals were separated by weighting in separatory funnels filled with  $\text{CHBr}_3$   
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( $\rho=2.84-2.86$ ). Filtered, dried and weighed samples were carefully smeared on slide in immersion oil so that 300-400 mineral grains became visible in Olympus BH-2 microscope. Light minerals were studied at magnification x200 using the immersion oil with refraction index 1.542-1.544. Heavy minerals were split in two parts and analyzed at the same or lower magnification using two types of immersion oil with refraction indices 1.700 -1.706 and 1.638-1.640. Mineral species were determined based on their morphology and optical properties (Sakharova and Cherkasov, 1970), specified minerals were counted visually and quantified in % from the total aleurite fraction (Lapina, 1977).

### **Sedimentary TOC (Total Organic Carbon)**

Sediment samples were frozen at -20 °C by means of ScanVac Coolsafe 1100 lyophilizer prior to onboard freeze-drying. Dried sediments were homogenized by Retch RM200 grinder. TOC content was measured using Shimadzu TOC-V CSN element analyzer equipped with Solid Sample Unit SSM-500A in an onshore lab.

### **Sedimentary gas extraction**

Interstitial gas samples were extracted with the degassing set SUOK-DG (Patent (19) RU (11) 2348931 (13) C1) including centrifugal pump, supersonic ejector and hollow stainless steel working volume. Pre-weighted sediment samples were loaded into the sampling chamber and completely re-suspended by degassed water under high pressure. Interstitial gas phase was separated from the homogenous sediment pulp by supersonic ejector and fed into the graded volume meter. After the volume measurements, isolated sedimentary gas was transferred into the crimped vials with butyl-rubber stoppers (Portnov et al., 2018).

### **Interstitial methane and C<sub>2</sub>-C<sub>5</sub> hydrocarbon (HC) gases**

Hydrocarbon (HC) gas composition (C<sub>1</sub>-C<sub>5</sub>) including 8 compounds (Table 1) was analyzed using Shimadzu 2014 gas chromatograph (GC) equipped with flame ionization detector (FID) and Restek Rt-Alumina BOND/Na<sub>2</sub>SO<sub>4</sub> wide-bore capillary column (i.d. 0.53 mm, length 50 m, film thickness 10 µm) attached to the packed injector. Helium was used as a carrier gas at a flow rate of 25 ml/min. Certified gas mixtures were used as external standards. The detection limit of the analysis was 50 ppb, for CH<sub>4</sub> the error of instrumental measurements did not exceed 5%.

### **Sedimentary n-Alkanes and isoprenoids (C<sub>10</sub>-C<sub>40</sub>)**

For determination of n-alkanes and isoprenoids (phytane and pristane) sediment samples were freeze-dried onboard, grinded and stored dry until the on-shore lab analysis. Hexane extraction of the total liquid hydrocarbons (HC) was performed using Soxterm SOX 16 system (Gerhardt). Extracts were rotary evaporated, the resulting dry residue re-suspended in 2 ml of hexane and transferred into a 2 ml vial with PTFE screw caps. GC-FID analysis was carried out on Shimadzu GC 2014 using Machinery Nagel Optima-1 capillary column (0.25 mm, 30 m) and helium as a carrier gas (1.5 ml/min flow rate). Samples were injected using Shimadzu AOS 20i autoinjector. Splitless mode of

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238 GC evaporator retained during 1 min after withdrawal of a syringe needle and split vent re-opened to  
239 provide a ratio 1:20. The GC oven was programmed from 100°C to 320°C at 6 °C/min. Additional  
240 flow of carrier gas (make-up) was supplied to FID at a rate 25 ml/min. Certified mixture of odd and  
241 even C<sub>10</sub>-C<sub>40</sub> n-alkanes and isoprenoids (Chiron) was used as external standard for quantification,  
242 aliquote of squalane was spiked into sediment sample prior to extraction for evaluation of HC  
243 recovery during sample preparation. The analytical precision estimated on multiple injections was  
244 within a range of 2-8%.  
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#### 249 **Pore water sulfate and chloride**

251 Sedimentary pore water was extracted by wet sediment centrifugation 6240 G using Haereus  
252 Megafuge 1 at in 50 ml plastic centrifugal tubes flushed with argon stream to minimize the pore  
253 water sulfide re-oxidation. Pore water samples represented by supernatants were filtered through a  
254 0.25 um glass microfiber syringe filter (GF/F) and transferred into a 2 ml Eppendorf tubes allowing  
255 no headspace and stored frozen until the onshore instrumental measurements.  
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257 Anions concentrations were measured using the Metrohm 940 Professional IC Vario ion  
258 chromatograph equipped with conductometry detector, MSM-A rotor suppressor unit and Metrosepp  
259 A Sup 5 -150/4.0 column. Samples were eluted by 5 mmol Na<sub>2</sub>CO<sub>3</sub>/NaHCO<sub>3</sub> solution at 1 ml/min  
260 flow rate. Prior to analysis, pore water samples were diluted 1:50 to reduce the seawater chloride  
261 level overloading the column. The minimum detection limit of the applied method was 10 ppb (for  
262 both anions), while the error did not exceed 4 %.  
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#### 268 **Total Alkalinity (Talk)**

269 Pore water samples were filtered through a 0.25 um glass syringe microfiber filter (GF/F) into  
270 10 ml borosilicate vials. Total Alkalinity (Talk) was determined by titration of pore water samples  
271 against H<sub>2</sub>SO<sub>4</sub> with indication by phenolphthalein and methyl orange at pH 8.3 and 4.4 respectively.  
272 pH was measured by Hanna HI 83141 pH-meter.  
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#### 276 **Stable isotopic composition of carbon and hydrogen in methane.**

277 The isotope measurements were performed by isotopic-ratio gas chromatography/mass  
278 spectrometry (GC/IRMS). The minimum methane concentrations required for measurements were 20  
279 ppmV and 200 ppmV for determination of <sup>13</sup>C/<sup>12</sup>C and <sup>2</sup>H/<sup>1</sup>H respectively.  
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281 Carbon isotope composition (<sup>13</sup>C/<sup>12</sup>C) in CH<sub>4</sub> was measured with an Agilent 6890N GC  
282 (Agilent Technologies, Santa Clara, US) interfaced to a Finnigan Delta S IRMS (Bremen, Germany)  
283 using a Finnigan GC-C II interface. The GC was equipped with Molsieve column (12 m, 0.32 mm  
284 i.d.) and an injection valve. Samples were calibrated against a certified standard; the resulting  
285 isotopic signatures were reported in δ notation and per mil (‰) values vs. VPDB (δ<sup>13</sup>C VPDB).  
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287 The hydrogen isotope measurements (<sup>2</sup>H/<sup>1</sup>H) were performed on an Agilent 7890A GC  
288 (Agilent Technologies, Santa Clara, US) interfaced to a MAT 253 IRMS (Thermo Scientific,  
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296 Bremen, Germany) using a GC-Isolink interface by Thermo. The GC was equipped with Molsieve  
 297 column (12 m, 0.32 mm i.d.) and an injection valve. Samples were calibrated against a certified  
 298 standard; the resulting isotopic signatures were reported in  $\delta$  notation and per mil (‰) values vs.  
 300 VSMOW as ( $\delta D$  VSMOW)  
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## 302 Results

### 303 High-resolution seismic (HRS/chirp)

304 A sub-seabed seismic amplitude anomaly, which was chosen for sediment coring (Fig. 1b), is  
 305 a ~120 m wide amplitude blanking zone, extending upward from the subbottom depth of >25 m and  
 306 reaching the seafloor in the water depth of 45 m. We interpret this amplitude anomaly as a zone of  
 307 focused vertical gas discharge, indicating active fluid flow in this area.  
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### 309 Lithology and grain size composition.

310 The upper part of T-04 sediment core (5-25 cmbsf) is formed by black clayey silt with H<sub>2</sub>S  
 311 odor and insignificant admixture of shelly detritus (Fig. 1c). Downwards, in the interval 25-105  
 312 cmbsf, the core comprises a layer of dark-grey clayey silt with inclusions of hydrotroillite. Grain size  
 313 composition changes from 6.5% of sand, 54.0% of silt and 39.4% of clay in the upper 5 cm thick  
 314 horizon, to 59.9 % of sand, 20.0 % of silt and 19.9% of clay in the bottom of the core (110 cmbsf).  
 315 The strongest changes in proportion of the grain-size fractions characterize the interval between 60  
 316 and 110 cmbsf (Fig. 1c).  
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### 318 TOC

319 Maximum TOC content along the core is measured at 5 cmbsf (0.67% wt). In 5-45 cmbsf  
 320 interval, TOC profile demonstrates almost constant value (0.63-0.67%), while deeper in the section,  
 321 we observed gradual depletion reaching minimum value 0.2 % at 110 cmbsf (Fig. 2a).  
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### 323 Composition of n-alkanes and isoprenoids (C<sub>10</sub>-C<sub>40</sub>)

324 n-Alkanes and isoprenoids were analyzed in the sediment sample from the uppermost horizon  
 325 (5 cmbsf) containing recently deposited organic matter (OM). There is a unimodal distribution of the  
 326 measured compounds with conventional predominance of odd homologues C<sub>27</sub>-C<sub>31</sub> (Fig. 2c) derived  
 327 from terrigenous vascular plants waxes (Peters et al., 2008). Terrigenous constituent represented by  
 328 n-alkanes of C<sub>25</sub>-C<sub>34</sub> series is characterized by carbon preference index (CPI C<sub>25</sub>-C<sub>34</sub> =  
 329  $(C_{25}+C_{27}+C_{29}+C_{31}+C_{33}) / (C_{24}+C_{26}+C_{28}+C_{30}+C_{32}) + (C_{25}+C_{27}+C_{29}+C_{31}+C_{33}) / (C_{26}+C_{28}+C_{30}+C_{32}+C_{34}) = 5.05$ ) indicating  
 330 an odd overprint of recent OM and no signs of migrated petrogenic constituent (Abrams, 2005). Low  
 331 molecular weight fraction (C<sub>14</sub>-C<sub>34</sub>) associated with autochthonous organic matter in oligotrophic  
 332 conditions of the Kara Sea is mainly supplied to the bottom sediments by primary production (Stein  
 333 and Fahl, 2004; Vetrov and Romankevich, 2011). Terrigenous/aquatic index (TAR=  
 334  $C_{27}+C_{29}+C_{31} / C_{15}+C_{17}+C_{19} = 10.5$ ) demonstrates the major contribution from terrigenous source and minor  
 335 from in situ production (Bourbonniere and Meyers, 1996). Isoprenoid ratio (Pr/Ph=0.1) suggests  
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355 anaerobic or poorly oxygenated depositional environment (Peters et al., 2008).

### 356 **Methane and C<sub>2</sub>-C<sub>5</sub> hydrocarbon gases**

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359 The lowermost sampling horizon (110 cmbsf) shows maximum methane concentration  
360 reaching 734 μM. In contrast, minimum concentration (0.87 μM) is observed in the uppermost core  
361 section (5 cmbsf) (Table 1). The CH<sub>4</sub> concentration plot exhibits sufficient variations along the  
362 sediment section (Fig. 3). The steepest gradient is observed in the interval 30-10 cmbsf. Despite the  
363 strong H<sub>2</sub>S odor, visual analysis of sediment core T-04 revealed no gas-induced cracks, voids or other  
364 features of interstitial gas oversaturation and ebullition. Maximum measured concentration/saturation  
365 of interstitial methane (0.734 mM) is twice lower than the saturation limit for dissolved methane at  
366 atmospheric pressure (~1.3 mM) (Yamamoto et al., 1976; Jørgensen et al., 2001).

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369 The relatively insignificant percentage of C<sub>2+</sub> compounds in the total HC gas composition  
370 (e.g. 0.03% at 110 cmbsf) suggests microbial origin of methane (Whiticar 1999; Abrams, 2005). The  
371 ratio of methane concentration to the total content of gaseous hydrocarbons (C<sub>1</sub>/ΣC<sub>2</sub>-C<sub>5</sub>) shows  
372 significant upward decrease from 3452 at 30 cmbsf to only 22 at 5 cmbsf (Fig. 3, Table 1).

### 373 **Pore water sulfate and chloride**

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375 Sulfate distribution shows significant decrease with depth from 25.1 mM at 30 cmbsf to 1.38  
376 mM at 110 cmbsf (Fig.3). The corresponding downcore profile is of concave up shape (Fig. 3).

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378 Chloride profile reveals the downcore depletion from 536.3 (5 cmbsf) to 121.87 (110 cmbsf)  
379 (Fig. 3).

### 380 **Total alkalinity (Talk)**

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382 The lowest Talk value (~7 mM) was measured in the uppermost interval 5-10 cm (Fig. 3).  
383 Below, it is characterized by a sharp downcore increase to 15.44 mM within 10-30 cmbsf, followed  
384 by a decrease from 12.27 to 10.75 mM within 45-110 cmbsf interval.

### 385 **Sedimentary mineral composition**

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387 Content of authigenic minerals, including carbonates and pyrite, which formed as a result of  
388 the net sulfate reduction, is of a particular interest in the current study. Among carbonates we were  
389 only able to detect trace amounts of siderite (< 0,2%) with uniform distribution within the sampled  
390 sediment section.

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392 Downcore pyrite concentration/content shows drastic increase within the 30-45 cmbsf  
393 interval. The highest concentration of 55.4 % (aleurite fraction) was measured at 30 cmbsf (Fig. 3).  
394 Above and below this interval the values of pyrite are within the background range of 0.8-5.4 %.  
395 Light microscopy showed that framboid made up the obviously predominant textural form of pyrite  
396 in the samples from 30 and 40 cmbsf enriched in this mineral. Morphological analysis indicated the  
397 primarily authigenic nature of pyrite in the sediments of T-04 core .

### 400 **Stable isotope composition of carbon and hydrogen in methane.**

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414 Carbon isotopic signatures ( $\delta^{13}\text{C}$ ) of  $\text{CH}_4$  range from -77.8 to -79.9 ‰ VPDB (Fig. 3).  
415 Hydrogen isotopic signatures ( $\delta\text{D}$ ) vary from -296 to -281 ‰ VSMOW.  $\delta^{13}\text{C}$  and  $\delta\text{D}$  values remain  
416 almost constant along the sediment core (Table 1). Methane content in the gas samples from the  
417 uppermost horizons (5 and 10 cmbsf) was not sufficient for the isotopic measurements.  
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## 421 **Discussion**

### 422 **Sources of the organic matter and their significance for early diagenesis**

423 Sediments of the West Yamal coast and adjacent shallow shelf undergo intensive coastal  
424 erosion, which results in the delivery of allochthonous OM (Stein and Fahl, 2004). OM enriched  
425 with terrigenous compounds is refractory to biochemical reactions and mostly preserved during  
426 early diagenesis (Stein and Fahl, 2004; Vetrov and Romankevich, 2011). In turn, contribution of  
427 fresh and reactive autochthonous OM, linked to the aquatic primary production is relatively low due  
428 to low productivity of cold and shallow waters of South Kara Sea. This portion of OM is  
429 presumably consumed by aerobic oxidation within the water column and at the seafloor. Thus, only  
430 few percent of the initial amount is left for anaerobic diagenesis (Vetrov and Romankevich, 2011;  
431 Rozanov, 2015). TOC content of 0.68% wt measured beneath the oxidized layer (5 cmbsf) in core  
432 T-04 is a typical regional background value for the Kara Sea shelf (Stein and Fahl, 2004; Seiter et  
433 al., 2014; Rozanov, 2015). However, the downward decrease in the sedimentary TOC is likely  
434 attributed to the lithological control rather than the OM mineralization. This becomes evident from  
435 the correlation between the TOC content and percentage of clay fraction (Fig 2b) retaining OM due  
436 to its high sorption capacity (Tyson, 1995; Secrieru and Oaie, 2009). Such mechanism causes a  
437 consistent lithologically controlled partitioning of sedimentary TOC and its downcore variations.  
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449 Results of our analysis of sedimentary lipid biomarkers (n-alkanes and isoprenoids) in the  
450 uppermost sampling interval (5 cmbsf) show significance of OM sources for the studied depositional  
451 environment. We observe prevalence of terrigenous OM indicated by unimodal distribution pattern  
452 of n-alkanes with domination of  $\text{C}_{29}$ ,  $\text{C}_{31}$  and  $\text{C}_{33}$  odd homologues. Contribution from  $\text{C}_{14}$ - $\text{C}_{25}$   
453 compounds representing the liable autochthonous OM in our samples is low to moderate (Fig 2c)  
454 (Meister et al., 2013; Vetrov and Romankevich, 2011). The drastic predominance of terrigenous  
455 organic matter does not provide favorable conditions for intensive organoclastic sulfate reduction  
456 (OSR) in the T-04 upper anoxic sediments, which may cause the detected subsurface sulfate  
457 depletion within 110 cmbsf (Jørgensen et al., 2001; Jørgensen, 2006; Meister et al., 2013).  
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### 463 **Methane, sulfate and AOM**

464 Results of our analyses from 126 coring stations show the average value of sedimentary  
465 methane concentration 4.58  $\mu\text{M}$  in the interval 85-120 cmbsf (n=126). This concentration may be  
466 used as a reliable background value for T-04 core, because the methodology of gas measurements  
467 was identical at all sampling locations. Measured maximum of methane concentration at the  
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474 corresponding depth of 110 cmbsf in T-04 core (734.2  $\mu\text{M}$ ) is 158 times higher than the background  
475 value, indicating focused methane-rich fluid flow in our study area.  
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477 Methane  $\delta^{13}\text{C}$  values in T-04 core range from -77.8 to -79.9 ‰ VPDB suggesting its microbial  
478 genesis. Additionally, gas composition in the samples with high methane content (i.e. less altered by  
479 the AOM) shows only trace input of heavy homologues, which is also typical for microbial source  
480 (Table 1) (Abrams, 2005). Yet, possible minor admixture of thermogenic gas should not be ruled out  
481 (Whiticar, 1999).  
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485 The inverse trends of sulfate and methane profiles (Fig. 3) indicate process of AOM in the  
486 bottom sediments (Iversen and Jørgensen, 1985). However, a concave-up sulfate profile in T-04 does  
487 not correspond with typical steady state AOM profile, characterized by the linear gradient of pore  
488 water sulfate, depicting the diffusive flux into the reaction zone (Iversen and Jørgensen, 1985;  
489 Borowski et al., 1996; Jørgensen et al., 2001). The subsurface concave up sulfate profiles are  
490 normally attributed to the OSR-dominated conditions, which is definitely not characteristic for T-04  
491 depositional environment. Therefore, it is evident that sulfate concentration in our study area is  
492 strongly affected by an external process. (Iversen and Jørgensen, 1985).  
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498 Hydrogen sulfide, produced by SR including both OSR and AOM, interacts with either  
499 dissolved iron or reactive solid-state iron oxides forming the precipitate of amorphous iron sulfide  
500 (FeS), which may successively convert to pyrite (FeS<sub>2</sub>). This mineral is stable enough to be preserved  
501 in the sediments for many millennia; therefore it presents a reliable proxy for paleo-environmental  
502 reconstructions (Sassen et al., 2004). Overproduction of pyrite is reported to mark the shallow SMT  
503 indicative of transitional diagenesis from lower to higher methane supply with corresponding  
504 redistribution of reactive compounds (Lin et al., 2016; Coffin et al., 2008). At the same time, there  
505 are examples of methane seepage settings in which pyrite precipitates above the thin zone of  
506 subsurface iron reduction, but not within the SMT. In this case, iron sulfide formation is restricted by  
507 spatial pattern of reactive Fe (II) concentration, as opposed to a more typical scenario when pyrite  
508 crystallization is restricted to the local discharge of H<sub>2</sub>S as result of AOM (Latour et al., 2018).  
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515 Downcore profile of the sedimentary pyrite (FeS<sub>2</sub>) in T-04, shows sufficient selective  
516 accumulation of this mineral within the depth interval 30-45 cmbsf (Fig. 3). Wherein, its maximum at  
517 30 cmbsf (55.4% of the net heavy minerals composition) appears to coincide with the maximum of  
518 pore water  $\text{Fe}^{2+}$  (15.44 mM), reflecting the release of the SR products (HCO<sub>3</sub><sup>2-</sup> and HS<sup>-</sup>). Given that  
519 AOM is a major driver of SR, this subbottom depth (30 cmbsf) indicates the position of the modern  
520 SMT. In this case, we define SMT by the depth of obviously enhanced AOM. The vertical extension  
521 of AOM zone above the SMT is confirmed directly by the highest methane depletion within the  
522 interval 10-30 cmbsf and indirectly - by conservative distribution of  $\delta^{13}\text{C}$  CH<sub>4</sub> below 30 cmbsf with  
523 no signs of AOM-linked isotopic fractionation (Whiticar, 1999). Unfortunately, lack of  $\delta^{13}\text{C}$  CH<sub>4</sub>  
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532 data in the subbottom depths < 30 cmbsf caused by insufficient methane concentration, does not  
533 allow tracing the expected oxidation shift above 30 cmbsf. At the same time, observed sulfate  
534 retaining below the established SMT (values don't decrease below 1.5 mM) indicates no basic  
535 conditions for methanogenesis. Elevated sulfate content below the SMT is often caused by natural  
536 reoxidation of sulfide under cryptic sulfur cycle operation (Holmkvist et al., 2011; Treude et al.,  
537 2014). However, in the studied core, extremely shallow SMT and the uneven boundary between  
538 sulfate and methane zone are more likely linked to pore water mixing as result of fresh water fluid  
539 flow.  
540

#### 541 **Chloride and pore water mixing**

542 Downcore distribution of biochemically inert pore water chloride is widely used for tracing  
543 the variations in pore water salinity indicative of conservative mixing due to fluid migration (Kastner  
544 et al., 2008; Pohlman et al, 2008). Chloride concentration in T-04 demonstrates 78% decrease (from  
545 536.36 to 121.87 mM) (Fig. 3). Concave up shape of chloride profile indicates advection-dominated  
546 flux with particular subsurface curvature defining the upper limit of a mixing zone at 10 cmbsf (Fig.  
547 3) (Schlüter et al., 2004). Discrimination between the effects of diagenetic reactions and conservative  
548 mixing on distribution of the involved pore water solutes are available from the mixing diagram  
549 (Pohlman et al., 2008) (Fig 4.). The observed linear correlation between sulfate and chloride reveals  
550 physical mixing as a major factor of sulfate depletion. Along with the significant mixing effect, the  
551 coinciding excursions of  $\text{SO}_4^{2-}$  (negative) and Talk (positive) obviously exhibit the effect of AOM.  
552 Absence of authigenic carbonates in the sediment core T-04 indicates that odd decrease in pore water  
553 Talk at 30 cmbsf is not driven by consumption during  $\text{CO}_2$  reduction, instead it probably reflects the  
554 dilution through conservative mixing. Thus, chloride distribution indicates upward advection of  
555 fresh/brackish water responsible for changing the pore water composition in diagenetic  
556 environments.  
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558 Considering the basic conditions for fluid transport within the studied sediment section, we  
559 cannot ignore the lithological control. The increase of sandy fraction in the lower part of the core  
560 indicates the increase in permeability basically required for pore water advection described by Darcy  
561 law (Schulz et al., 1994).  
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#### 563 **Stable isotope composition of methane and its possible sources**

564 The conventional CD diagram based on compilation of  $\delta^{13}\text{C}$  and  $\delta\text{D}$  values can be used for  
565 differentiation between the sources of interstitial methane (Whiticar, 1999). In our case, the isotopic  
566 signatures in T-04 methane samples are compared with two reference sites (Fig.1, Fig 5). First  
567 reference is from methane samples extracted from the ground ice onshore Yamal Peninsula, near  
568 Marre Sale (Streletskaia et al., 2018) (Fig. 1a, 5). Second is a marine reference constituted by  
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591 methane sampled from gas charged sediment at pingo-like feature site (PLF-2), probably linked to  
592 gas hydrate decomposition (Serov et al., 2015).  
593

594  
595 Isotopic signature of the ground-ice derived methane appears within a transition field of CD  
596 diagram apparently indicating coexistence of acetoclastic methanogenesis and CO<sub>2</sub> reduction typical  
597 for organic-rich humid landscapes of Arctic tundra (Lofton et al., 2015). In marine conditions, CO<sub>2</sub>  
598 reduction becomes the only possible pathway, since as competitive substrate acetate is depleted by  
599 SR before entering the zone of methane generation (Whiticar, 1999). Fig. 5 shows that similarly to  
600 positive reference values of methane samples from the Marre Sale ground ice, T-04 methane appears  
601 within a transitional field with slightly lower values of  $\delta^{13}\text{C}$  (Fig. 5; black dotted line). Being  
602 composed ultimately by the migrated pool, interstitial methane in the target site is likely to involve a  
603 remote terrestrial (fresh-water) source partly inheriting its characteristic isotope signatures. **In this  
604 case, <sup>2</sup>H discrimination resulted from a kinetic effect of acetoclastic methanogenesis, is coupled  
605 with utilization of <sup>2</sup>H depleted fresh water source for CO<sub>2</sub> reduction pathway. On the contrary, <sup>2</sup>H  
606 enriched methane from the PLF-2 site, generated exclusively by the CO<sub>2</sub> reduction in typical marine  
607 (i.e. sulfate rich) environment shows a corresponding position within CD-plot and provides a  
608 negative reference, indicative of different isotopic composition** (Fig. 5; blue dotted line) (Whiticar,  
609 1999).  
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#### 619 **Geochemical data synthesis with implication to the observed fluid migration**

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621 Our geochemical analyses indicate presence of dissolved methane flux and fresh/brackish  
622 water advection co-occurring within the sediment section. Thus, seismic amplitude blanking anomaly  
623 observed on the chirp data is caused by discharge of methane-saturated fresh water fluid. The  
624 apparent depth of the seismic amplitude blanking anomaly (> 25 m) indicates a relatively deep  
625 location of fluid source connected with the shallow sediment section by sub-vertically developed  
626 conduit (Fig 1b).  
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#### 630 **Possible mechanisms explaining the fluid flow system at T-04 site.**

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632 We consider two mechanisms, which may drive the advection-dominated fluid flow system at  
633 T-04 site offshore Yamal Peninsula: thawing subsea permafrost and terrestrial ground water  
634 discharge.  
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637 It has been shown that relict subsea permafrost offshore Yamal Peninsula is an extension of  
638 thick organic-rich permafrost complex developed across the western Yamal during ancient sea  
639 regressions (Vasiliev et al., 2015; Streletskaya et al., 2018). Therefore, thawing subsea permafrost  
640 may drive circulation of methane-charged brackish water initiated by convection between ambient  
641 saline water flows and less dense fresh/brackish water from thawing permafrost (salt fingering)  
642 (Osterkamp, 2001; Frederick and Buffet, 2015). Shallow drilling into subsea permafrost in the Laptev  
643 sea (Buor-Haya bay) (Overduin et al., 2015) showed that excess of dissolved methane released by  
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651 thawed permafrost was almost entirely consumed by the AOM immediately above the boundary  
652 between unfrozen and frozen sediment sections at the depth of approximately 25 m below the  
653 seafloor. The insufficient amount of residual methane above the reaction zone was remarkable for  
654 dramatic depletion in  $^{12}\text{C}$  (Overduin et al., 2015). It is likely that intense advection-dominated  
655 methane transport rate significantly overrides the biochemical consumption of dissolved methane in  
656 T-04 core. Additionally, the increased buoyancy of methane charged water should stimulate the  
657 upward fluid migration driven by density gradient (Park, 1990; Osterkamp, 2001). Similar  
658 observations of permafrost-related seabed water discharge are extremely scarce. One of them was  
659 detected in the Beaufort Sea, where brackish water discharge has been documented and attributed to  
660 seasonal thawing of subsea permafrost; however there was no observations of related methane  
661 transport (Dimova et al., 2015).  
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664 Alternative mechanism, which may drive the fluid flow at T-04 site, is the subbottom  
665 groundwater discharge (SGD). Lateral transportation of methane (as well as other solutes and  
666 nutrients) by ground water flows is a well-studied phenomenon (Bugna et al., 1996). Frederick and  
667 Buffet (2015) provided a comprehensive numerical model involving SGD related to subsea  
668 permafrost at the Beaufort Sea shelf. This model shows that sub-permafrost ground flows can move  
669 laterally seaward along the lower permafrost boundary towards its outer edge. There, fresh water  
670 flow changes its direction to subvertical providing the quasi-stationary mixing interface significantly  
671 further offshore compared to non-permafrost conditions, where the discharge distance is mainly  
672 determined by topography-driven hydraulic head (Frederick and Buffet, 2015). In this scenario,  
673 continuous permafrost offshore central Yamal Peninsula may reach 45 m isobaths which is  
674 significantly more extensive compared to the existing estimate (20 mbsl) (Portnov et al., 2013,  
675 Portnov, 2014).  
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678 The above scenarios describe two different mechanisms, driving the fluid flow system in the  
679 T-04 site offshore Yamal Peninsula. These scenarios are not mutually exclusive, as such, both  
680 mechanisms may co-exist, providing complex geochemical signature, which we registered in our  
681 sediment samples.  
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### 684 **Conclusions**

- 685 • High resolution seismic amplitude anomaly has been verified by high methane  
686 concentration indicating modern shallow fluid flow system offshore Yamal Peninsula.
- 687 • Non steady state diagenesis becomes evident from the distribution pattern of reactive  
688 compounds suggesting the physical process imposed on local biogeochemical  
689 environment. Our data indicate extremely narrow AOM zone with sharp and shallow  
690 SMT located at 30 cm subbottom depth with significant temporal variations in  
691 methane discharge recorded in pyrite ( $\text{FeS}_2$ ) distribution. Concave up pore water  
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709 chloride profile depicts an upward fresh/brackish water advection and the upper limit of  
710 subsurface mixing between 5 and 10 cmbsf.  
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- 712 • Terrestrial/fresh water origin of interstitial methane from the sampled location has  
713 been revealed by stable isotopic signatures ( $\delta^{13}\text{C}$  and  $\delta\text{D}$ ) similar to methane, extracted  
714 from ground ice of adjacent onshore setting used as a reference (Marre Sale site).  
715 Thus, the apparent explanation for the detected geochemical features of subsurface  
716 migration is pore water advection of fresh/brackish water sufficiently enriched in  
717 methane.  
718
- 719 • We propose two potential mechanisms driving the observed fluid flow: convection of  
720 sub-sea permafrost thaw water and sub-permafrost lateral ground water discharge. The  
721 latter may indicate the outer limit of continuous permafrost extent offshore central  
722 Yamal Peninsula at 45 mbsl isobath.

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## FIGURES CAPTIONS

Fig 1. (a) Location of the offshore sampling station T-04 (this study) and sampling sites at Marre Sale and PLF-2 (Streletskaya et al., 2018; Serov et al., 2015) provided onshore and offshore reference data for comparison of CH<sub>4</sub> stable isotopic compositions; (b) chirp section showing shallow amplitude anomaly in the chirp data; (c) Lithology and grain-size composition of T-04 sediment core.

Fig. 2. Organic matter (OM) of T-04 sediment core (a) TOC (%) downcore distribution; (b) TOC (%) plotted against clay content in the sediments; (c) distribution of n-alkanes ( $C_{10}$ - $C_{35}$ ) and isoprenoids (phytane and pristane) in hexane extract of the sediment sampled at 5 cmbsf. The ratio indices (CPI, TAR, Pr/Ph) are specified in the black frame.

Table 1. Interstitial hydrocarbon gases (C<sub>1</sub>-C<sub>4</sub>) in T-04 sediments (molecular composition and stable isotopic composition of methane ( $\delta^{13}\text{C}$  and  $\delta\text{D}$  ))

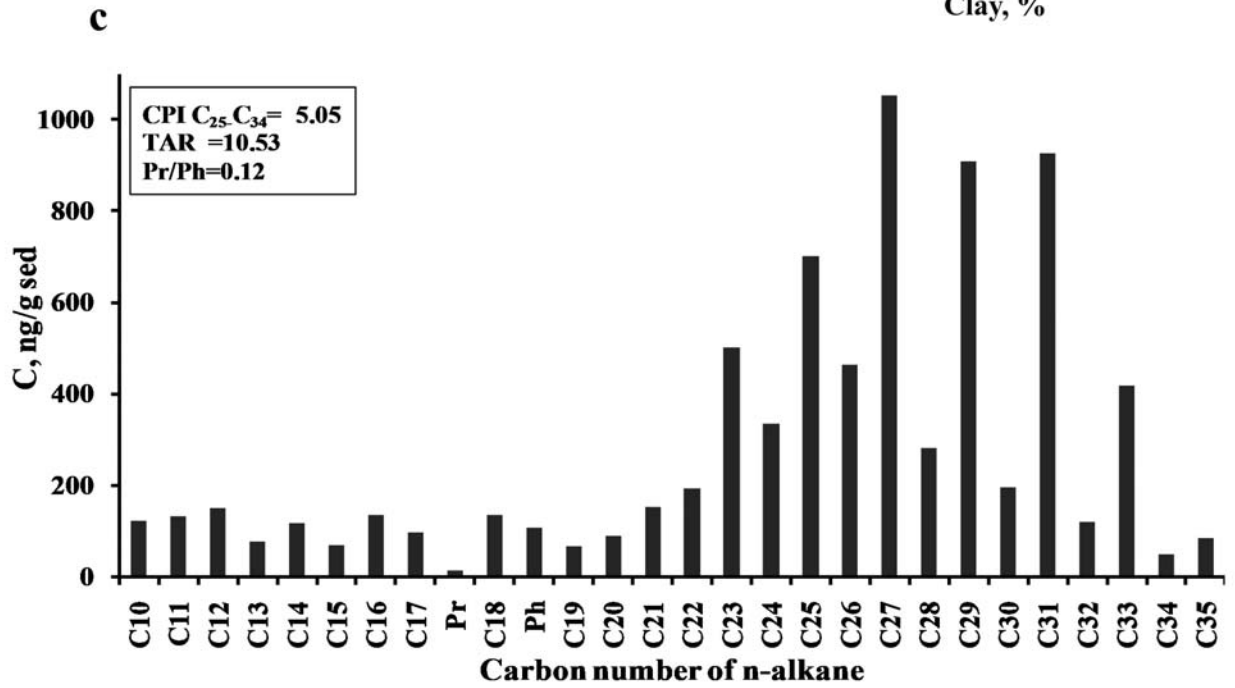
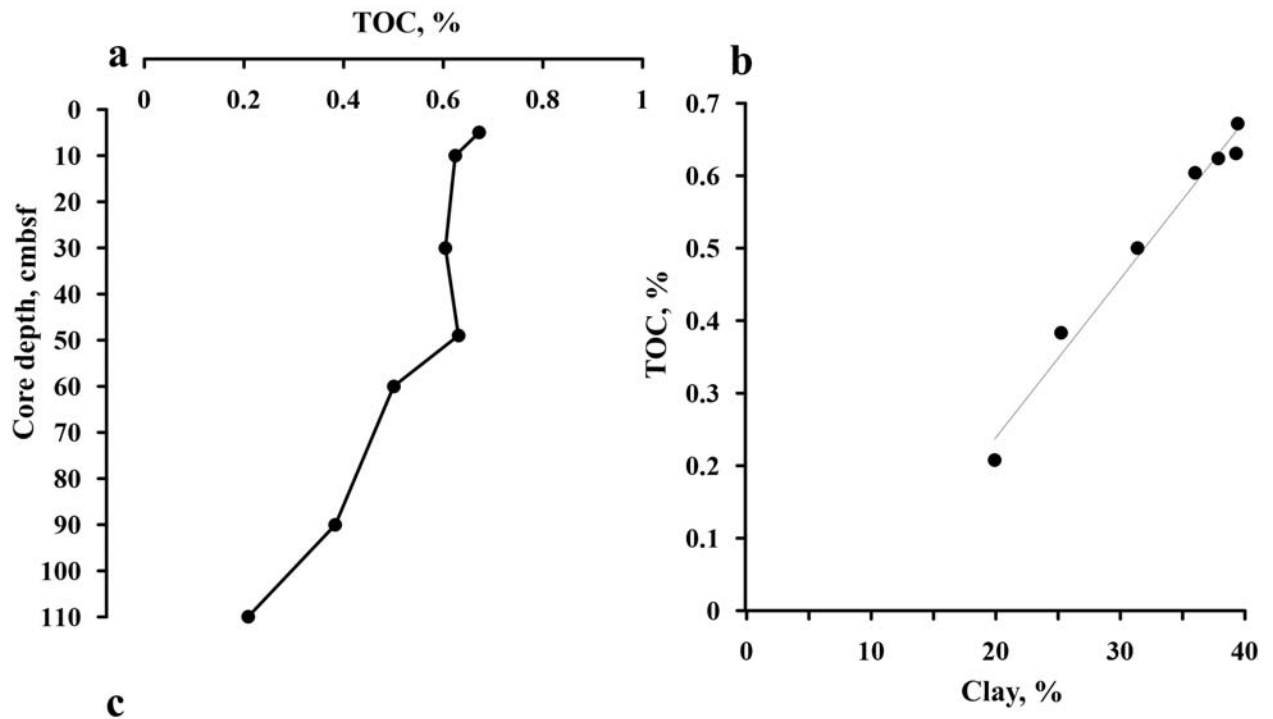
Fig. 3. Gas geochemistry and pore water chemistry in T-04 sediment core. Grey rectangle indicates a zone of anaerobic oxidation of methane (AOM), grey arrow points at sulfate-methane transition (SMT).

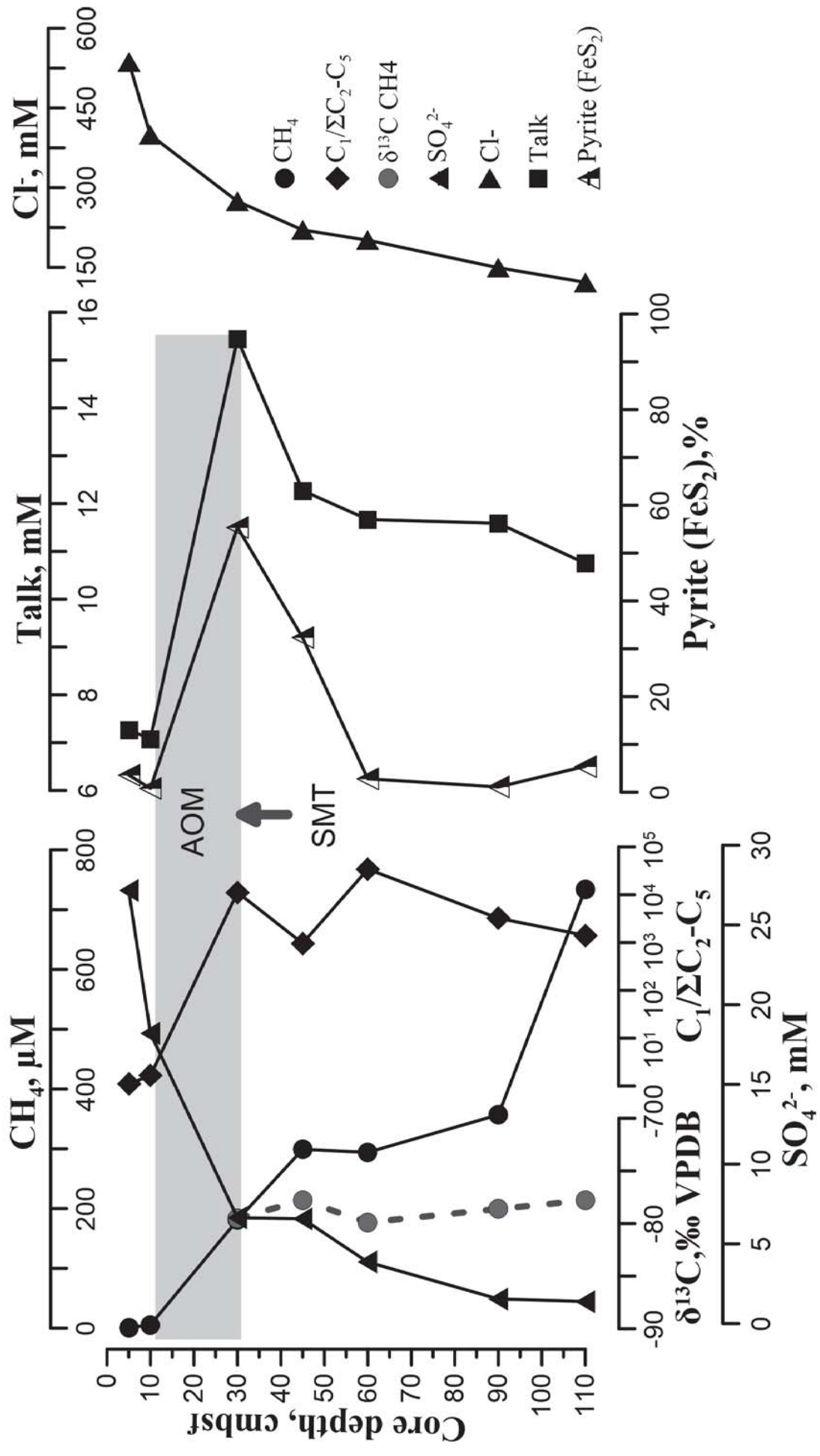
Fig. 4. Mixing diagram (after Pohlman et al., 2008), including sulfate and Talk (mM) plotted as functions of chloride distribution (mM). Coinciding negative excursion of sulfate and positive of Talk are outlined by grey ellipse marking AOM against a background of conservative mixing reflected in linear correlation of chloride and sulfate.

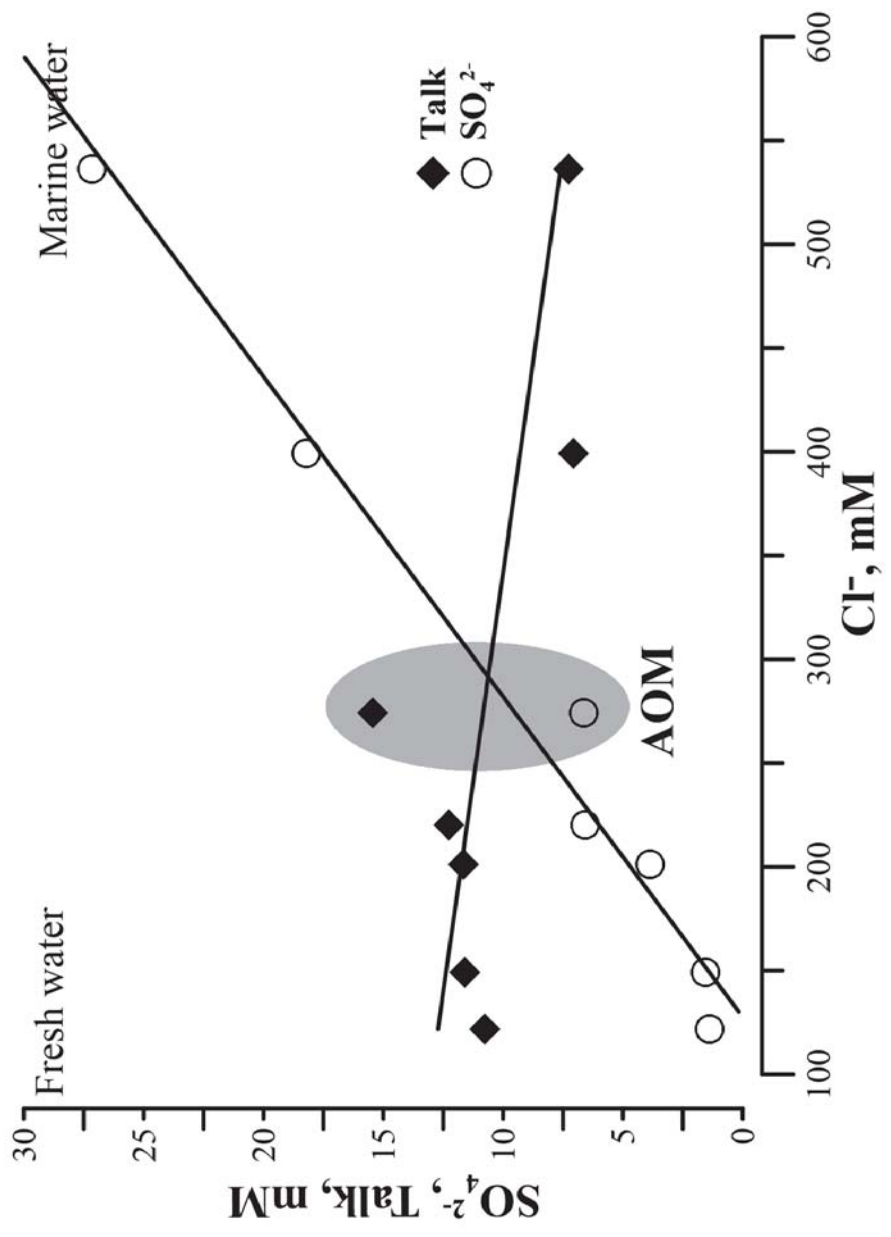
Fig. 5. CD diagram of interstitial methane (after Whiticar, 1999), including the data from T-04 core and published data (ground ice from Marre Sale site and PLF-2) for onshore and offshore reference respectively (Streletskaya et al., 2018; Serov et al., 2015). Brown dashed ellipse shows samples with methane of presumably terrestrial (onshore) origin, while blue dashed ellipse shows samples with methane of typical marine (offshore) genesis.

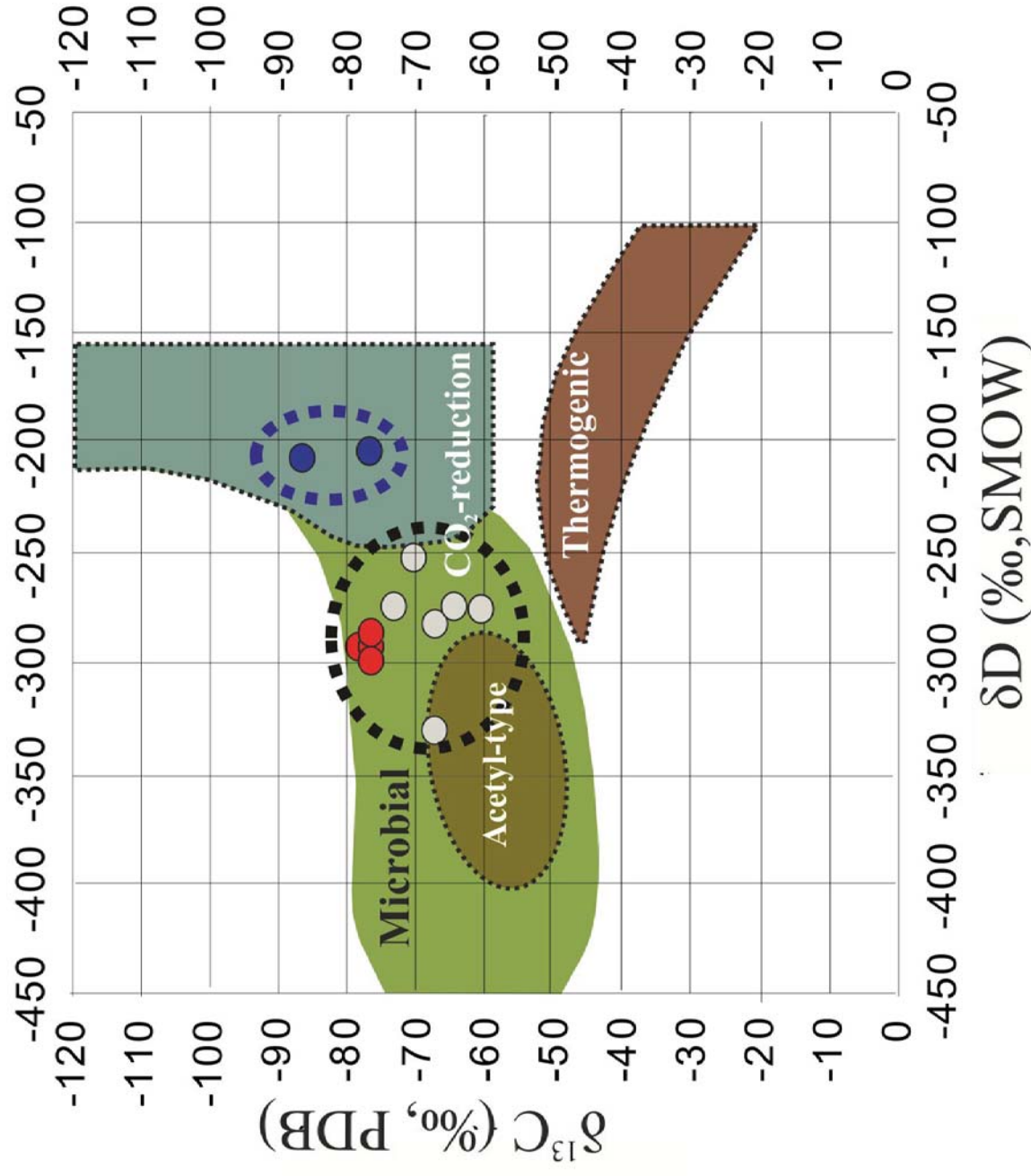












- T-04 ● Ground ice of Marre-Sale ● PLF2  
(Streletskaaya et al., 2016) (Serov et al., 2015)

Subbottom depth, cm	5	10	30	45	60	90	110
Methane CH <sub>4</sub> , μM	0.87	5.42	180.80	168.96	298.26	356.78	734.39
Ethane C <sub>2</sub> H <sub>6</sub> , nM	6.65	10.99	24.08	37.41	42.81	73.39	107.08
Ethylene C <sub>2</sub> H <sub>4</sub> , nM	3.87	3.35	4.13	2.69	4.46	6.17	15.79
Propane C <sub>3</sub> H <sub>8</sub> , nM	4.36	5.59	6.74	10.10	9.23	11.02	55.51
Propylene C <sub>3</sub> H <sub>6</sub> , nM	0.67	1.40	2.60	1.45	2.47	3.61	3.90
i-Buthane i-C <sub>4</sub> H <sub>10</sub> , nM	1.67	1.22	2.92	1.85	2.65	3.36	23.72
n-Buthane n-C <sub>4</sub> H <sub>10</sub> , nM	1.86	1.66	2.70	1.74	2.76	2.67	24.07
i-Buthylene i-C <sub>4</sub> H <sub>8</sub> , nM	1.57	1.08	1.47	1.42	1.42	1.60	3.22
C <sub>1</sub> /ΣC <sub>2</sub> -C <sub>5</sub>	43	215	4051	2983	4534	3505	3149
δ <sup>13</sup> C, ‰ PDB			-22	-16.4	-12.7	-14.95	-30
δD, ‰ SMOW			-287	-288	-296	-295	-281