

1 Identifying global vs. basinal controls on Paleoproterozoic organic 2 carbon and sulfur isotope records

3 K. Paiste^{a,*}, A. Lepland^{a,b,c}, A. L. Zerkle^d, K. Kirsimäe^c, T. Kreitsmann^c, K. Mänd^{c,e}, A. E.
4 Romashkin^f, D. V. Rychanchik^f, A. R. Prave^d

5 ^aCAGE—Centre for Arctic Gas Hydrate, Environment and Climate, Department of
6 Geosciences, UiT The Arctic University of Norway, 9037 Tromsø, Norway.

7 ^bGeological Survey of Norway (NGU), 7491 Trondheim, Norway.

8 ^cDepartment of Geology, University of Tartu, 50411 Tartu, Estonia.

9 ^dSchool of Earth and Environmental Sciences and Centre for Exoplanet Science, University of
10 St Andrews, St Andrews, KY16 9AL Scotland/UK.

11 ^eDepartment of Earth & Atmospheric Sciences, University of Alberta, Edmonton, Alberta T6G
12 2E3, Canada.

13 ^fInstitute of Geology, Karelian Science Centre, Pushkinskaya 11, 185610 Petrozavodsk, Russia.

14 *Corresponding author: email kart.paiste@ut.ee

15 ¹Present address: Department of Geology, University of Tartu, 50411 Tartu, Estonia.

16 Abstract

17 Paleoproterozoic sedimentary successions are important archives of the redox evolution of
18 Earth's atmosphere and oceans. Efforts to unravel the dynamics of our planet's early
19 oxygenation from this archive rely on various geochemical proxies, including stable carbon and
20 sulfur isotopes. However, ancient metasedimentary rocks often experienced early- and late-
21 stage (bio)geochemical processes making it difficult to discern primary environmental signals
22 from bulk-rock $\delta^{13}\text{C}_{\text{org}}$ and $\delta^{34}\text{S}$ values. Such complexity in carbon and sulfur isotope
23 systematics contributes to uncertainty about the redox structure of Paleoproterozoic oceans. A
24 currently popular idea is that, following the Great Oxidation Event, global changes led to low-
25 oxygen environments and temporally fluctuating ocean redox conditions that lasted until the
26 Neoproterozoic. The volcano-sedimentary rocks of the Onega Basin have figured prominently
27 in this concept, particularly the exceptionally organic-rich rocks of the 1.98 Ga Zaonega
28 Formation. However, a growing body of evidence shows that local depositional processes acted
29 to form the $\delta^{13}\text{C}_{\text{org}}$ and pyrite $\delta^{34}\text{S}$ records of the Zaonega Formation, thus calling for careful
30 assessment of the global significance of these isotope records. Placing new and existing organic

31 carbon and sulfur isotope data from the Zaonega Formation into the context of basin history
32 and by comparing those results with key Paleoproterozoic successions of the Francevillian
33 Basin (Gabon), the Pechenga Greenstone Belt (NW Russia) and the Animikie Basin (Canada),
34 we show that the stratigraphic $\delta^{13}\text{C}_{\text{org}}$ and pyrite $\delta^{34}\text{S}$ trends can be explained by local
35 perturbations in biogeochemical carbon and sulfur cycling without requiring global drivers.
36 Despite their temporal disparity, we also demonstrate that individual successions share certain
37 geological traits (e.g. magmatic and/or tectonic activity, hydrocarbon generation, basin
38 restriction) suggesting that their pyrite $\delta^{34}\text{S}$ and $\delta^{13}\text{C}_{\text{org}}$ trends were governed by common
39 underlying mechanisms (e.g. similar basinal evolution and biogeochemical feedbacks) and are
40 not necessarily unique to certain time intervals. We further show that pyrites in these
41 successions that are most likely to capture ambient seawater sulfate isotopic composition have
42 consistent $\delta^{34}\text{S}$ values of 15–18‰, which hints at remarkable stability in the marine sulfur cycle
43 over most of the Paleoproterozoic Era.

44 **Keywords**

45 Paleoproterozoic, Sulfur cycle, Carbon cycle, Ocean redox, Great Oxidation Event, Onega
46 Basin

47 **1. Introduction**

48 The Neoproterozoic and Paleoproterozoic eras saw drastic perturbations in Earth system processes,
49 spanning from the geodynamic, the emergence of continents and the initiation of supercontinent
50 cycles (Reddy and Evans, 2009), to the climatic, as recorded by several episodes of global
51 glaciations (Kopp et al., 2005; Young et al., 1998). In concert with those were major changes
52 in Earth's surface environments: widespread accumulation of iron formations (Bekker et al.,
53 2014; Cloud, 1973; Holland, 1978; Klein, 2005; Konhauser et al., 2002, 2017), disappearance
54 of detrital pyrite and uraninite grains (Berkner and Marshall, 1965; Cloud, 1968; Holland,
55 2006), loss of mass-independent sulfur isotope fractionation (Farquhar et al., 2000; Guo et al.,
56 2009; Luo et al., 2016), the large-magnitude positive $\delta^{13}\text{C}_{\text{carb}}$ excursion of the Lomagundi-Jatuli
57 Event (LJE; Baker and Fallick, 1989; Karhu and Holland, 1996), accumulation of exceptionally
58 organic-rich sediments of the Shunga Event (Melezhik et al., 1999; Kump, 2011; Strauss et al.,
59 2013), and the appearance of phosphorous-rich sedimentary deposits (Lepland et al., 2013;
60 Papineau, 2010).

61 These changes are thought to have occurred as a consequence of the build-up of atmospheric
62 oxygen to above 0.001% present atmospheric levels at c. 2.4–2.3 Ga (Fig. 1; Bekker et al.,
63 2004; Gumsley et al., 2017; Luo et al., 2016; but see also Ohmoto et al., 2014). Known as the
64 Great Oxidation Event (GOE; Holland, 2006), this change towards oxidative surface conditions
65 is postulated as having intensified continental weathering and prompted an increased riverine
66 flux of sulfate, phosphate and metals into the global ocean (e.g. Konhauser et al., 2011). In turn,
67 the increase in the availability of electron acceptors (e.g. O₂, NO₃⁻, Fe³⁺, Mn⁴⁺, SO₄²⁻) and
68 macro/micro nutrients (e.g. PO₄³⁻, Cu, Ni, Zn) is thought to have created the necessary
69 conditions for establishment of new ecological niches exploited by microbes co-evolving with
70 Earth's oxygenation (e.g. Zerkle et al., 2017). Various geochemical datasets though, including
71 stable sulfur and carbon isotopes, imply that the oxygenation of Earth's atmosphere was
72 dynamic and marked by spatiotemporal heterogeneities (Large et al., 2019; Luo et al., 2016;
73 Ossa Ossa et al., 2018; Planavsky et al., 2018, 2012; Scott et al., 2014; Sheen et al., 2018).
74 Nevertheless, what is agreed is that a substantial increase in seawater sulfate (SWS)
75 concentrations to > 2.5 mM (Bekker et al., 2006; Reuschel et al., 2012b; Schröder et al., 2008)
76 or even > 10 mM (Blättler et al., 2018) occurred during the c. 2.2–2.0 Ga LJE as a consequence
77 of an oxygenated atmosphere-ocean system. The redox structure of Proterozoic oceans,
78 however, remains debated with interpretations ranging from ocean deoxygenation and
79 widespread euxinia at the end of the LJE (Luo et al., 2016; Ossa Ossa et al., 2018; Planavsky
80 et al., 2012; Scott et al., 2014) to episodic deep-water oxygenation and spatiotemporal ocean
81 redox variability that persisted until the late Neoproterozoic (Diamond et al., 2018; Planavsky
82 et al., 2018; Sheen et al., 2018; Sperling et al., 2015, 2014).

83 Given that microbial sulfate reduction (MSR) coupled to oxidation of organic carbon exert great
84 influence on the redox balance of Earth's surface environments (Fike et al., 2015; Jørgensen,
85 1979), concomitant changes in sedimentary records of global organic carbon and sulfur cycling
86 can provide important insights to the redox state of past oceans (Canfield et al., 2013; Fike et
87 al., 2015; Kump and Garrels, 1986). Biological processes preferentially utilise ¹²C and ³²S
88 isotopes over ¹³C and ³⁴S, respectively, leaving the residual source pools enriched and the
89 products depleted in the heavier isotopes. In low temperature environments the kinetic isotope
90 fractionations imparted by microorganisms typically surpass the extent of abiotic isotope
91 effects. These differences between biological and abiotic isotope effects can be used to trace
92 carbon and sulfur fluxes through various reservoirs. Furthermore, biogeochemical sulfur
93 cycling can produce small mass-dependent fractionations in the minor sulfur isotope ratios

94 ($\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}$) that can provide additional information about different processes governing
95 sulfur cycling even when preserved $\delta^{34}\text{S}$ values overlap (Canfield et al., 2010; Johnston, 2011;
96 Johnston et al., 2005; Seal, 2006; Zerkle et al., 2016, 2010). Therefore, coupled organic carbon
97 ($\delta^{13}\text{C}_{\text{org}}$) and multiple sulfur isotope data ($\delta^{34}\text{S}$, $\Delta^{33}\text{S}$, $\Delta^{36}\text{S}$) can be used as indirect
98 paleoenvironmental proxies for microbial processes and biogeochemical cycling during periods
99 of significant change in Earth's surface environments. However, we recognise that stable
100 isotope values can also be influenced by post-depositional geological processes (e.g. diagenesis,
101 metamorphism) that can significantly affect the extent of fractionation between the oxidised
102 and reduced species and, hence, the preservation of original isotope signatures within the rock
103 record (Aller et al., 2010, 2008; Fike et al., 2015). Screening for pristine geochemical signals
104 is especially important given the rarity of well-preserved Paleoproterozoic sedimentary basins.

105 Geochemical trends, including organic carbon and multiple sulfur isotopes and their global
106 importance, from the c. 1.98 Ga Zaonega Formation in the Onega Basin of northwest Russia
107 have figured prominently in investigations of the post-LJE time. Here we combine new and
108 existing records of S-C systematics, from drill cores through the Zaonega Formation to assess
109 ideas about the redox state of the Paleoproterozoic ocean following the LJE. We compare these
110 records to other key Paleoproterozoic successions, the Francevillian Basin of Gabon, Pechenga
111 Greenstone Belt of northwest Russia and Animikie Basin in Canada, to deconvolve local- from
112 global-scale signals. These successions in particular have shaped ideas about the post-LJE
113 world and their geochemical records and depositional models have helped refine understanding
114 of this critical time period in Earth's history.

115 **2. The Zaonega Formation in the Onega Basin**

116 Situated in the eastern part of the Fennoscandian Shield in NW Russia, the Paleoproterozoic
117 Onega Basin (Fig. 2) contains one of the most complete volcano-sedimentary successions that
118 encompasses both the end of the LJE and the succeeding Shunga Event (Melezhik et al., 1999;
119 Kump, 2011; Strauss et al., 2013); the former is mostly recorded by the carbonate rocks of the
120 Tulomozero Formation and the latter by the organic-rich rocks of the Zaonega Formation
121 (Melezhik and Hanski, 2013). Deposition during the LJE followed a phase of rifting associated
122 with the opening of the Paleoproterozoic Kola Ocean and Svecofennian Sea (Lahtinen et al.,
123 2008; Melezhik et al., 2015) and occurred in sabkha-coastal plain and shallow-marine peritidal
124 carbonate platforms restricted by stromatolite reefs as represented by the Tulomozero
125 Formation (Blättler et al., 2018; Melezhik et al., 2015; Melezhik and Hanski, 2013; Morozov

126 et al., 2010). Continued basinal subsidence created a deep-water shelf-slope-basin during the
127 deposition of the overlying Zaonega Formation (Črne et al., 2014; Melezhik et al., 2015),
128 marked by organic-rich siliciclastic and carbonate rocks interlayered with numerous syn-
129 depositional mafic lava flows and co-magmatic intrusions (Črne et al., 2013a, 2013b; Melezhik
130 et al., 1999; Strauss et al., 2013). Subsequently, reduction in the rate of subsidence and sub-
131 aqueous volcanism led to the filling of the basin by hundreds-of-meters of basalts of the Suisari
132 Formation intercalated with thin organic-rich shale beds (Krupenik et al., 2011; Melezhik et al.,
133 2015; Morozov et al., 2010) with final closure of the basin marked by deposition of turbiditic
134 greywackes and siltstones, many containing pyrobitumen clasts (subaerial seeps), in the
135 lacustrine settings of the Kondopoga Formation (Melezhik et al., 2015). The entire succession
136 underwent greenschist facies metamorphism and deformation into a series of open folds cut by
137 high-angle faults during the 1.89–1.79 Ga Svecofennian orogeny (Ojakangas et al., 2001;
138 Stepanova et al., 2014).

139 The lower and upper parts of the Zaonega Formation consist of intercalated siliciclastic
140 (dominating in the lower part) and carbonate (dominating in the upper part) rocks whereas the
141 middle part contains massive organic-rich rocks (locally termed “shungite”) considered to be
142 one of the world’s oldest petrified oilfields (Melezhik et al., 2013, 1999): total organic carbon
143 (TOC) can reach up to 40 wt.% in shungite rock and as high as 90 wt.% in pyrobitumen-filled
144 veins (Melezhik et al., 2013, 2004, 1999). Oil generation and hydrocarbon migration is
145 attributed to syndepositional magmatism that promoted thermal maturation of the organic
146 matter along the contacts of magmatic bodies (Črne et al., 2013a, 2013b; Qu et al., 2018, 2012).
147 Organic-rich rocks in the middle part of the Zaonega succession have characteristic $\delta^{13}\text{C}_{\text{org}}$
148 values of $< -30\text{‰}$ (Kump, 2011; Lepland et al., 2014; Qu et al., 2018, 2012) and their presence
149 and stratigraphic correlation is well documented in several outcrops and drill cores across the
150 entire Onega Basin (Filippov and Yesipko, 2016; Galdobina, 1987; Melezhik et al., 2013).

151 The overwhelmingly fine-grained textures, dearth of large current-generated sedimentary
152 structures, and rare emplacement of coarser-grained sediment-gravity flow and turbidity-
153 current deposits indicate that deposition occurred predominantly in relatively low-energy
154 environments (Črne et al., 2014, 2013a, 2013b; Galdobina, 1987; Melezhik et al., 2015).
155 Although paleobathymetry is difficult to assess, depths must have remained consistently below
156 storm-wave base during deposition of much of the Zaonega Formation.

157 2. 1 Age of the Zaonega Formation

158 Deposition of the Zaonega Formation in part coincided with the termination of the LJE, which
159 is recorded in the conformably underlying Tulomozero Formation and the lowermost part of
160 the Zaonega Formation (Melezhik et al., 2015; Morozov et al., 2010). This termination occurred
161 at c. 2.06 Ga based on correlations of LJE-bearing rocks across Fennoscandia (Karhu, 1993;
162 Karhu and Holland, 1996) and U-Pb radiometric ages on post-LJE bearing rocks: 2058 ± 2 Ma
163 (Melezhik et al., 2007) and 2056.6 ± 0.8 Ma (Martin et al., 2015) for the Kolosjoki Sedimentary
164 Formation in the Pechenga Greenstone Belt; 2055.5 ± 2.3 Ma for the Il'mozero Sedimentary
165 Formation in the Imandra-Varzuga Greenstone Belt (Martin et al., 2013a, 2013b); and a 2050
166 ± 8 Ma age on felsic volcanites in the Peräphoja Belt, Finland (Perttunen and Vaasjoki, 2001).
167 A younger age of the Zaonega Formation is provided by the emplacement ages of igneous
168 rocks: 1919 ± 18 Ma (Priyatkina et al., 2014), 1956 ± 5 Ma (Stepanova et al., 2014) and 1961.6
169 ± 5.1 Ma (Martin et al., 2015) for dykes that cross-cut the Zaonega Formation; 1969 ± 18 Ma
170 (Puchtel et al., 1998) and 1988 ± 34 Ma (Puchtel et al., 1999) on gabbro sills in the overlying
171 Suisari Formation.

172 While the above ages bracket deposition of the Zaonega Formation between c. 2.06 and 1.98
173 Ga, its exact age remains unresolved. A carbonate whole rock Pb-Pb age of 2090 ± 70 Ma from
174 the underlying Tulomozero Formation (Ovchinnikova et al., 2007) and a 2050 Ma Re-Os age
175 on organic-rich siltstone from the Zaonega Formation (reported in a conference abstract by
176 Hannah et al., 2008) indicate that deposition occurred essentially synchronous with the end of
177 the LJE. More recently, Martin et al. (2015) reported single zircon ages of 1982.0 ± 4.5 Ma for
178 a tuff in the lower Zaonega Formation and 1967.6 ± 3.5 Ma for a detrital zircon grain in the
179 overlying Kondopoga Formation. These upper constraints are within error of zircon ages of
180 1976 ± 9 Ma (Puchtel et al., 1998) and 1975.3 ± 2.8 Ma (Martin et al., 2015) that have been
181 determined for mafic magmatic rocks of the Jangozero Formation below the Tulomozero
182 Formation, interpreted by Martin et al. (2015) as a lava flow. However, our recent field
183 observations at the outcrop area near Hirvas (NW Russia), where the samples were collected
184 for dating by both Puchtel et al. (1998) and Martin et al. (2015), indicate that this mafic body
185 is a sill rather than a lava flow; hence, the ages derived from it cannot be used for defining the
186 maximum depositional age of the Tulomozero and Zaonega formations.

187 In summary, i) the Zaonega Formation was deposited conformably over the Tulomozero
188 Formation; ii) carbonates and calcareous mudstones in the lowermost part of the Zaonega
189 Formation host the termination of the LJE ($\delta^{13}\text{C}_{\text{carb}}$ shift from c. 9‰ to 0‰; Melezhik et al.,

190 2015) that occurred across Fennoscandia at c. 2.06 Ga; iii) syndepositional magmatism—
191 resulting co-magmatic sills with pepperite contacts and deposition of tuffs—place accumulation
192 of the exceptionally organic-rich rocks in the middle part of the Zaonega Formation at c. 1.98
193 Ga, iv) an upper age limit of c. 1.98 Ga is further corroborated by the emplacement ages of the
194 overlying Suisari Formation igneous and volcanic rocks. Taken together, and until more refined
195 age constraints can be determined, we consider c. 1.98 Ga as the depositional age of the Zaonega
196 Formation.

197 *2. 2 New samples and data integrated with published data*

198 In this study we integrate new $\delta^{13}\text{C}_{\text{org}}$ data on 117 samples from the drill core of the Onega
199 Parametric Hole (OPH, 62.1559 N, 34.4073 E) with published geochemical data from OPH and
200 other drill cores including FAR-DEEP 12AB, FAR-DEEP 13A, OnZap1, OnZap3, C-175 and
201 C-5190 (Fig. 3, 4, 5). The OPH core was drilled between 2007–2009 in the southeastern part of
202 the Onega Basin and intersects c. 3000 m of Paleoproterozoic rocks and 500 m of the Archean
203 basement (Fig. 2; Krupenik et al., 2011; Morozov et al., 2010). Here we mainly focus on the
204 2115–525 m depth interval in the OPH core that comprises the Zaonega Formation and most of
205 the sedimentary intervals of the overlying Suisari Formation. As part of the International
206 Continental Drilling Program’s Fennoscandian Arctic Russia-Drilling Early Earth Project
207 (FAR-DEEP) in 2007, FAR-DEEP core 13A (240 m deep, 62.5891 N, 34.9273 E) was drilled
208 near the village of Shunga and FAR-DEEP core 12AB (504 m deep, 62.4947 N, 35.2887 E) c.
209 25 km south-east from the village (Fig. 2). Both cores intersect the middle and upper part of the
210 Zaonega Formation. Additional drilling was performed near the village of Shunga in 2012 when
211 two 60 m long cores, OnZap1 (62.5870 N, 34.9310 E) and OnZap3 (62.5920 N, 34.9280 E),
212 were recovered (Fig. 2). Together these two cores form a 102 m composite section that
213 corresponds to the upper part of the Zaonega Formation. Detailed descriptions of the OPH,
214 FAR-DEEP and OnZap cores are available in Krupenik et al. (2011), Črne et al. (2013a, 2013b)
215 and Paiste et al. (2018), respectively. An additional Zaonega data set has been published from
216 core C-175 (220 m thick interval; Scott et al., 2014) and core C-5190 (842 m thick interval;
217 Melezhik et al., 2013; Scott et al., 2014); these are located 500 m to the northwest and c. 15 km
218 south of the FAR-DEEP 12AB location, respectively (Fig. 2). Combined, these cores span the
219 entire Zaonega Formation but the most detail is provided for the organic-rich intervals in the
220 middle part of the Formation (Fig. 5).

221 **3. Methods**

222 The organic carbon (C_{org}) isotope composition of decarbonated residues was determined by
223 flash combustion using an Elemental Analyser Continuous Flow Isotope Ratio Mass
224 Spectrometer (EA-CF-IRMS: ThermoScientific Delta V Plus with Costech EA) at the NERC
225 Life Sciences Mass Spectrometry Facility at the Scottish Universities Environmental Research
226 Centre. The organic carbon isotope data are reported relative to the Vienna PeeDee Belemnite
227 (V-PDB) standard using delta notation ($\delta^{13}C_{org} = 1000 \cdot (^{13}R_{organic-C}/^{13}R_{V-PDB} - 1)$, where ^{13}R is
228 $^{13}C/^{12}C$). Sample reproducibility was determined by replicate analyses of the international
229 standard USGS40 L-glutamic acid ($\delta^{13}C = -26.39 \pm 0.04\%$ V-PDB) which yielded $\delta^{13}C$ values
230 of $-26.19 \pm 0.04\%$. The $\delta^{13}C$ values have been corrected for the 0.2‰ offset between the
231 USGS40 measurements obtained during analysis and the accepted value for the standard. The
232 sulfur isotopes reported from earlier studies are expressed as $\delta^{34}S = 1000 \cdot (^{34}R_{sample}/^{34}R_{V-CDT} -$
233 $1)$ and $\Delta^{33}S = \delta^{33}S - 1000 \cdot (1 + \delta^{34}S/1000)^{0.515} - 1$). Details of chemical methods and isotope
234 analyses are outlined in Scott et al. (2014) and Paiste et al. (2018, 2020).

235 **4. Results for OPH core and basin-wide correlations**

236 The OPH samples record $\delta^{13}C_{org}$ values ranging from -41.01‰ to -23.65‰ and define a
237 stratigraphic profile marked by a prominent negative excursion ($\delta^{13}C_{org}$ values of $<-30\%$) in
238 the 1226–1080 m interval and near invariant values below and above that excursion. Further,
239 this negative $\delta^{13}C_{org}$ excursion is associated with a positive $\delta^{34}S$ excursion (Fig. 4). Examining
240 all the available $\delta^{13}C_{org}$, $\delta^{34}S$ and $\Delta^{33}S$ core records for the OPH, FAR-DEEP and OnZap cores
241 show that the middle and upper parts of the Zaonega Formation are typified by similar
242 lithofacies, including a basin-wide dolomite marked by a prominent increase in P_2O_5 at the
243 lower contact of that unit (Fig. 3). Further, in all those cores, the strongly ^{13}C -depleted biomass
244 recording $\delta^{13}C_{org}$ values of $<-30\%$ occurs in a specific interval in the middle Zaonega
245 Formation. Consequently, these independent criteria provide a firm basis for making
246 correlations across the Onega Basin (Filippov and Yesipko, 2016; Melezhik et al., 2013) and
247 dividing the Zaonega Formation into Members A, B and C (Fig. 5, Paiste et al., 2020). Direct
248 correlation with cores C-175 and C-5190 studied by Scott et al. (2014) is hampered by the lack
249 of lithological descriptions by those workers. However, given the stratigraphic consistency and
250 uniqueness of the pronounced negative $\delta^{13}C_{org}$ excursion present in all other Zaonega cores, the
251 occurrence of $\delta^{13}C_{org}$ values $<-30\%$ in the depth intervals from 184–16 m in C-5190 and 219–
252 17 m in C-175 indicates their correlation with Member B and the depth interval between 296–

253 184 m in C-5190 with Zaonega Member A (Fig. 5). These correlations are in good agreement
254 with those proposed between cores C-5190 and FAR-DEEP 12AB by Melezhik et al. (2013).

255 In detail, the Zaonega Formation houses a pronounced negative $\delta^{13}\text{C}_{\text{org}}$ excursion with values
256 as low as -42‰ in Member B, whereas Members A and C are characterised by relatively
257 invariant $\delta^{13}\text{C}_{\text{org}}$, $-26.5 \pm 2.4\text{‰}$ (at 1SD level, here and elsewhere) and $-29.7 \pm 3.5\text{‰}$,
258 respectively. The sulfur isotope data show that Member A is marked by uniformly negative
259 $\delta^{34}\text{S}$ (c. $-11.9 \pm 10.4\text{‰}$) and positive $\Delta^{33}\text{S}$ (c. $0.02 \pm 0.04\text{‰}$) values that shift to positive $\delta^{34}\text{S}$
260 ($17.8 \pm 7.5\text{‰}$) and mainly negative $\Delta^{33}\text{S}$ ($-0.02\text{‰} \pm 0.03$) values in Member B ($\Delta^{33}\text{S}$ data shown
261 in Paiste et al., 2020). In Member B the values oscillate around a baseline of c. 18‰ in $\delta^{34}\text{S}$ and
262 c. -0.02‰ in $\Delta^{33}\text{S}$ but several sharp positive excursions in $\delta^{34}\text{S}$, up to 44‰, and negative
263 excursions in $\Delta^{33}\text{S}$, down to -0.12‰ occur. On average sulfides in Member C have $\delta^{34}\text{S}$ values
264 of $15.5\text{‰} \pm 8.4$ and $\Delta^{33}\text{S}$ of $-0.01\text{‰} \pm 0.02$ but record a negative trend to c. 3‰ in $\delta^{34}\text{S}$ and a
265 positive trend to c. 0‰ in $\Delta^{33}\text{S}$ towards the top of the member. All these isotope shifts are
266 accompanied by concomitant changes in TOC and TS abundances, that are relatively low in the
267 lower part of Member A and in Member C but exhibit concentrations up to 65 wt.% TOC and
268 11 wt.% TS in Member B (Fig. 5).

269 **5. Discussion**

270 **5.1 Organic carbon and sulfur isotope records**

271 *5.1.1 Carbon fixation pathways and the $\delta^{13}\text{C}_{\text{org}}$ record*

272 The most common organic matter source in marine sediments throughout Earth's history is the
273 biomass of CO₂-fixing autotrophic organisms with $\delta^{13}\text{C}_{\text{org}}$ values between -10‰ to -30‰
274 (Preuß et al., 1989; Zerkle et al., 2005). This is corroborated by the relative consistency of
275 $\delta^{13}\text{C}_{\text{org}}$ (-25 to -35‰) and $\delta^{13}\text{C}_{\text{carb}}$ (c. 0‰) records throughout Earth's history, reflecting the
276 antiquity of autotrophic CO₂-fixing metabolisms and fairly uniform biomass burial and
277 carbonate deposition through geologic time (Hayes, 1993; Hayes and Waldbauer, 2006; Marais,
278 2001; Schidlowski, 2001; Zerkle et al., 2005). However, natural environments may also host
279 mixed biological communities, some of which can produce biomass significantly more depleted
280 in ¹³C such as those utilising CH₄ from either thermogenic ($\delta^{13}\text{C}_{\text{CH}_4}$ from c. -20‰ to -50‰) or
281 biogenic ($\delta^{13}\text{C}_{\text{CH}_4}$ from c. -50‰ to -90‰) sources (Sackett, 1978; Schoell, 1983; Whiticar,
282 1999) and thereby result in sedimentary organic matter containing various $\delta^{13}\text{C}_{\text{org}}$ signals with
283 values lower than -30‰.

284 Sequences with significant ^{13}C -depleted biomass ($\delta^{13}\text{C}_{\text{org}} < -30\text{‰}$) are most notably
285 documented in the 2.7–2.6 Ga Fortescue Formation (Hayes, 1983; Schidlowski, 2001), the c.
286 2.1 Ga Francevillian D Formation (Gauthier-Lafaye and Weber, 2003, 1989), and the c. 1.98
287 Ga Zaonega Formation (Kump, 2011; Melezhik et al., 1999). Such excursions have been
288 explained variably as a global-scale negative isotope shift in the atmospheric CO_2 source via
289 widespread exhumation and weathering of buried organic carbon (Kump, 2011), by increased
290 methane oxidation (Yudovich et al., 1991) or as an increase of microbial methane cycling
291 driving pervasive methanogenesis and subsequent methanotrophy (Bengston and Hayes, 1994;
292 Hayes, 1983; Knoll and Canfield, 1998). However, basin-scale (or smaller) kerogen alteration
293 and devolatilisation of ^{13}C -depleted hydrocarbons could equally provide a suite of carbon
294 sources to sustain microbial communities, consequently lowering the bulk $\delta^{13}\text{C}_{\text{org}}$ of
295 sedimentary organic matter (Qu et al., 2012). Another way to produce ^{13}C -depleted biomass is
296 by enriching the local dissolved inorganic carbon (DIC) pool in ^{12}C by recycling of organic
297 carbon in restricted and/or stratified environments. For example, organic matter with values as
298 low as -45‰ $\delta^{13}\text{C}_{\text{org}}$ have been reported in a modern mountain lake and -34‰ $\delta^{13}\text{C}_{\text{org}}$ in a
299 Norwegian fjord (Rau, 1978; van Breugel et al., 2005).

300 Metamorphism can also alter the isotope composition of primary organic matter but these
301 isotope effects are generally small ($< 3\text{‰}$) and will typically shift the preserved $\delta^{13}\text{C}_{\text{org}}$ to more
302 positive values (Clayton, 1991; Hayes, 1993; Lewan, 1983; Schidlowski, 2001). Significant
303 negative shifts of residual graphite $\delta^{13}\text{C}_{\text{org}}$ are theoretically possible by oxidation of
304 carbonaceous matter in prograde metamorphic conditions where $>90\%$ of carbon is removed
305 by extreme CO_2 degassing at high temperatures ($>500^\circ\text{C}$; Eiler et al., 1997). In addition,
306 formation of abiogenic organic matter through deposition from carbonic fluids can yield low
307 $\delta^{13}\text{C}_{\text{org}}$ values, but such processes facilitating deposition from fluids also occur at temperature
308 and pressure regimes exceeding low grade, greenschist facies metamorphic conditions (Luque
309 et al., 1998). Therefore, the $\delta^{13}\text{C}_{\text{org}}$ values preserved in organic matter in ancient sedimentary
310 rocks of the Fortescue, Francevillian D and Zaonega Formations can be considered to carry
311 primary signatures reflecting switching between different carbon sources and metabolic
312 pathways that are induced by local and/or global processes. The principal way to discriminate
313 between local and global mechanisms is to place stable isotope records into a depositional
314 context.

315 *5. 1. 2 Global and local processes governing the sulfur isotope system*

316 On geological time scales, seawater sulfate (SWS) concentrations depend on the redox status
317 of the atmosphere-ocean system and the relative fluxes of sulfur into and out of the oceans.
318 Sulfur enters the oceans mainly via riverine input and is removed as oxidised (e.g. gypsum,
319 anhydrite, barite, carbonate/phosphate associated sulfate) and reduced (e.g. pyrite, marcasite,
320 pyrrhotite, sphalerite, chalcopyrite) minerals and organic sulfur (Raven et al., 2018, 2016).
321 These sulfur carriers form as a result of microbial activity (e.g. microbial reduction, oxidation,
322 and/or disproportionation of sulfur compounds) and abiogenic processing (e.g. evaporation,
323 thermogenic sulfate reduction, photolysis), which generate distinct isotope effects (Berner,
324 1964; Berner and Raiswell, 1983; Canfield, 2001; Johnston et al., 2007, 2005; Seal, 2006; Sim
325 et al., 2011). The difference between the sulfur isotope compositions of sulfate and sulfide
326 minerals in sediments and their relative burial fluxes affect the isotopic compositions of both
327 the SWS reservoir and the different sulfur sinks. Ideally, variations in the sulfur isotope
328 composition of sedimentary sulfur species could, thus, be used as a proxy for past ocean
329 chemistry.

330 Reconstructions of deep-time sulfur cycling are often limited to pyrite records because
331 evaporite minerals, which more directly record SWS isotopic composition, are rarely preserved.
332 Although pyrite is stable under a range of geological conditions, diagenetic or hydrothermal
333 fluid flow or regional metamorphism can contribute to sulfur isotope variations in pyrite records
334 (Craig, 1993; Meyer et al., 2017; Wagner and Boyce, 2006). Such late-stage overprinting of the
335 primary pyrite $\delta^{34}\text{S}$ signals can be identified by proper petrographic and geochemical screening
336 of samples (Cui et al., 2018; Fischer et al., 2014; Gregory et al., 2017; Large et al., 2014; Ono
337 et al., 2009; Picard et al., 2018; Wagner and Boyce, 2006; Xiao et al., 2010). Therefore, relying
338 solely on sedimentary pyrite records in paleoenvironmental reconstructions requires careful
339 evaluation of mechanisms that affect its isotope composition at the time of deposition.

340 Strictly anaerobic MSR typically occurs within the shallow sediments where a combination of
341 environmental factors (e.g. organic carbon concentrations, temperature, salinity, pH) can affect
342 sulfate reduction rates, and sedimentary processes and characteristics such as sedimentation
343 rate, porosity, permeability, and sediment reworking can limit sulfate transfer between pore
344 space and the overlying water column (Aller et al., 2010, 2008; Bradley et al., 2016; Canfield
345 et al., 2010; Farquhar et al., 2007; Fike et al., 2015; Ries et al., 2009; Sim et al., 2011).
346 Generally, sedimentary pyrites with negative $\delta^{34}\text{S}$ and positive $\Delta^{33}\text{S}$ values are produced during
347 steady-state conditions where microbial discrimination against ^{34}S is most expressed (Canfield

348 et al., 2010; Rees et al., 1978; Seal, 2006). Conversely, if sulfate transport is limited, Rayleigh
349 distillation effects attenuate biological fractionations and the $\Delta^{33}\text{S}-\delta^{34}\text{S}$ of pore-water sulfide
350 will approximate the initial SWS isotope composition (Berner, 1964; Goldhaber and Kaplan,
351 1975; Johnston et al., 2007; Jørgensen, 1979; Zaback et al., 1993). In addition, the final
352 composition of pyrite also depends on reactive iron availability and the fraction of sulfide that
353 is reoxidised. Even in a closed system, ^{34}S -depleted pyrites will accumulate if only a fraction
354 of pore-water sulfide is scavenged by reactive iron. Moreover, in sediments open to sulfate
355 transport, sulfate limitation can develop as a result of high organic matter mineralisation rates
356 by MSR when sulfate is rapidly consumed, reduced and effectively captured as sedimentary
357 sulfide species (Aller et al., 2010; Berner, 1964; Canfield, 1991; Claypool, 2004; Fike et al.,
358 2015; Goldhaber and Kaplan, 1975; Jørgensen, 1979; Zaback et al., 1993).

359 Physical reworking of sediments (e.g. storms, bioturbation) can disturb redox-stratified
360 sediments and cause mixing between distinct sulfide pools that have undergone some degree of
361 Rayleigh distillation. If formerly reduced sediments at some depth in the sedimentary column
362 are placed into oxidising conditions, full or partial oxidation of pore-water sulfide by abiogenic
363 or biogenic processes can impart additional fractionations of $< 5\%$ amplitude (Balci et al., 2007;
364 Fry et al., 1988; Zerkle et al., 2009) leaving the residual sulfide enriched in ^{34}S . Further reactions
365 between the modified sulfide and reduced iron can generate pyrites that preserve smaller sulfur
366 isotope fractionations than predicted for MSR alone. Thus, shallow water environments, where
367 sediment reworking is most intense, typically record pyrites with more positive and variable
368 isotope signatures than their counterparts in deeper water settings. It has even been suggested
369 that within such environments frequent disturbance of sediments and partial oxidation of sulfide
370 can generate “superheavy pyrites” with $\delta^{34}\text{S}$ values that exceed those of ambient seawater
371 sulfate (Ries et al., 2009). Overall, steadily accumulating pelagic shales are considered less
372 likely to be affected by sedimentary reworking and more open to exchange with the overlying
373 water, thus the sulfur isotope records therein are not as affected by limitations on sulfate
374 diffusion (Fike et al., 2015; Pasquier et al., 2017; Ries et al., 2009).

375 However, this might not always be the case. For example, solid phase sulfur ($\Sigma\text{S} = \text{FeS}, \text{FeS}_2,$
376 S_0) in the Amazon-Guianas inner shelf muds has unusually positive $\delta^{34}\text{S}$ signatures in areas
377 episodically reworked by currents and/or wave action and negative $\delta^{34}\text{S}$ signatures where
378 bioturbation occurs (Aller et al., 2010). If these sulfur phases were converted into pyrite and
379 preserved in the rock record, it would be easy to misinterpret the nature of the processes that
380 lead to such isotope signatures. We acknowledge that bioturbation had a limited effect on

381 sedimentary iron and sulfur cycling prior to colonisation of the seafloor by benthic fauna in the
382 early Phanerozoic (Canfield and Farquhar, 2009). Based on that, it is reasonable to assume that
383 Precambrian nonbioturbated sediments would be more conducive to closed system sulfur
384 cycling.

385 Another important aspect to be considered is the preservation of marine sedimentary records.
386 Plate tectonics has effectively recycled Earth's crust and significantly reduced the likelihood of
387 preserving deep-marine continental margins (Condie et al., 2017; Condie and Kröner, 2008;
388 Hildebrand et al., 2018; Reddy and Evans, 2009). As the result, in Archean and Proterozoic
389 successions the preservation of sedimentary successions is strongly biased towards marginal
390 basins that may compromise the use of shales as representative archives of steady-state
391 conditions (Woodcock, 2004).

392 Given difficulties that arise with variability of sulfate transport in early diagenetic systems and
393 sedimentary facies, it has been suggested that pyrites that precipitate in euxinic conditions are
394 more likely to indicate open-system sulfate reduction (Canfield et al., 2010; Lyons, 1997). This
395 suggestion relies on the simple assumption that euxinia could indicate MSR within the water
396 column, shifting the locus of pyrite formation from the sedimentary environment into the water
397 column. Such pyrites should reflect stratigraphically consistent ^{34}S -depletions if sulfate is not
398 limiting but approach SWS isotopic composition once sulfate limitation leads to smaller
399 biological fractionations. Low apparent fractionation from seawater has, therefore, been used
400 to argue for very low sulfate concentrations, since it has been assumed that biological
401 discrimination against ^{34}S is suppressed at sulfate concentrations <0.2 mM (Bradley et al., 2016;
402 Habicht et al., 2002; Werne et al., 2003). This has given rise to the view that Proterozoic SWS
403 concentrations were significantly lower than those of modern (28 mM) oceans (Canfield, 2001;
404 Canfield and Raiswell, 1999; Habicht et al., 2002; Halevy et al., 2012; Havig et al., 2017; Kah
405 et al., 2004; Lowenstein et al., 2003; Luo et al., 2015; Planavsky et al., 2012). However, this
406 notion has been refuted by theoretical predictions (Wing and Halevy, 2014) and observations
407 of $>20\%$ fractionations between sulfate and sulfide associated with MSR in settings with
408 extremely low sulfate concentrations (<0.01 mM; Crowe et al., 2014).

409 In the modern oxygenated marine realm euxinic conditions are limited to isolated basins (e.g.
410 the Black Sea), settings with restricted water circulation (fjords) or upwelling zones like
411 Benguela, off-shore of Namibia (Meyer and Kump, 2008). In restricted basins, the sulfate
412 reservoir can undergo Rayleigh distillation and evolve towards a more ^{34}S -enriched end-

413 member if sulfide production and retention outpace sulfate supply (Fike et al., 2015; Gomes
414 and Hurtgen, 2013; Gomes and Johnston, 2017; Johnston et al., 2008; Pufahl et al., 2010). These
415 effects can be reversed, for example, if a more stable connection with a larger reservoir is
416 restored or if sulfide sinks are decreased. Consequently, oscillations in basinal circulation and
417 concomitant changes in sulfate levels would produce stratigraphic successions with opposing
418 $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$ trends that are not directly reflective of changes in global ocean chemistry. Thus,
419 unless it can be convincingly shown that euxinic conditions developed in open basins, such
420 sulfur isotope records are likely to depict changes in the local environment, which may or may
421 not track global trends.

422 Identifying unambiguously euxinic water column conditions and pyrite formation pathways in
423 paleorecords is another concern. Geochemical proxies that explore the behaviour of iron in
424 sedimentary environments (e.g. Fe/Al and Fe speciation proxy) have been used as indicators of
425 paleoenvironmental redox conditions (e.g. Clarkson et al., 2014; Poulton and Canfield, 2005).
426 The general premise is that enrichments of iron relative to the terrigenous background and its
427 prevalent partitioning to iron sulfide minerals in sedimentary rocks are compatible with euxinic
428 waters (e.g. Raiswell et al., 1994; Raiswell and Canfield, 2012). While this concept can be
429 applied in isolated basins such as the Black Sea, iron distribution patterns in modern open-
430 marine high productivity settings like the Peruvian and Namibian shelf are often unrelated to
431 water column chemistry and instead reflect the combined effects of local sedimentological and
432 biogeochemical processes (Rico and Sheldon, 2019; Scholz, 2018; Scholz et al., 2019). Another
433 aspect to consider is overprinting of the original proxy signatures, since iron is mobile under
434 most diagenetic and metamorphic conditions and readily redistributed among different mineral
435 phases (Clarkson et al., 2014; Slotznick et al., 2018). Consequently, connecting iron-based
436 proxy signatures to water column chemistry and pyrite formation pathways is challenging and
437 should be supported with additional lines of evidence (e.g. lithological observations,
438 petrography, mineralogical and redox sensitive trace element analyses).

439 Even further, basins that are fully open to the ocean can also exhibit significant regional
440 variability in sedimentary sulfide isotope compositions. For example, small fractionations
441 between $\delta^{34}\text{S}$ of sulfate and sulfide are reported in the modern Benguela upwelling system along
442 the Namibian coast where iron limitation and rapid oxidation of organic carbon promote
443 episodic water column euxinia (Brüchert, 2004; Dale et al., 2009). Moreover, the distribution
444 and development of sulfidic waters along the Namibian shelf relate to regional differences in
445 the geochemical gradients and microbial communities, e.g. the sulfide-oxidising genera

446 *Beggiatoa* vs. *Thiomargarita* (Brüchert, 2004; Dale et al., 2009). Thus, even if it can be
447 established that pyrites precipitated from euxinic waters, their $\delta^{34}\text{S}$ values might not directly
448 relate to global seawater sulfate levels and isotopic composition. All in all, sedimentary pyrites
449 can provide insights to ancient ocean conditions, but careful evaluation of the depositional
450 environment, facies, microbial consortia, basinal configuration and late-stage diagenesis are
451 necessary for any global conclusions to be drawn.

452 **5.2 Organic carbon and sulfur isotopes tracking environmental changes in** 453 **the Paleoproterozoic: Lessons from the Zaonega Formation**

454 *5.2.1 Hydrocarbon migration and alteration of the $\delta^{13}\text{C}_{\text{org}}$ record*

455 Sedimentation of the Zaonega Formation was contemporaneous with magmatism (lava flows,
456 sills with peperite contacts) that produced high heat gradients and hydrothermal circulation.
457 This, in turn, induced thermal cracking of sedimentary organic matter, hydrocarbon generation,
458 oil migration, and formation of ubiquitous pyrobitumen veining and impregnation of mudstones
459 in the upper part of Member A and in Member B (Črne et al., 2013a, 2013b; Melezhik et al.,
460 1999; Qu et al., 2018, 2012). Detailed geochemical investigations show that these Members
461 have distinctive $\delta^{13}\text{C}_{\text{org}}$ values that can be pinpointed in the $\delta^{13}\text{C}_{\text{org}}$ values of organic matter in
462 pyrobitumen veins and their sedimentary host rocks. This fingerprinting confirms that the
463 migrating organic matter was locally sourced and that pyrobitumen impregnation and post-
464 depositional alteration had only a limited (c. 4‰) effect on the preserved $\delta^{13}\text{C}_{\text{org}}$ stratigraphic
465 record (Qu et al., 2012, 2018).

466 The relatively smooth stratigraphic $\delta^{13}\text{C}_{\text{org}}$ trend is, however, interrupted in the 156–136 m
467 interval in the FAR-DEEP 12AB core by the presence of a massive C_{org} -rich rock (Qu et al.,
468 2012). Its isotopic characteristics ($\delta^{13}\text{C}_{\text{org}} = -26\text{‰}$, $\delta^{34}\text{S} = -18.7\text{‰}$, and $\Delta^{33}\text{S} = 0.023\text{‰}$) are
469 distinct from adjacent under- and overlying sediments ($\delta^{13}\text{C}_{\text{org}} = \text{c. } -30\text{‰}$, $\delta^{34}\text{S} = >10.6\text{‰}$ $\delta^{34}\text{S}$,
470 and $\Delta^{33}\text{S} = <-0.015\text{‰}$) but match those of the sedimentary units occurring more than 100 m
471 lower in the Zaonega Formation ($\delta^{13}\text{C}_{\text{org}} = -26\text{‰}$, $\delta^{34}\text{S} = -22.3\text{‰}$ $\delta^{34}\text{S}$, and $\Delta^{33}\text{S} = 0.026\text{‰}$)
472 (Fig. 4). Those relationships are readily reconcilable by recognising that this seemingly
473 anomalous 20 m interval of massive rocks represents hydrocarbons and fluidised sediments,
474 including pyrite, that migrated from Member A upward to the seafloor during accumulation of
475 Member B. Thermal alteration and remobilisation of organic matter, however, cannot be offered
476 as a mechanism to explain the c. 15‰ negative $\delta^{13}\text{C}_{\text{org}}$ excursion encompassed in Member B.
477 While such processes can alter $\delta^{13}\text{C}_{\text{org}}$ values, as observed in pyrobitumen-rich veins and

478 contact zones with magmatic bodies of Member A, the produced shifts are positive and rather
479 small (a few per mil), inconsistent with Member B. Thus, the most straightforward explanation
480 of the pronounced negative $\delta^{13}\text{C}_{\text{org}}$ excursion in Member B is a change in primary biomass.

481 *5. 2. 2 Previous interpretations of the Zaonega Formation C_{org} and S records*

482 Previous studies of the upper Zaonega Formation (e.g. Scott et al., 2014) inferred that Fe
483 speciation data and $\delta^{34}\text{S}$ values (as low as -19‰) of pyrites in strata from Member A recorded
484 conditions compatible with a sizeable sulfate reservoir but that ^{34}S -enriched pyrites ($\delta^{34}\text{S}$ up to
485 +27‰) from strata corresponding to Member B precipitated from an euxinic water column and
486 that this change coincided with a global drawdown in oxygen levels. They further suggested
487 that the negative $\delta^{13}\text{C}_{\text{org}}$ shift that coincides with the ^{34}S -enriched sulfides expressed a
488 biogeochemical response to deoxygenation and low sulfate levels, allowing methane flux and
489 anaerobic oxidation of methane (AOM) in the water column. Selenium (Kipp et al., 2017),
490 molybdenum (Asael et al., 2018) and multi-proxy (Ossa Ossa et al., 2018) studies of c. 2.1–2.0
491 Ga sedimentary rocks, including sections from the Zaonega Formation Member B, have also
492 been used to argue for a global drawdown in oxygen levels at that time. However, instead of
493 the progressive increase in pyrite $\delta^{34}\text{S}$ and decrease in $\Delta^{33}\text{S}$ values that would occur from a
494 unidirectional, worldwide decrease in sulfate concentrations, the high resolution data show that
495 Member B is marked by short-lived excursions around a baseline of c. 18‰ $\delta^{34}\text{S}$ and c. -0.02‰
496 $\Delta^{33}\text{S}$. Further, molybdenum, uranium, rhenium and selenium enrichments and uranium isotope
497 compositions (Kipp et al., 2020; Mänd et al., 2020; Partin et al., 2013; Sheen et al., 2018), as
498 well as negative Ce anomalies in apatite (Joosu et al., 2015) and best preserved carbonate rocks
499 (Kreitsmann et al., 2020) in the upper Zaonega Formation have been used to argue for prevailing
500 oxygenated conditions after the LJE (Mänd et al., 2020). Support for a large oceanic oxidant
501 pool during the Zaonega time was also found by a most recent selenium isotope study by Kipp
502 et al. (2020). Lastly, the coinciding four-fold increase in TOC concentrations and the negative
503 $\delta^{13}\text{C}_{\text{org}}$ excursion in Member B has been explained by a local bloom in methanotrophic
504 organisms (Qu et al., 2012, 2018). Influence of methanotrophy on the basinal DIC pool can also
505 be found from the relatively large up to 4‰ variations in $\delta^{13}\text{C}_{\text{carb}}$ values (as low as -6.3‰) in
506 individual carbonate beds of Member B (Kreitsmann et al., 2019). In that Member B is
507 characterised by syndepositional magmatism, hydrocarbon generation and seepage and changes
508 in depositional conditions, it is not surprising that geochemical studies of the upper Zaonega
509 Formation have produced contradictory interpretations.

510 An open-system steady-state modelling approach to the pyrite sulfur isotope record of the
511 Zaonega Formation demonstrated that the positive $\delta^{34}\text{S}$ and opposing $\Delta^{33}\text{S}$ data in Member B
512 can be explained by increased pyrite burial as a consequence of locally high sulfate reduction
513 rates that ultimately led to diminished fractionation from seawater (Paiste et al., 2020). These
514 results are consistent with the concomitant enrichments in TOC and TS seen in the upper part
515 of Member A and lower to middle part of Member B (Fig. 5) and are suggestive of AOM
516 coupled to MSR (Paiste et al., 2018; Qu et al., 2018, 2012). In contrast to previous work, these
517 studies highlight the importance of local rather than global processes acting on the carbon and
518 sulfur cycles in the Onega Basin. Considering the intraplate rift or active continental margin
519 depositional setting (Črne et al., 2014; Melezhik et al., 2015) local control on sulfur and carbon
520 cycling during deposition of the Zaonega Formation is feasible.

521 *5.2.3 New interpretations of the previous C_{org} and sulfur records*

522 In that syndepositional magmatic-volcanic activity generated hydrocarbon formation, it is likely
523 that upward migrating methane produced from the organic matter of Member A fuelled
524 sediment- or seafloor-housed methanotrophy during deposition of Member B. Thus, the
525 coinciding four-fold increase in TOC, close association of strongly ^{13}C -depleted organic matter
526 and ^{34}S -enriched sulfides within Member B suggest that a surge of carbon sources increased
527 pore-water MSR fostering sulfate limitation and near quantitative uptake of sulfate.
528 Alternatively, methanotrophy may have been established in an euxinic water column on the
529 assumption that SWS levels were very low as previously suggested (Scott et al., 2014). In both
530 of these scenarios, small fractionations from sulfate are expected and pyrites within Member B
531 are most likely to have captured the SWS isotope composition, which was between 6–18‰
532 $\delta^{34}\text{S}_{\text{SO}_4}$ and -0.05 – -0.02‰ $\Delta^{33}\text{S}_{\text{SO}_4}$ based on modelling results (Scott et al., 2014; Paiste et al.,
533 2020).

534 Besides helping to promote high primary productivity, active magmatism-volcanism would
535 have modified basinal geography and influenced sedimentation patterns. Reshaping of the
536 depositional setting and establishment of shallower conditions is implied by a higher frequency
537 of thick dolostone beds in the upper part of the Zaonega Formation (Melezhik et al., 2015) and
538 subsequent emplacement of the Suisari Formation lavas. Unstable redox conditions, as
539 indicated by the large spatio-temporal variability in elemental proxies for the upper part of the
540 Zaonega Formation (e.g. Asael et al., 2018; Kipp et al., 2017; Mänd et al., 2020; Paiste et al.,
541 2018; Scott et al., 2014) also corroborates a semirestricted hydrographic regime because both
542 well-oxygenated open basins and stably stratified restricted basins with anoxic/euxinic deep

543 waters would, in contrast, record relatively constant redox conditions (Algeo and Li, 2020).
544 Specific sediment biogeochemical conditions can, however, induce temporal and regional
545 redox heterogeneity in continent-margin upwelling systems (Algeo and Li, 2020; Dale et al.,
546 2009; Waldron and Probyn, 1991). Similar to modern upwelling zones, the occurrence of
547 abundant phosphates (Lepland et al., 2014), high redox sensitive trace element concentrations
548 as well as elevated uranium isotope ratios in the middle part of the Zaonega Formation point
549 towards locally variable redox conditions in the backdrop of an oxygenated ocean (Mänd et al.,
550 2020). On the other hand, Kreitsmann et al. (2020) has proposed that the preservation of
551 prominent Eu_{SN} anomalies in the Zaonega Formation carbonates and positive and negative Eu_{SN}
552 anomalies in apatites (Joosu et al., 2015) are suggestive of deposition in a semi-restricted rather
553 than fully open-marine embayment, because otherwise the excess Eu would have been rapidly
554 diluted. Taken together the large variability in elemental redox proxies for the upper part of the
555 Zaonega Formation is best explained by sediment accumulation in a semi-confined embayment
556 that had intermittent access to the open ocean. The modern Guaymas Basin in the Gulf of
557 California is a representative example of a dynamic early rifting environment where seasonal
558 upwelling, high heat gradients, hydrothermal fluid flow and syndepositional hydrocarbon
559 generation induce regional redox heterogeneity (Chang et al., 2015; Curray, 1982; Scholz et al.,
560 2019). Thus, instead of the common approach of using the Black Sea as a paleo-marine model
561 system, the Guaymas Basin could serve as a closer analogue for deposition in the Onega Basin
562 during the Zaonega time.

563 The combination of added carbon sources, shallowing and/or the establishment of a physical
564 barrier within the basin could have led to sulfate limitation due to an increased pyrite sink. This
565 would have allowed the basinal sulfate to undergo Rayleigh distillation and explain pyrites in
566 Member B and lower part of Member C that exceed the oceanic SWS isotope composition.
567 However, in order to maintain high rates of MSR over 150–200 meters of strata and sustain a
568 baseline of c. 18‰ $\delta^{34}S$ and c. -0.02‰ $\Delta^{33}S$ of sedimentary sulfides, relatively constant sulfate
569 concentrations had to be maintained. That MSR was not sulfate-limited is further implied by
570 the relatively high but variable TS/TOC ratios that mostly fall between 0.1 and 0.5 in Member
571 B (Melezhik et al., 1999; Paiste et al., 2020), even though organic matter mobilization,
572 impregnation of the sediments with pyrobitumen and hydrocarbon spillage on the seafloor
573 likely lowered TS/TOC ratios. In addition, the pronounced positive >30‰ $\delta^{34}S$ excursions in
574 Member B that occur over a few m of strata are unlikely to reflect rapid fluctuations (in a range
575 of << millions of years) in the size of the SWS reservoir, even if the residence time for sulfate

576 in the Proterozoic ocean was shorter than in the modern (13 Ma; Berner, 2001). Therefore, it is
577 likely that the Onega Basin was periodically cut off and/or sulfate consumption episodically
578 exceeded supply from a larger reservoir. Another possibility is that a separate generation of
579 late-stage pyrite—identified in intervals with $>30\text{‰}$ $\delta^{34}\text{S}$ values (Paiste et al., 2018)—could
580 contribute ^{34}S -enriched sulfur to the bulk pyrite S isotope signature. Whether different pyrite
581 morphologies in the Zaonega Formation carry distinct $\delta^{34}\text{S}$ signatures and to what extent this
582 could affect the bulk pyrite $\delta^{34}\text{S}$ analysis is currently unknown and remains to be determined
583 by future studies. In any case, it is doubtful that the anomalous $>30\text{‰}$ $\delta^{34}\text{S}$ values reflect abrupt
584 fluctuations in the SWS reservoir. This renders previous interpretations suggesting water
585 column methanotrophy in response to very low global SWS levels in Member B equivocal.
586 Further doubts arise from the fact that such interpretations require a close association between
587 the most positive $\delta^{34}\text{S}$ and negative $\delta^{13}\text{C}_{\text{org}}$ in the Zaonega Formation; instead what is observed
588 is that intervals with $\delta^{34}\text{S}$ values $>30\text{‰}$ occur in the upper part of Member B where steadily
589 increasing $\delta^{13}\text{C}_{\text{org}}$ values from -42‰ to -29‰ suggest increasing contributions of normal
590 marine CO_2 -fixing autotrophic biomass.

591 If the Zaonega Formation records global perturbations in the carbon and sulfur cycles, then the
592 ^{13}C -depleted organic matter and ^{34}S -enriched pyrites in Member B represent the postulated
593 global deoxygenation event; however, such conditions must have been short-lived as normal
594 marine biomass and more open-system MSR resumed in Member C. Thus, instead of oscillating
595 atmospheric oxygen levels, what is more likely is that changes in the basin dynamics and
596 depositional environment affected both microbial and sedimentary processes, and it is these that
597 were mutually responsible for generating the $\delta^{13}\text{C}_{\text{org}}$ and multiple sulfur isotope trends within
598 the Onega Basin.

599 **5. 3 Global vs local influence on the seawater sulfate levels and its isotope** 600 **composition in the Paleoproterozoic**

601 There is abundant evidence to suggest that seawater sulfate concentrations significantly
602 increased during the GOE and the 2.2–2.1 Ga LJE (e.g. Guo et al., 2009; Reuschel et al., 2012b;
603 Planavsky et al., 2012; Scott et al., 2014; Luo et al., 2015; Havig et al., 2017; Blättler et al.,
604 2018). However, broadly similar chemostratigraphic trends with positive and upwards
605 increasing $\delta^{34}\text{S}$ values characterize several post-LJE Paleoproterozoic sedimentary successions
606 such as the Onega Basin (Kump et al., 2011; Scott et al., 2014), the Francevillian Basin (Kump
607 et al., 2011; Canfield et al., 2013), the Pechenga Greenstone Belt (Reuschel et al., 2012a) and

608 younger strata such as the Animikie Basin (Poulton et al., 2004, 2010) (Fig. 6). These trends
609 have been proposed to reflect a contraction of the global seawater sulfate reservoir, redox
610 stratified oceans and a return to more reducing surface environments (Canfield et al., 2013;
611 Havig et al., 2017; Johnston et al., 2006; Ossa Ossa et al., 2018; Planavsky et al., 2012; Poulton
612 et al., 2004; Scott et al., 2014). However, we and other workers have questioned the global
613 nature of these preserved $\delta^{34}\text{S}$ records by demonstrating that sedimentary processes, facies
614 changes and basinal restriction could equally govern sulfur cycling and isotope trends within
615 these basins (Pufahl et al., 2010; Reuschel et al., 2012a). Evaluating the meaning of pyrite $\delta^{34}\text{S}$
616 records becomes even more important in light of recent time-series investigations of trace
617 element composition of marine pyrite grains in black shales that have been used to infer that
618 atmospheric oxygen levels remained persistently high until at least 1.8 Ga (Large et al., 2014,
619 2019; Mukherjee and Large, 2016).

620 *5. 3. 1 Similar Paleoproterozoic organic carbon and sulfur isotope records - global vs local* 621 *underlying mechanisms*

622 Kump et al. (2011) argued that the distinct negative trend of $\delta^{13}\text{C}_{\text{org}}$ values in sedimentary
623 successions of the upper part of both the Francevillian and Onega Basins signified a global and
624 synchronous two-step shift towards strongly ^{13}C -depleted biomass at the end of the LJE. The
625 compilation of Kump et al. (2011) used data from the FAR-DEEP 12AB core from the Onega
626 Basin, compared it with data from the Francevillian reference stratigraphic column (Gauthier-
627 Lafaye and Weber, 2003), and noted that the two $\delta^{13}\text{C}_{\text{org}}$ records exhibit similar trends, but are
628 off-set by c. 7‰. However, ascribing outcrop and drill core material from the Francevillian
629 Basin to specific stratigraphic intervals of the reference lithostratigraphic column has proven to
630 be challenging. For example, the unit containing the most ^{13}C -depleted biomass ($\delta^{13}\text{C}_{\text{org}} < -$
631 45‰) has been variably assigned to the Francevillian C (FC) Formation (Gauthier-Lafaye and
632 Weber, 2003) or to the FD Formation (Canfield et al., 2013). The lack of robust stratigraphic
633 correlation and disparate representations of the Francevillian reference lithostratigraphic
634 column by different authors complicate direct comparisons with other basins.

635 Canfield et al. (2013) and Ossa-Ossa et al. (2018) have demonstrated that the first negative
636 $\delta^{13}\text{C}_{\text{org}}$ shift (from c. -25‰ to -35‰) in the middle part of the Francevillian B Formation (FB1)
637 coincides with a positive shift in pyrite $\delta^{34}\text{S}$ values (from c. -20‰ to 10‰, Fig. 6). The sulfur
638 isotopes in the upper FB Formation (FB2) return to predominantly negative values ranging from
639 -34‰ to 5‰. The overlying FC Formation exhibits more variable $\delta^{34}\text{S}$ values, with carbonate
640 samples containing typically isotopically heavier pyrite ($\delta^{34}\text{S}$ from 12‰ to 59‰) than shales

641 ($\delta^{34}\text{S}$ from -4‰ to 16‰). According to Canfield et al. (2013), the FD Formation at the top of
642 the Francevillian succession houses the second step of the negative $\delta^{13}\text{C}_{\text{org}}$ shift representing
643 the most ^{13}C -depleted biomass ($\delta^{13}\text{C}_{\text{org}}$ values between -40‰ and -50‰) and pyrite $\delta^{34}\text{S}$ values
644 typically ranging between -5‰ to 8‰ (with the exception of one 24‰ data point). By
645 combining Francevillian $\delta^{13}\text{C}_{\text{org}}$ and $\delta^{34}\text{S}$ records, Ossa-Ossa et al. (2018) argued for negative
646 coupling of the two isotope systems caused by shoaling of anoxic deep waters and sulfate
647 limitation that was accompanied by two global episodes of ocean-atmosphere deoxygenation.
648 While the opposing behaviour of the $\delta^{13}\text{C}_{\text{org}}$ and $\delta^{34}\text{S}$ trends during the first negative $\delta^{13}\text{C}_{\text{org}}$
649 shift in the FB Formation is evident, the available data do not support such co-variation during
650 the second negative $\delta^{13}\text{C}_{\text{org}}$ shift in the upper FC and FD Formations. Notably, within the FC
651 and FD Formations the most ^{13}C -depleted biomass does not typically associate with the highest
652 $\delta^{34}\text{S}$ values. Using the interpretation of Scott et al. (2014) from the Zaonega Formation it was
653 reasoned that low-sulfate conditions resulted in an overall increase in methane production by
654 methanogens and a shift of methanotrophy into the water column, thus explaining the negative
655 $\delta^{13}\text{C}_{\text{org}}$ shift and pyrites with highly positive $\delta^{34}\text{S}$ values (Ossa-Ossa et al., 2018). However, the
656 decoupling of the $\delta^{34}\text{S}$ and $\delta^{13}\text{C}_{\text{org}}$ records in the upper part of the Francevillian succession does
657 not favour such reasoning nor a pronounced contraction of the marine sulfate reservoir during
658 the second negative $\delta^{13}\text{C}_{\text{org}}$ shift. Instead, these results suggest that the observed $\delta^{13}\text{C}_{\text{org}}$ and
659 $\delta^{34}\text{S}$ trends can be produced by both local and global processes – their implication for large-
660 scale changes in Earth's surface environments then rests mainly on comparisons with other
661 basins of similar age (Ossa-Ossa et al., 2018).

662 Such comparisons, however, are fraught with difficulty in the Paleoproterozoic and verifying
663 the global synchronicity of short-term (<10 Ma) geochemical trends is in most cases
664 irresolvable by currently available geochronology methods due to precision and accuracy
665 uncertainties. This is the case with the interpretation that a singular event encompassing large
666 fluctuations in SWS concentrations was the underlying cause of the organic carbon and sulfur
667 isotope records on two different cratons, as derived from the upper FC and FD Formations
668 (Ossa Ossa et al., 2018) and the Zaonega Formation (Kump, 2011; Scott et al., 2014). The
669 maximum age for most of the Francevillian succession is determined by a zircon $^{207}\text{Pb}/^{206}\text{Pb}$
670 age of 2191 ± 13 Ma derived from the N'goutou magmatic complex that intrudes the
671 Francevillian FA, FB and possibly FC Formations (Sawaki et al., 2017; Weber et al., 2016),
672 whereas the minimum age is defined as 2083 ± 6 Ma by zircon ages from a welded ignimbrite
673 tuff in the FD Formation in the Lastrousville sub-basin (Horie et al., 2005). Current age

674 constraints make the Zaonega Formation younger, at minimum c. 30 Ma (Hannah et al., 2008)
675 and at maximum c. 100 Ma (Martin et al., 2015; Weber and Gauthier-Lafaye, 2013) and indicate
676 that instead of a single global event, the geochemical conditions in the Francevillian and Onega
677 Basins were not coeval. We suggest that a similar style of basinal evolution for the intracratonic
678 Francevillian and the Onega Basins could be responsible for the similar $\delta^{13}\text{C}_{\text{org}}$ and $\delta^{34}\text{S}$ trends
679 recorded at different time periods in these basins.

680 A recent review of the evolution of the c. 2.0 Ga Eburnean Orogeny (Weber et al., 2016)
681 suggests that the depositional history of the Francevillian sedimentary basin is comparable to
682 the Onega Basin, although the latter was magmatically more active. According to their
683 interpretation the black shales of the FB Formation accumulated in confined and restricted
684 environments during a period of tectonic instability and submarine volcanism. The overlying
685 FC Formation, which has been thought to record the falling limb of the LJE, is characterised by
686 widespread emersion and deposition of stromatolitic and evaporitic carbonates in very shallow
687 environments, similar to the Tulomozero Formation of the Onega Basin. Moreover, the
688 common occurrence of bitumen filled fractures in the lower part of the FC Formation has been
689 related to syn-depositional hydrocarbon migration and seepage that likely invigorated
690 methanotrophy and its contribution to the sedimentary biomass (Ossa-Ossa et al., 2018). In a
691 larger tectonic framework the FA–FC Formations accumulated during a phase of basin opening,
692 whereas the FD Formation represents sedimentation in a foreland basin towards the end of the
693 Eburnean Orogeny (Weber et al., 2016). Pyrite precipitation within the FD Formation was
694 interpreted to have occurred under euxinic conditions (Canfield et al., 2013) and the strongly
695 ^{13}C -depleted biomass was related by Kump et al. (2011) to increased weathering of LJE organic
696 carbon. However, this scenario was subsequently dismissed, since the negative $\delta^{13}\text{C}_{\text{org}}$ shift in
697 the underlying FB Formation occurs during accumulation of isotopically heavy LJE carbonates
698 (Ossa-Ossa et al., 2018). Instead, Ossa-Ossa et al. (2018) forwarded a scenario whereby
699 increased methane cycling occurred in low-sulfate environments, similar to the interpretation
700 of the FC Formation S and C_{org} isotope profiles. Even so, hydrocarbon migration and seepage
701 are typically restricted to environments where high heat gradients are generated locally due to
702 active magmatism and/or tectonic processes that can further impact depositional patterns and
703 nutrient fluxes within confined areas. Given the similar history of basinal evolution, we suggest
704 that the observed range of $\delta^{34}\text{S}$ and $\delta^{13}\text{C}_{\text{org}}$ in the Francevillian can therefore be explained by
705 local processes without necessitating global oscillations in the sulfur and carbon cycles, as we
706 have demonstrated for the Zaonega Formation.

707 In addition to the Onega and Francevillian Basins, pyrite sulfur isotope records from other
708 Paleoproterozoic metasedimentary basins have also been subjects of debate about the meaning
709 of the isotopic information and if the size and isotopic composition of the contemporaneous
710 SWS reservoir can be estimated from bulk-rock pyrite $\delta^{34}\text{S}$ records alone. The organic-rich
711 sediments of the Pilgijärvi Sedimentary Formation in the Pechenga Greenstone Belt record
712 pyrites with positive $\delta^{34}\text{S}$ values up to 18.7‰ that accumulated in a tectonically active basin
713 bound to an oceanic rift system (Melezhik and Sturt, 1994; Reuschel et al., 2012a). Until
714 recently, the Zaonega and Pilgijärvi Sedimentary Formation were considered time equivalent;
715 however, new age constraints suggest that the latter is significantly younger (1.92 Ga instead
716 of 1.98 Ga; Martin et al., 2015). While some authors have linked ^{34}S -enriched pyrites in the
717 Pilgijärvi Sedimentary Formation to global fluctuations in seawater sulfate levels (Planavsky
718 et al., 2012; Having et al., 2017), Reuschel et al. (2012b) suggested that a combination of high
719 sedimentation rates, increased organic carbon loading and highly reactive iron availability
720 resulted in rapid pore water MSR and effective pyrite burial (Claypool, 2004). In such closed-
721 system conditions, if all sulfate entering the pore waters was reduced and scavenged by reduced
722 iron, the preserved sedimentary pyrites with the most positive $\delta^{34}\text{S}$ would have captured the
723 sulfur isotope signature of the contemporaneous seawater (Johnston et al., 2006; Zaback et al.,
724 1993). While this interpretation requires a more thorough assessment, it is intriguing that pyrites
725 in the organic-rich Member B of the Zaonega Formation typically have very similar $\delta^{34}\text{S}$ values
726 of c. 18‰.

727 Younger Paleoproterozoic successions such as the Gunflint and Rove Formations of the
728 Animikie Basin have, likewise, been used to estimate a $\delta^{34}\text{S}$ signature of c. 17‰ for seawater
729 sulfate at 1.84 Ga (Poulton et al., 2004, 2010). The Rove Formation was further interpreted to
730 reflect low seawater sulfate concentrations under an oxygen-poor atmosphere. This conclusion
731 was largely based on the assumption that water circulation between the ocean and the Animikie
732 Basin was unrestricted and that the precipitated pyrites reflect relatively small fractionations
733 from seawater sulfate. Pyrites in the Gunflint Formation with average $\delta^{34}\text{S}$ values of $8.4\text{‰} \pm$
734 5.4‰ were interpreted to have formed within shallow sediments, while pyrites with $\delta^{34}\text{S}$ values
735 of $17.2\text{‰} \pm 5.1\text{‰}$ in the overlying Rove Formation were suggested to have precipitated from
736 an euxinic water column. Taken together with other lines of evidence, the overall small
737 fractionations between the estimated SWS composition and the positive $\delta^{34}\text{S}$ values were
738 suggested to indicate SWS concentrations between 0.2 and 2.4 mM during accumulation of the
739 Rove Formation (Poulton et al., 2004). However, a subsequent study of the time-equivalent

740 Michigamme Formation in the Baraga Group was used to argue that the Animikie Basin became
741 increasingly isolated during its evolution and that the sulfur isotope profiles reflect basinal
742 rather than global processes (Pufahl et al., 2010). Disseminated pyrites with average $\delta^{34}\text{S}$ values
743 of $6.3 \pm 1.0\text{‰}$ throughout the Michigamme Formation were used to argue for freshening effects
744 in the riverine delta areas that resulted in lateral gradients along the coast of the Animikie
745 foreland. While this interpretation has been debated (Poulton et al., 2010; Fralick et al., 2011),
746 further work in the Animikie Basin has strengthened the impression of important spatial and
747 temporal variability in geochemical profiles across the area (Fralick et al., 2017; Poulton et al.,
748 2010; Poulton and Canfield, 2011).

749 Expanding on their earlier work on the Rove Formation, Poulton et al. (2010) analysed several
750 correlative core sections extending from the shallow- to deep-water settings of the Animikie
751 Basin to further assess a redox stratified open-shelf environment with oxygenated surface,
752 euxinic mid-depth and ferruginous deep waters. This change in iron proxy data across the
753 Animikie shelf was accompanied by a negative shift in pyrite $\delta^{34}\text{S}$ from c. 17‰ in the shallow-
754 shelf (Rove Formation) to $5.5 \pm 3.6\text{‰}$ in the correlative deeper-shelf (Virginia Formation).
755 Following a similar reasoning as in Poulton et al. (2004), samples with an euxinic iron
756 speciation signature were interpreted to capture the $\delta^{34}\text{S}$ signature of seawater sulfate, whereas
757 those with ferruginous signatures were thought to reflect open-system MSR in sediments. Given
758 these associations, a relatively small c. 15‰ fractionation between sulfate and sulfide was
759 estimated and considered characteristic of a limited SWS reservoir that suppressed biological
760 sulfur cycling (Poulton et al., 2010). Enrichments of redox-sensitive metals and positive Cr
761 isotope signatures of meteoric calcite cements in the Gunflint Formation have also been used
762 to argue for oxygenated terrestrial conditions and an imbalance between the redox-state of the
763 ocean and atmosphere, assuming the former remained reducing (Fralick et al., 2017).
764 Conversely, a more recent multi-proxy study of the Stambaugh Formation that represents the
765 deepest-water facies of the Animikie Basin has provided evidence for deep-water oxygenation
766 at c. 1.85 Ga (Planavsky et al., 2018). The latter authors propose that variations in the
767 geochemical signatures of the metasedimentary rocks of the Stambaugh Formation resemble
768 those in the modern Baltic Sea, where local anoxia can develop due to fluctuations in basin
769 connectivity with the larger ocean. While Poulton et al. (2010) and Fralick et al. (2011) argue
770 against basin isolation, Planavsky et al. (2018) offer an alternative scenario in which periodic
771 incursions of low oxygen, suboxic and oxic waters are responsible for the observed redox
772 variability in the Animikie Basin. Although it is not clear if the 1.84 Ga Rove and 1.85 Ga

773 Stambaugh Formations are in part time-equivalent, they are considered time-transgressive on a
774 regional scale (Planavsky et al., 2018), which has some interesting implications for the reported
775 $\delta^{34}\text{S}$ data.

776 The bulk pyrite sulfur isotope data from the deep-water Stambaugh Formation range from -21.6
777 to $+2.2\text{‰}$ (average $-6.3 \pm 6.5\text{‰}$) in $\delta^{34}\text{S}$, as opposed to the dominantly positive shallow-shelf
778 pyrites with $\delta^{34}\text{S}$ of around 17‰ in the Rove Formation. Additionally, the deeper-shelf pyrites
779 of the Virginia Formation carry $\delta^{34}\text{S}$ signatures that fall between the more shore-proximal Rove
780 Formation and most shore-distal Stambaugh Formation pyrites. Applying the modern analogy
781 and the isotopic contrast between shallow- and deep-water settings (e.g. Fike et al., 2015;
782 Pasquier et al., 2017) the discrepancy in $\delta^{34}\text{S}$ between Stambaugh, Virginia and Rove
783 Formations could imply that facies differences are driving variability in the sulfur isotope
784 records within the Animikie Basin. Moreover, if the $\delta^{34}\text{S}$ value of c. 17‰ from the Rove
785 Formation represents a SWS signature as previously proposed, then a fractionation of c. 38‰
786 between sulfate and sulfide could be estimated. Such a c. 38‰ offset between pyrite and SWS
787 is commonly observed in more recent sulfate-rich marine environments. This estimate supports
788 the conclusion of Poulton et al. (2004) that sulfate concentrations within the Animikie Basin
789 exceeded the 0.2 mM threshold generally required to induce large fractionations during MSR.
790 We caution, however, that the upper limit for sulfate concentrations needs further evaluation,
791 as the attenuated fractionation between sulfate and sulfide in the shallower environments is
792 likely depositionally controlled, rather than directly reflective of global fluctuations in SWS
793 concentrations.

794 Taken together, the sulfur isotope profiles of the Francevillian Basin, Onega Basin, Pechenga
795 Greenstone Belt and the Animikie Basin record environmental changes within basins having
796 complex sedimentation histories and variable environmental conditions all of which operated
797 at disparate time intervals. Despite the influence of local factors on the pyrite records, it may
798 be possible to estimate the $\delta^{34}\text{S}$ composition of ambient SWS by thoroughly screening for
799 intervals where quantitative reduction of sulfate to sulfide and conversion into pyrite can be
800 inferred. However, this screening requires a thorough understanding of the basinal evolution
801 and local sedimentary and biological processes that can mutually influence the expression of
802 stable isotope signatures in sedimentary minerals and organic matter. Thus, if typical $\delta^{34}\text{S}$
803 values of $15\text{--}18\text{‰}$ in pyrites from the Zaonega, Pilgujärvi Sedimentary, and Rove Formations
804 reflect an isotope composition similar to the initial SWS, it could imply that the global sulfate
805 reservoir and its isotope composition was relatively stable throughout much of the

806 Paleoproterozoic. This stability, if confirmed, would suggest that the underlying mechanisms
807 that govern pyrite formation and a sizeable SWS reservoir with stable $\delta^{34}\text{S}$ values may have
808 been already established during or shortly after the initial rise in atmospheric oxygen levels,
809 and have remained largely unchanged.

810 **Conclusions**

811 Exploiting the pyrite $\delta^{34}\text{S}$ and $\delta^{13}\text{C}_{\text{org}}$ records from recently drilled cores, bisecting the entire
812 Zaonega Formation of the Onega Basin, we have undertaken a careful investigation of these
813 proxies in the context of depositional history to constrain global and/or local influences on
814 sulfur and carbon cycling as recorded in Paleoproterozoic rock records. Considering the
815 emerging understanding of sulfur isotope effects associating with sedimentary and biological
816 processes, the entire Zaonega Formation pyrite isotope record can be explained by local
817 depositional changes and corresponding microbial feedbacks without the need to invoke large-
818 scale fluctuations in ocean sulfate concentrations and isotope composition. Likewise, the
819 negative $\delta^{13}\text{C}_{\text{org}}$ excursion encompassed in the organic-rich rocks of the middle Zaonega
820 Formation can be explained by a blooming methanotrophic community responding to a surge
821 of carbon sources to the local environment as a consequence of syn-depositional hydrocarbon
822 seepage. In turn, this invigorated MSR causing a high demand for sulfate that overwhelmed its
823 supply from an external pool. However, an influx of sulfate had to be maintained in order to
824 account for the high pyrite burial and MSR rates throughout most of the Zaonega Formation,
825 indicating that SWS levels were not in a state of decline as suggested by previous workers.

826 Further comparisons between the Zaonega Formation pyrite $\delta^{34}\text{S}$ and $\delta^{13}\text{C}_{\text{org}}$ records and the
827 older Francevillian Formations in Gabon reveals that similar but temporally unrelated
828 conditions marked these diachronous successions in which high primary productivity and
829 dynamic basinal conditions governed the sulfur and organic carbon cycles subsequently
830 producing broadly similar stable isotope records; they do not represent global perturbations in
831 the ocean-atmosphere redox system and, in fact, this is a likely commonality between other
832 Paleoproterozoic basins such as the Pechenga Greenstone Belt and Animikie Basin. However,
833 screening for pyrites that infer small fractionations from the initial seawater sulfate has revealed
834 consistent $\delta^{34}\text{S}$ values of 15–18‰ in the Zaonega, Pilgijärvi Sedimentary, and Rove
835 Formations, which could indicate a relatively stable SWS reservoir throughout much of the
836 Paleoproterozoic. All be it speculative, it highlights the importance of interpreting pyrite $\delta^{34}\text{S}$
837 and $\delta^{13}\text{C}_{\text{org}}$ records in a context of sedimentological evidence, facies changes and late-stage

838 diagenesis underpinned by firm chronological constraints in each individual Paleoproterozoic
839 succession.

840 **Acknowledgments**

841 We are grateful to B. A. Wing and D. A. Fike for early discussion of the manuscript. Rona
842 McGill is acknowledged for organic carbon isotope analyses at the NERC Life Sciences Mass
843 Spectrometry Facility at the Scottish Universities Environmental Research Centre. We thank
844 R. R. Large and P. W. Fralick for valuable feedback and appreciate K. Pedersen for editorial
845 handling. The research is part of the Centre for Arctic Gas Hydrate, Environment and Climate
846 and was supported by the Research Council of Norway through its Centres of Excellence
847 funding scheme grant No. 223259. A. L. Z. acknowledges support from a Natural Environment
848 Council Standard Grant NE/J023485/2. K. K, A. L., T. K and K. M were supported by the
849 Estonian Science Agency grant PRG447 and Estonian Center of Analytical Chemistry. K. P.
850 was supported by the European Regional Development Fund and the programme Mobilitas
851 Plus grant MOBJD542. A. E. R and D. V. R. acknowledges support from the state assignment
852 of IG KarRC RAS.

853 **References**

- 854 Algeo, T.J., Li, C., 2020. Redox classification and calibration of redox thresholds in
855 sedimentary systems. *Geochimica et Cosmochimica Acta*.
856 <https://doi.org/10.1016/j.gca.2020.01.055>
- 857 Aller, R.C., Blair, N.E., Brunskill, G.J., 2008. Early diagenetic cycling, incineration, and burial
858 of sedimentary organic carbon in the central Gulf of Papua (Papua New Guinea). *Journal*
859 *of Geophysical Research* 113. <https://doi.org/10.1029/2006JF000689>
- 860 Aller, R.C., Madrid, V., Chistoserdov, A., Aller, J.Y., Heilbrun, C., 2010. Unsteady diagenetic
861 processes and sulfur biogeochemistry in tropical deltaic muds: Implications for oceanic
862 isotope cycles and the sedimentary record. *Geochimica et Cosmochimica Acta* 74,
863 4671–4692. <https://doi.org/10.1016/j.gca.2010.05.008>
- 864 Asael, D., Rouxel, O., Poulton, S.W., Lyons, T.W., Bekker, A., 2018. Molybdenum record from
865 black shales indicates oscillating atmospheric oxygen levels in the early
866 Paleoproterozoic. *Am J Sci* 318, 275–299. <https://doi.org/10.2475/03.2018.01>
- 867 Baker, A.J., Fallick, A.E., 1989. Evidence from Lewisian limestones for isotopically heavy
868 carbon in two-thousand-million-year-old sea water. *Nature* 337, 352–354.
869 <https://doi.org/10.1038/337352a0>
- 870 Balci, N., Shanks, W.C., Mayer, B., Mandernack, K.W., 2007. Oxygen and sulfur isotope
871 systematics of sulfate produced by bacterial and abiotic oxidation of pyrite. *Geochimica*
872 *et Cosmochimica Acta* 71, 3796–3811. <https://doi.org/10.1016/j.gca.2007.04.017>
- 873 Bekker, A., Holland, H.D., Wang, P.L., Rumble, D., Stein, H.J., Hannah, J.L., Coetzee, L.L.,
874 Beukes, N.J., 2004. Dating the rise of atmospheric oxygen. *Nature* 427, 117–120.
875 <https://doi.org/10.1038/nature02260>

- 876 Bekker, A., Karhu, J.A., Kaufman, A.J., 2006. Carbon isotope record for the onset of the
877 Lomagundi carbon isotope excursion in the Great Lakes area, North America.
878 *Precambrian Res* 148, 145–180. <https://doi.org/10.1016/j.precamres.2006.03.008>
- 879 Bekker, A., Planavsky, N.J., Krapež, B., Rasmussen, B., Hofmann, A., Slack, J.F., Rouxel, O.J.,
880 Konhauser, K.O., 2014. 9.18 - Iron Formations: Their Origins and Implications for
881 Ancient Seawater Chemistry, in: Holland, H.D., Turekian, K.K. (Eds.), *Treatise on*
882 *Geochemistry* (Second Edition). Elsevier, Oxford, pp. 561–628.
883 <https://doi.org/10.1016/B978-0-08-095975-7.00719-1>
- 884 Bengtson, S., Hayes, J., 1994. *Early Life on Earth*. Nobel Symposium. Columbia University
885 Press, New York.
- 886 Berkner, L.V., Marshall, L.C., 1965. N.A.S. Symposium on the Evolution of the Earth's
887 Atmosphere: HISTORY OF MAJOR ATMOSPHERIC COMPONENTS. *Proceedings*
888 *of the National Academy of Sciences* 53, 1215–1226.
889 <https://doi.org/10.1073/pnas.53.6.1215>
- 890 Berner, R.A., 2001. Modeling atmospheric O₂ over Phanerozoic time.
- 891 Berner, R.A., 1964. An idealized model of dissolved sulfate distribution in recent sediments.
892 *Geochimica et Cosmochimica Acta* 28, 1497–1503.
- 893 Berner, R.A., Raiswell, R., 1983. Burial of organic carbon and pyrite sulfur in sediments over
894 phanerozoic time: a new theory. *Geochimica et Cosmochimica Acta* 47, 855–862.
895 [https://doi.org/10.1016/0016-7037\(83\)90151-5](https://doi.org/10.1016/0016-7037(83)90151-5)
- 896 Blättler, C.L., Claire, M.W., Prave, A.R., Kirsimäe, K., Higgins, J.A., Medvedev, P.V.,
897 Romashkin, A.E., Rychanchik, D.V., Zerkle, A.L., Paiste, K., Kreitsmann, T., Millar,
898 I.L., Hayles, J.A., Bao, H., Turchyn, A.V., Warke, M.R., Lepland, A., 2018. Two-
899 billion-year-old evaporites capture Earth's great oxidation. *Science* eaar2687.
900 <https://doi.org/10.1126/science.aar2687>
- 901 Bradley, A.S., Leavitt, W.D., Schmidt, M., Knoll, A.H., Girguis, P.R., Johnston, D.T., 2016.
902 Patterns of sulfur isotope fractionation during microbial sulfate reduction. *Geobiology*
903 14, 91–101. <https://doi.org/10.1111/gbi.12149>
- 904 Brüchert, V., 2004. Physiological and ecological aspects of sulfur isotope fractionation during
905 bacterial sulfate reduction, in: *Special Paper 379: Sulfur Biogeochemistry - Past and*
906 *Present*. Geological Society of America, pp. 1–16. [https://doi.org/10.1130/0-8137-](https://doi.org/10.1130/0-8137-2379-5.1)
907 [2379-5.1](https://doi.org/10.1130/0-8137-2379-5.1)
- 908 Canfield, D.E., 2001. Biogeochemistry of Sulfur Isotopes. *Reviews in Mineralogy and*
909 *Geochemistry* 43, 607–636. <https://doi.org/10.2138/gsrmg.43.1.607>
- 910 Canfield, D.E., 1991. Sulfate Reduction in Deep-Sea Sediments. *Am J Sci* 291, 177–188.
- 911 Canfield, D.E., Farquhar, J., 2009. Animal evolution, bioturbation, and the sulfate concentration
912 of the oceans. *PNAS* 106, 8123–8127. <https://doi.org/10.1073/pnas.0902037106>
- 913 Canfield, D.E., Farquhar, J., Zerkle, A.L., 2010. High isotope fractionations during sulfate
914 reduction in a low-sulfate euxinic ocean analog. *Geology* 38, 415–418.
915 <https://doi.org/10.1130/G30723.1>
- 916 Canfield, D.E., Ngombi-Pemba, L., Hammarlund, E.U., Bengtson, S., Chaussidon, M.,
917 Gauthier-Lafaye, F., Meunier, A., Riboulleau, A., Rollion-Bard, C., Rouxel, O., Asael,
918 D., Pierson-Wickmann, A.-C., El Albani, A., 2013. Oxygen dynamics in the aftermath
919 of the Great Oxidation of Earth's atmosphere. *Proceedings of the National Academy of*
920 *Sciences* 110, 16736–16741. <https://doi.org/10.1073/pnas.1315570110>
- 921 Canfield, D.E., Raiswell, R., 1999. The evolution of the sulfur cycle. *Am J Sci* 299, 697–723.
922 <https://doi.org/10.2475/ajs.299.7-9.697>
- 923 Chang, A.S., Pichevin, L., Pedersen, T.F., Gray, V., Ganeshram, R., 2015. New insights into
924 productivity and redox-controlled trace element (Ag, Cd, Re, and Mo) accumulation in
925 a 55 kyr long sediment record from Guaymas Basin, Gulf of California: Trace elements

- 926 in Guaymas Basin. *Paleoceanography* 30, 77–94.
 927 <https://doi.org/10.1002/2014PA002681>
- 928 Clarkson, M.O., Poulton, S.W., Guilbaud, R., Wood, R.A., 2014. Assessing the utility of Fe/Al
 929 and Fe-speciation to record water column redox conditions in carbonate-rich sediments.
 930 *Chemical Geology* 382, 111–122. <https://doi.org/10.1016/j.chemgeo.2014.05.031>
- 931 Claypool, G.E., 2004. Ventilation of marine sediments indicated by depth profiles of pore water
 932 sulfate and $\delta^{34}\text{S}$, in: Hill, R.J., Leventhal, J., Aizenshtat, Z., Baedeker, M.J., Claypool,
 933 G., Eganhouse, R., Goldhaber, M., Peters, K. (Eds.), *The Geochemical Society Special*
 934 *Publications, Geochemical Investigations in Earth and Space Science: A Tribute to*
 935 *Isaac R. Kaplan*. Elsevier, pp. 59–65. [https://doi.org/10.1016/S1873-9881\(04\)80007-5](https://doi.org/10.1016/S1873-9881(04)80007-5)
- 936 Clayton, C., 1991. Carbon isotope fractionation during natural gas generation from kerogen.
 937 *Marine and Petroleum Geology* 8, 232–240. [https://doi.org/10.1016/0264-](https://doi.org/10.1016/0264-8172(91)90010-X)
 938 [8172\(91\)90010-X](https://doi.org/10.1016/0264-8172(91)90010-X)
- 939 Cloud, P., 1973. Paleocological Significance of the Banded Iron-Formation. *Economic*
 940 *Geology* 68, 1135–1143. <https://doi.org/10.2113/gsecongeo.68.7.1135>
- 941 Cloud, P.E., 1968. Atmospheric and Hydrospheric Evolution on the Primitive Earth: Both
 942 secular accretion and biological and geochemical processes have affected earth's
 943 volatile envelope. *Science* 160, 729–736. <https://doi.org/10.1126/science.160.3829.729>
- 944 Condie, K.C., Arndt, N., Davaille, A., Puetz, S.J., 2017. Zircon age peaks: Production or
 945 preservation of continental crust? *Geosphere* 13, 227–234.
 946 <https://doi.org/10.1130/GES01361.1>
- 947 Condie, K.C., Kröner, A., 2008. When did plate tectonics begin? Evidence from the geologic
 948 record, in: *Special Paper 440: When Did Plate Tectonics Begin on Planet Earth?*
 949 *Geological Society of America*, pp. 281–294. [https://doi.org/10.1130/2008.2440\(14\)](https://doi.org/10.1130/2008.2440(14))
- 950 Craig, J.R., 1993. The Metamorphism of Pyrite and Pyritic Ores: An Overview. *Mineralogical*
 951 *Magazine* 57, 3–18. <https://doi.org/10.1180/minmag.1993.057.386.02>
- 952 Črne, A.E., Melezhik, V.A., Lepland, A., Fallick, A.E., Prave, A.R., Brasier, A.T., 2014.
 953 Petrography and geochemistry of carbonate rocks of the Paleoproterozoic Zaonega
 954 Formation, Russia: Documentation of C-13-depleted non-primary calcite. *Precambrian*
 955 *Res* 240, 79–93. [https://doi.org/DOI 10.1016/j.precamres.2013.10.005](https://doi.org/DOI%2010.1016/j.precamres.2013.10.005)
- 956 Črne, A.E., Melezhik, V.A., Prave, A.R., Lepland, A., Romashkin, A.E., Rychanchik, D.V.,
 957 Hanski, E.J., Luo, Z., 2013a. Zaonega Formation: FAR-DEEP Holes 12A and 12B, and
 958 Neighbouring quarries, in: Melezhik, V.A., Prave, A.R., Fallick, A.E., Hanski, E.J.,
 959 Lepland, A., Kump, L.R., Strauss, H. (Eds.), *Reading the Archive of Earth's*
 960 *Oxygenation: Volume 2: The Core Archive of the Fennoscandian Arctic Russia -*
 961 *Drilling Early Earth Project*, *Frontiers in Earth Sciences*. Springer, pp. 946–1007.
- 962 Črne, A.E., Melezhik, V.A., Prave, A.R., Lepland, A., Romashkin, A.E., Rychanchik, D.V.,
 963 Hanski, E.J., Luo, Z., 2013b. Zaonega Formation: FAR-DEEP Hole 13A, in: Melezhik,
 964 V.A., Prave, A.R., Fallick, A.E., Hanski, E.J., Lepland, A., Kump, L.R., Strauss, H.
 965 (Eds.), *Reading the Archive of Earth's Oxygenation: Volume 2: The Core Archive of*
 966 *the Fennoscandian Arctic Russia - Drilling Early Earth Project*, *Frontiers in Earth*
 967 *Sciences*. Springer, pp. 1008–1046.
- 968 Crockford, P.W., Kunzmann, M., Bekker, A., Hayles, J., Bao, H., Halverson, G.P., Peng, Y.,
 969 Bui, T.H., Cox, G.M., Gibson, T.M., Wörndle, S., Rainbird, R., Lepland, A., Swanson-
 970 Hysell, N.L., Master, S., Sreenivas, B., Kuznetsov, A., Krupenik, V., Wing, B.A., 2019.
 971 Claypool continued: Extending the isotopic record of sedimentary sulfate. *Chemical*
 972 *Geology*. <https://doi.org/10.1016/j.chemgeo.2019.02.030>
- 973 Crowe, S.A., Paris, G., Katsev, S., Jones, C., Kim, S.-T., Zerkle, A.L., Nomosatryo, S., Fowle,
 974 D.A., Adkins, J.F., Sessions, A.L., Farquhar, J., Canfield, D.E., 2014. Sulfate was a

975 trace constituent of Archean seawater. *Science* 346, 735–739.
976 <https://doi.org/10.1126/science.1258966>

977 Cui, H., Kitajima, K., Spicuzza, M.J., Fournelle, J.H., Denny, A., Ishida, A., Zhang, F., Valley,
978 J.W., 2018. Questioning the biogenicity of Neoproterozoic superheavy pyrite by SIMS.
979 *American Mineralogist* 103, 1362–1400. <https://doi.org/10.2138/am-2018-6489>

980 Curray, J.R., Moore, D.G., et al., 1982. Initial Reports of the Deep Sea Drilling Project, 64,
981 Initial Reports of the Deep Sea Drilling Project. U.S. Government Printing Office.
982 <https://doi.org/10.2973/dsdp.proc.64.1982>

983 Dale, A.W., Brüchert, V., Alperin, M., Regnier, P., 2009. An integrated sulfur isotope model
984 for Namibian shelf sediments. *Geochimica et Cosmochimica Acta* 73, 1924–1944.
985 <https://doi.org/10.1016/j.gca.2008.12.015>

986 Diamond, C.W., Planavsky, N.J., Wang, C., Lyons, T.W., 2018. What the ~1.4 Ga Xiamaling
987 Formation can and cannot tell us about the mid-Proterozoic ocean. *Geobiology* 16, 219–
988 236. <https://doi.org/10.1111/gbi.12282>

989 Eiler, J.M., Mojzsis, S.J., Arrhenius, G., 1997. Carbon isotope evidence for early life. *Nature*
990 386, 665.

991 Farquhar, J., Bao, H., Thiemens, M., 2000. Atmospheric Influence of Earth’s Earliest Sulfur
992 Cycle. *Science* 289, 756–758. <https://doi.org/10.1126/science.289.5480.756>

993 Farquhar, J., Johnston, D.T., Wing, B.A., 2007. Implications of conservation of mass effects on
994 mass-dependent isotope fractionations: Influence of network structure on sulfur isotope
995 phase space of dissimilatory sulfate reduction. *Geochimica et Cosmochimica Acta* 71,
996 5862–5875. <https://doi.org/10.1016/j.gca.2007.08.028>

997 Farquhar, J., Zerkle, A.L., Bekker, A., 2014. Geologic and Geochemical Constraints on Earth’s
998 Early Atmosphere, in: *Treatise on Geochemistry*. Elsevier, pp. 91–138.
999 <https://doi.org/10.1016/B978-0-08-095975-7.01304-8>

1000 Fike, D.A., Bradley, A.S., Rose, C.V., 2015. Rethinking the Ancient Sulfur Cycle. *Annu Rev*
1001 *Earth Pl Sc* 43, 593–622. <https://doi.org/10.1146/annurev-earth-060313-054802>

1002 Filippov, M.M., Yesipko, O.A., 2016. Geological-geophysical marker horizons of the
1003 Paleoproterozoic Onega Structure. *Proceedings of the Karelian Research Centre of the*
1004 *Russian Academy of Sciences*.

1005 Fischer, W.W., Fike, D.A., Johnson, J.E., Raub, T.D., Guan, Y., Kirschvink, J.L., Eiler, J.M.,
1006 2014. SQUID-SIMS is a useful approach to uncover primary signals in the Archean
1007 sulfur cycle. *Proceedings of the National Academy of Sciences* 111, 5468–5473.
1008 <https://doi.org/10.1073/pnas.1322577111>

1009 Fralick, P., Planavsky, N., Burton, J., Jarvis, I., Addison, W.D., Barrett, T.J., Brumpton, G.R.,
1010 2017. Geochemistry of Paleoproterozoic Gunflint Formation carbonate: Implications
1011 for hydrosphere-atmosphere evolution. *Precambrian Research* 290, 126–146.
1012 <https://doi.org/10.1016/j.precamres.2016.12.014>

1013 Fralick, P.W., Poulton, S.W., Canfield, D.E., 2011. Does the Paleoproterozoic Animikie Basin
1014 record the sulfidic ocean transition?: COMMENT. *Geology* 39, e241–e241.
1015 <https://doi.org/10.1130/G31747C.1>

1016 Fry, B., Ruf, W., Gest, H., Hayes, J.M., 1988. Sulfur isotope effects associated with oxidation
1017 of sulfide by O₂ in aqueous solution. *Chemical Geology: Isotope Geoscience section*
1018 73, 205–210. [https://doi.org/10.1016/0168-9622\(88\)90001-2](https://doi.org/10.1016/0168-9622(88)90001-2)

1019 Galdobina, L.P., 1987. The Ludikovi superhorizon, in: Sokolov, V. A. (Ed.), *Geology of*
1020 *Karelia*. Nauka (Science), Leningrad, pp. 59–67.

1021 Gauthier-Lafaye, F., Weber, F., 2003. Natural nuclear fission reactors: time constraints for
1022 occurrence, and their relation to uranium and manganese deposits and to the evolution
1023 of the atmosphere. *Precambrian Research* 120, 81–100. [https://doi.org/10.1016/S0301-9268\(02\)00163-8](https://doi.org/10.1016/S0301-9268(02)00163-8)

1024

- 1025 Gauthier-Lafaye, F., Weber, F., 1989. The Francevillian (Lower Proterozoic) uranium ore
 1026 deposits of Gabon. *Economic Geology* 84, 2267–2285.
 1027 <https://doi.org/10.2113/gsecongeo.84.8.2267>
- 1028 Goldhaber, M.B., Kaplan, I.R., 1975. Controls and consequences of sulfate reduction rates in
 1029 recent marine sediments. *Soil Science* 119.
- 1030 Gomes, M.L., Hurtgen, M.T., 2013. Sulfur isotope systematics of a euxinic, low-sulfate lake:
 1031 Evaluating the importance of the reservoir effect in modern and ancient oceans. *Geology*
 1032 41, 663–666. <https://doi.org/10.1130/G34187.1>
- 1033 Gomes, M.L., Johnston, D.T., 2017. Oxygen and sulfur isotopes in sulfate in modern euxinic
 1034 systems with implications for evaluating the extent of euxinia in ancient oceans.
 1035 *Geochimica et Cosmochimica Acta* 205, 331–359.
 1036 <https://doi.org/10.1016/j.gca.2017.02.020>
- 1037 Gregory, D.D., Lyons, T.W., Large, R.R., Jiang, G., Stepanov, A.S., Diamond, C.W., Figueroa,
 1038 M.C., Olin, P., 2017. Whole rock and discrete pyrite geochemistry as complementary
 1039 tracers of ancient ocean chemistry: An example from the Neoproterozoic Doushantuo
 1040 Formation, China. *Geochimica et Cosmochimica Acta* 216, 201–220.
 1041 <https://doi.org/10.1016/j.gca.2017.05.042>
- 1042 Gumsley, A.P., Chamberlain, K.R., Bleeker, W., Soderlund, U., Kock, M.D.O., Larsson, E.R.,
 1043 Bekker, A., 2017. Timing and tempo of the Great Oxidation Event. *P Natl Acad Sci*
 1044 *USA* 114, 1811–1816. <https://doi.org/10.1073/pnas.1608824114>
- 1045 Guo, Q., Strauss, H., Kaufman, A.J., Schröder, S., Gutzmer, J., Wing, B., Baker, M.A., Bekker,
 1046 A., Jin, Q., Kim, S.-T., Farquhar, J., 2009. Reconstructing Earth’s surface oxidation
 1047 across the Archean-Proterozoic transition. *Geology* 37, 399–402.
 1048 <https://doi.org/10.1130/G25423A.1>
- 1049 Habicht, K.S., Gade, M., Thamdrup, B., Berg, P., Canfield, D.E., 2002. Calibration of sulfate
 1050 levels in the Archean Ocean. *Science* 298, 2372–2374. <https://doi.org/DOI>
 1051 [10.1126/science.1078265](https://doi.org/10.1126/science.1078265)
- 1052 Halevy, I., Peters, S.E., Fischer, W.W., 2012. Sulfate Burial Constraints on the Phanerozoic
 1053 Sulfur Cycle. *Science* 337, 331–334. <https://doi.org/10.1126/science.1220224>
- 1054 Hannah, J.L., Stein, H.J., Zimmerman, A., Yang, G., Melezhik, V.A., Filippov, M.M., Turgeon,
 1055 S.C., Creaser, R.A., 2008. Re-Os geochronology of shungite: A 2.05 Ga fossil oil field
 1056 in Karelia. Presented at the 33rd International Geological Congress, Goldschmidt
 1057 Conference Abstracts, Oslo.
- 1058 Havig, J.R., Hamilton, T.L., Bachan, A., Kump, L.R., 2017. Sulfur and carbon isotopic
 1059 evidence for metabolic pathway evolution and a four-stepped Earth system progression
 1060 across the Archean and Paleoproterozoic. *Earth-Science Reviews* 174, 1–21.
 1061 <https://doi.org/10.1016/j.earscirev.2017.06.014>
- 1062 Hayes, J.M., 1993. Factors controlling ¹³C contents of sedimentary organic compounds:
 1063 Principles and evidence. *Marine Geology, Marine Sediments, Burial, Pore Water*
 1064 *Chemistry, Microbiology and Diagenesis* 113, 111–125. [https://doi.org/10.1016/0025-](https://doi.org/10.1016/0025-3227(93)90153-M)
 1065 [3227\(93\)90153-M](https://doi.org/10.1016/0025-3227(93)90153-M)
- 1066 Hayes, J.M., 1983. Geochemical evidence bearing on the origin of aerobiosis, a speculative
 1067 hypothesis.
- 1068 Hayes, J.M., Waldbauer, J.R., 2006. The carbon cycle and associated redox processes through
 1069 time. *Philos Trans R Soc Lond B Biol Sci* 361, 931–950.
 1070 <https://doi.org/10.1098/rstb.2006.1840>
- 1071 Hildebrand, R.S., Whalen, J.B., Bowring, S.A., 2018. Resolving the crustal composition
 1072 paradox by 3.8 billion years of slab failure magmatism and collisional recycling of
 1073 continental crust. *Tectonophysics* 734–735, 69–88.
 1074 <https://doi.org/10.1016/j.tecto.2018.04.001>

- 1075 Holland, H.D., 1978. The chemistry of the atmosphere and oceans. Wiley, New York 351. ISBN
1076 10: 0471035092
- 1077 Holland, H.D., 2006. The oxygenation of the atmosphere and oceans. *Philos T R Soc B* 361,
1078 903–915. <https://doi.org/DOI 10.1098/rstb.2006.1838>
- 1079 Johnston, D.T., 2011. Multiple sulfur isotopes and the evolution of Earth's surface sulfur cycle.
1080 *Earth-Sci Rev* 106, 161–183. <https://doi.org/DOI 10.1016/j.earscirev.2011.02.003>
- 1081 Johnston, D.T., Farquhar, J., Canfield, D.E., 2007. Sulfur isotope insights into microbial sulfate
1082 reduction: When microbes meet models. *Geochimica et Cosmochimica Acta* 71, 3929–
1083 3947. <https://doi.org/DOI 10.1016/j.gca.2007.05.008>
- 1084 Johnston, D.T., Farquhar, J., Summons, R.E., Shen, Y., Kaufman, A.J., Masterson, A.L.,
1085 Canfield, D.E., 2008. Sulfur isotope biogeochemistry of the Proterozoic McArthur
1086 Basin. *Geochimica et Cosmochimica Acta* 72, 4278–4290.
1087 <https://doi.org/10.1016/j.gca.2008.06.004>
- 1088 Johnston, D.T., Farquhar, J., Wing, B.A., Kaufman, A., Canfield, D.E., Habicht, K.S., 2005.
1089 Multiple sulfur isotope fractionations in biological systems: A case study with sulfate
1090 reducers and sulfur disproportionators. *Am J Sci* 305, 645–660. [https://doi.org/DOI](https://doi.org/DOI 10.2475/ajs.305.6-8.645)
1091 [10.2475/ajs.305.6-8.645](https://doi.org/DOI 10.2475/ajs.305.6-8.645)
- 1092 Johnston, D.T., Poulton, S.W., Fralick, P.W., Wing, B.A., Canfield, D.E., Farquhar, J., 2006.
1093 Evolution of the oceanic sulfur cycle at the end of the Paleoproterozoic. *Geochimica et*
1094 *Cosmochimica Acta* 70, 5723–5739. <https://doi.org/DOI 10.1016/j.gca.2006.08.001>
- 1095 Joosu, L., Lepland, A., Kirsimae, K., Romashkin, A.E., Roberts, N.W., Martin, A.P., Crne,
1096 A.E., 2015. The REE-composition and petrography of apatite in 2 Ga Zaonega
1097 Formation, Russia: The environmental setting for phosphogenesis. *Chemical Geology*
1098 395, 88–107. <https://doi.org/10.1016/j.chemgeo.2014.11.013>
- 1099 Jørgensen, B.B., 1979. A theoretical model of the stable sulfur isotope distribution in marine
1100 sediments. *Geochimica et Cosmochimica Acta* 43, 363–374.
1101 [https://doi.org/10.1016/0016-7037\(79\)90201-1](https://doi.org/10.1016/0016-7037(79)90201-1)
- 1102 Kah, L.C., Lyons, T.W., Frank, T.D., 2004. Low marine sulphate and protracted oxygenation
1103 of the Proterozoic biosphere. *Nature* 431, 834–838. <https://doi.org/10.1038/nature02974>
- 1104 Karhu, J.A., 1993. Paleoproterozoic evolution of the carbon isotope ratios of sedimentary
1105 carbonates in the Fennoscandian Shield. Geological Survey of Finland, Bulletin 371 1–
1106 87.
- 1107 Karhu, J.A., Holland, H.D., 1996. Carbon isotopes and the rise of atmospheric oxygen. *Geology*
1108 24, 867–870. [https://doi.org/Doi 10.1130/0091-7613\(1996\)024<0867:Ciatro>2.3.Co;2](https://doi.org/Doi 10.1130/0091-7613(1996)024<0867:Ciatro>2.3.Co;2)
- 1109 Kipp, M.A., Lepland, A., Buick, R., 2020. Redox fluctuations, trace metal enrichment and
1110 phosphogenesis in the ~2.0 Ga Zaonega Formation. *Precambrian Research* 105716.
1111 <https://doi.org/10.1016/j.precamres.2020.105716>
- 1112 Kipp, M.A., Stüeken, E.E., Bekker, A., Buick, R., 2017. Selenium isotopes record extensive
1113 marine suboxia during the Great Oxidation Event. *PNAS* 114, 875–880.
1114 <https://doi.org/10.1073/pnas.1615867114>
- 1115 Klein, C., 2005. Some Precambrian banded iron-formations (BIFs) from around the world:
1116 Their age, geologic setting, mineralogy, metamorphism, geochemistry, and origins.
1117 *American Mineralogist* 90, 1473–1499. <https://doi.org/10.2138/am.2005.1871>
- 1118 Knoll, A.H., Canfield, D.E., 1998. Isotopic Inferences on Early Ecosystems. *The*
1119 *Paleontological Society Papers* 4, 212–243.
1120 <https://doi.org/10.1017/S1089332600000449>
- 1121 Konhauser, K.O., Hamade, T., Raiswell, R., Morris, R.C., Grant Ferris, F., Southam, G.,
1122 Canfield, D.E., 2002. Could bacteria have formed the Precambrian banded iron
1123 formations? *Geology* 30, 1079. [https://doi.org/10.1130/0091-](https://doi.org/10.1130/0091-7613(2002)030<1079:CBHFTP>2.0.CO;2)
1124 [7613\(2002\)030<1079:CBHFTP>2.0.CO;2](https://doi.org/10.1130/0091-7613(2002)030<1079:CBHFTP>2.0.CO;2)

- 1125 Konhauser, K.O., Lalonde, S.V., Planavsky, N.J., Pecoits, E., Lyons, T.W., Mojzsis, S.J.,
 1126 Rouxel, O.J., Barley, M.E., Rosiere, C., Fralick, P.W., Kump, L.R., Bekker, A., 2011.
 1127 Aerobic bacterial pyrite oxidation and acid rock drainage during the Great Oxidation
 1128 Event. *Nature* 478, 369–373. <https://doi.org/10.1038/nature10511>
- 1129 Konhauser, K.O., Planavsky, N.J., Hardisty, D.S., Robbins, L.J., Warchola, T.J., Haugaard, R.,
 1130 Lalonde, S.V., Partin, C.A., Oonk, P.B.H., Tsikos, H., Lyons, T.W., Bekker, A.,
 1131 Johnson, C.M., 2017. Iron formations: A global record of Neoproterozoic to
 1132 Palaeoproterozoic environmental history. *Earth-Science Reviews* 172, 140–177.
 1133 <https://doi.org/10.1016/j.earscirev.2017.06.012>
- 1134 Kopp, R.E., Kirschvink, J.L., Hilburn, I.A., Nash, C.Z., 2005. The Paleoproterozoic snowball
 1135 Earth: A climate disaster triggered by the evolution of oxygenic photosynthesis. *PNAS*
 1136 102, 11131–11136. <https://doi.org/10.1073/pnas.0504878102>
- 1137 Kreitsmann, T., Külaviir, M., Lepland, A., Paiste, K., Paiste, P., Prave, A.R., Sepp, H.,
 1138 Romashkin, A.E., Rychanchik, D.V., Kirsimäe, K., 2019. Hydrothermal
 1139 dedolomitisation of carbonate rocks of the Paleoproterozoic Zaonega Formation, NW
 1140 Russia — Implications for the preservation of primary C isotope signals. *Chemical*
 1141 *Geology* 512, 43–57. <https://doi.org/10.1016/j.chemgeo.2019.03.002>
- 1142 Krupenik, V.A., Akhmedov, A.M., Sveshnikova, K.Y., 2011. Isotopic composition of carbon,
 1143 oxygen and sulfur in the Ludicovian and Jatulian rocks, in: Glushanin, L.V., Sharov,
 1144 N.V., Shchiptsov, V.V. (Eds.), *The Onega Paleoproterozoic Structure (Geology,
 1145 Tectonics, Deep Structure, Minerogeny)*. Institute of Geology, Karelian Research
 1146 Centre RAS, Petrozavodsk, pp. 250–255.
- 1147 Kump, L.R., 2011. Isotopic Evidence for Massive Oxidation of Organic Matter Following the
 1148 Great Oxidation Event. *Science* 334, 1694–1696.
 1149 <https://doi.org/10.1126/science.1213999>
- 1150 Kump, L.R., Garrels, R.M., 1986. Modeling atmospheric O₂ in the global sedimentary redox
 1151 cycle. *American Journal of Science* 286, 337–360.
 1152 <https://doi.org/10.2475/ajs.286.5.337>
- 1153 Lahtinen, R., Garde, A.A., Melezhik, V.A., 2008. Paleoproterozoic Evolution of Fennoscandia
 1154 and Greenland. *Episodes* 31, 20–28.
- 1155 Large, R.R., Halpin, J.A., Danyushevsky, L.V., Maslennikov, V.V., Bull, S.W., Long, J.A.,
 1156 Gregory, D.D., Lounejeva, E., Lyons, T.W., Sack, P.J., McGoldrick, P.J., Calver, C.R.,
 1157 2014. Trace element content of sedimentary pyrite as a new proxy for deep-time ocean-
 1158 atmosphere evolution. *Earth Planet Sc Lett* 389, 209–220.
 1159 <https://doi.org/10.1016/j.epsl.2013.12.020>
- 1160 Large, R.R., Mukherjee, I., Gregory, D., Steadman, J., Corkrey, R., Danyushevsky, L.V., 2019.
 1161 Atmosphere oxygen cycling through the Proterozoic and Phanerozoic. *Mineralium*
 1162 *Deposita* 54, 485–506. <https://doi.org/10.1007/s00126-019-00873-9>
- 1163 Lepland, A., Joosu, L., Kirsimäe, K., Prave, A.R., Romashkin, A.E., Crne, A.E., Martin, A.P.,
 1164 Fallick, A.E., Somelar, P., Upraus, K., Mand, K., Roberts, N.M.W., van Zuilen, M.A.,
 1165 Wirth, R., Schreiber, A., 2014. Potential influence of sulphur bacteria on
 1166 Palaeoproterozoic phosphogenesis. *Nat Geosci* 7, 20–24.
 1167 <https://doi.org/10.1038/Ngeo2005>
- 1168 Lepland, A., Melezhik, V.A., Papineau, A.E., Romashkin, A.E., Joosu, L., 2013. The Earliest
 1169 Phosphorites – Radical Change in the Phosphorus Cycle during the Palaeoproterozoic,
 1170 in: Melezhik, V.A., Prave, A.R., Fallick, A.E., Kump, L.R., Strauss, H., Lepland, A.,
 1171 Hanski, E. (Eds.), *Reading the Archive of Earth’s Oxygenation: Volume 3: Global
 1172 Events and the Fennoscandian Arctic Russia - Drilling Early Earth Project*, *Frontiers in*
 1173 *Earth Sciences*. Springer, pp. 1275–1296.

- 1174 Lewan, M.D., 1983. Effects of thermal maturation on stable organic carbon isotopes as
 1175 determined by hydrous pyrolysis of Woodford Shale. *Geochimica et Cosmochimica*
 1176 *Acta* 47, 1471–1479. [https://doi.org/10.1016/0016-7037\(83\)90306-X](https://doi.org/10.1016/0016-7037(83)90306-X)
- 1177 Lowenstein, T.K., Hardie, L.A., Timofeeff, M.N., Demicco, R.V., 2003. Secular variation in
 1178 seawater chemistry and the origin of calcium chloride basinal brines. *Geology* 31, 857.
 1179 <https://doi.org/10.1130/G19728R.1>
- 1180 Luo, G.M., Ono, S.H., Beukes, N.J., Wang, D.T., Xie, S.C., Summons, R.E., 2016. Rapid
 1181 oxygenation of Earth's atmosphere 2.33 billion years ago. *Sci Adv* 2.
 1182 <https://doi.org/UNSP e1600134> 10.1126/sciadv.1600134
- 1183 Luo, G.M., Ono, S.H., Huang, J.H., Algeo, T.J., Li, C., Zhou, L., Robinson, A., Lyons, T.W.,
 1184 Xie, S.C., 2015. Decline in oceanic sulfate levels during the early Mesoproterozoic.
 1185 *Precambrian Res* 258, 36–47. <https://doi.org/10.1016/j.precamres.2014.12.014>
- 1186 Luque, F.J., Pasteris, J.D., Wopenka, B., Rodas, M., Barrenechea, J.F., 1998. Natural fluid-
 1187 deposited graphite; mineralogical characteristics and mechanisms of formation.
 1188 *American Journal of Science* 298, 471–498. <https://doi.org/10.2475/ajs.298.6.471>
- 1189 Lyons, T.W., 1997. Sulfur isotopic trends and pathways of iron sulfide formation in upper
 1190 Holocene sediments of the anoxic Black Sea. *Geochimica et Cosmochimica Acta* 61,
 1191 3367–3382. [https://doi.org/Doi 10.1016/S0016-7037\(97\)00174-9](https://doi.org/Doi 10.1016/S0016-7037(97)00174-9)
- 1192 Lyons, T.W., Reinhard, C.T., Planavsky, N.J., 2014. The rise of oxygen in Earth's early ocean
 1193 and atmosphere. *Nature* 506, 307–315. <https://doi.org/10.1038/nature13068>
- 1194 Mänd, K., Lalonde, S.V., Robbins, L.J., Thoby, M., Paiste, K., Kreitsmann, T., Paiste, P.,
 1195 Reinhard, C.T., Romashkin, A.E., Planavsky, N.J., Kirsimäe, K., Lepland, A.,
 1196 Konhauser, K.O., 2020. Palaeoproterozoic oxygenated oceans following the
 1197 Lomagundi–Jatuli Event. *Nature Geoscience*. [https://doi.org/10.1038/s41561-020-](https://doi.org/10.1038/s41561-020-0558-5)
 1198 0558-5
- 1199 Marais, D.J.D., 2001. Isotopic Evolution of the Biogeochemical Carbon Cycle During the
 1200 Precambrian. *Reviews in Mineralogy and Geochemistry* 43, 555–578.
 1201 <https://doi.org/10.2138/gsrmg.43.1.555>
- 1202 Martin, Adam P., Condon, D.J., Prave, A.R., Lepland, A., 2013. A review of temporal
 1203 constraints for the Palaeoproterozoic large, positive carbonate carbon isotope excursion
 1204 (the Lomagundi–Jatuli Event). *Earth-Science Reviews* 127, 242–261.
 1205 <https://doi.org/10.1016/j.earscirev.2013.10.006>
- 1206 Martin, A.P., Condon, D.J., Prave, A.R., Melezhik, V.A., Lepland, A., Fallick, A.E., 2013.
 1207 Dating the termination of the Palaeoproterozoic Lomagundi–Jatuli carbon isotopic event
 1208 in the North Transfennoscandian Greenstone Belt. *Precambrian Research* 224, 160–168.
 1209 <https://doi.org/10.1016/j.precamres.2012.09.010>
- 1210 Martin, A.P., Prave, A.R., Condon, D.J., Lepland, A., Fallick, A.E., Romashkin, A.E.,
 1211 Medvedev, P.V., Rychanchik, D.V., 2015. Multiple Palaeoproterozoic carbon burial
 1212 episodes and excursions. *Earth Planet Sc Lett* 424, 226–236.
 1213 <https://doi.org/10.1016/j.epsl.2015.05.023>
- 1214 Melezhik, V.A., Fallick, A.E., Brasier, A.T., Lepland, A., 2015. Carbonate deposition in the
 1215 Palaeoproterozoic Onega basin from Fennoscandia: a spotlight on the transition from
 1216 the Lomagundi–Jatuli to Shunga events. *Earth-Sci Rev* 147, 65–98.
 1217 <https://doi.org/10.1016/j.earscirev.2015.05.005>
- 1218 Melezhik, V.A., Fallick, A.E., Filippov, M.M., Deines, Y.E., Črne, A.E., Lepland, A., Brasier,
 1219 A.T., Strauss, H., 2013. Giant Palaeoproterozoic Petrified Oil Field in the Onega Basin,
 1220 in: *Reading the Archive of Earth's Oxygenation: Volume 3: Global Events and the*
 1221 *Fennoscandian Arctic Russia - Drilling Early Earth Project*, *Frontiers in Earth Sciences*.
 1222 Springer, pp. 1202–1212.

- 1223 Melezhik, V.A., Fallick, A.E., Filippov, M.M., Larsen, O., 1999. Karelian shungite—an
 1224 indication of 2.0-Ga-old metamorphosed oil-shale and generation of petroleum:
 1225 geology, lithology and geochemistry. *Earth-Science Reviews* 47, 1–40.
 1226 [https://doi.org/10.1016/S0012-8252\(99\)00027-6](https://doi.org/10.1016/S0012-8252(99)00027-6)
- 1227 Melezhik, V.A., Filippov, M.M., Romashkin, A.E., 2004. A giant Palaeoproterozoic deposit of
 1228 shungite in NW Russia: genesis and practical applications. *Ore Geol Rev* 24, 135–154.
 1229 <https://doi.org/DOI.10.1016/j.oregeorev.2003.08.003>
- 1230 Melezhik, V.A., Hanski, E.J., 2013. Palaeotectonic and Palaeogeographic Evolution of
 1231 Fennoscandia in the Early Palaeoproterozoic, in: *Reading the Archive of Earth's*
 1232 *Oxygenation*, *Frontiers in Earth Sciences*. Springer, Berlin, Heidelberg, pp. 111–178.
 1233 https://doi.org/10.1007/978-3-642-29682-6_5
- 1234 Melezhik, V.A., Huhma, H., Condon, D.J., Fallick, A.E., Whitehouse, M.J., 2007. Temporal
 1235 constraints on the Paleoproterozoic Lomagundi-Jatuli carbon isotopic event. *Geology*
 1236 35, 655–658. <https://doi.org/10.1130/G23764A.1>
- 1237 Melezhik, V.A., Sturt, B.A., 1994. General geology and evolutionary history of the early
 1238 proterozoic Polmak-Pasvik-Pechenga-Imandra/Varzuga-Ust'Ponoy greenstone belt in
 1239 the northeastern Baltic Shield. *Earth-Science Reviews* 36, 205–241.
 1240 [https://doi.org/10.1016/0012-8252\(94\)90058-2](https://doi.org/10.1016/0012-8252(94)90058-2)
- 1241 Meyer, K.M., Kump, L.R., 2008. Oceanic Euxinia in Earth History: Causes and Consequences.
 1242 *Annu. Rev. Earth Planet. Sci.* 36, 251–288.
 1243 <https://doi.org/10.1146/annurev.earth.36.031207.124256>
- 1244 Meyer, N.R., Zerkle, A.L., Fike, D.A., 2017. Sulphur cycling in a Neoproterozoic microbial mat.
 1245 *Geobiology* 15, 353–365. <https://doi.org/10.1111/gbi.12227>
- 1246 Morozov, A.F., Hakhaev, B.N., Petrov, O.V., Gorbachev, V.I., Tarkhanov, G.B., Tsvetkov,
 1247 L.D., Erinchek, Y.M., Akhmedov, A.M., Krupenik, V.A., Sveshnikova, K.Y., 2010.
 1248 Rock-salts in Palaeoproterozoic strata of the Onega depression of Karelia (based on data
 1249 from the Onega parametric drillhole). *Transection of Academy of Sciences* 435, 230–
 1250 233.
- 1251 Mukherjee, I., Large, R.R., 2016. Pyrite trace element chemistry of the Velkerri Formation,
 1252 Roper Group, McArthur Basin: Evidence for atmospheric oxygenation during the
 1253 Boring Billion. *Precambrian Research* 281, 13–26.
 1254 <https://doi.org/10.1016/j.precamres.2016.05.003>
- 1255 Ohmoto, H., Watanabe, Y., Lasaga, A.C., Naraoka, H., Johnson, I., Brainard, J., Chorney, A.,
 1256 2014. Oxygen, iron, and sulfur geochemical cycles on early Earth: Paradigms and
 1257 contradictions, in: *Earth's Early Atmosphere and Surface Environment*. Geological
 1258 Society of America. [https://doi.org/10.1130/2014.2504\(09\)](https://doi.org/10.1130/2014.2504(09))
- 1259 Ojakangas, R.W., Marmo, J.S., Heiskanen, K.I., 2001. Basin evolution of the Paleoproterozoic
 1260 Karelian Supergroup of the Fennoscandian (Baltic) Shield. *Sediment Geol* 141, 255–
 1261 285. [https://doi.org/Doi.10.1016/S0037-0738\(01\)00079-3](https://doi.org/Doi.10.1016/S0037-0738(01)00079-3)
- 1262 Ono, S., Beukes, N.J., Rumble, D., 2009. Origin of two distinct multiple-sulfur isotope
 1263 compositions of pyrite in the 2.5Ga Klein Naute Formation, Griqualand West Basin,
 1264 South Africa. *Precambrian Research* 169, 48–57.
 1265 <https://doi.org/10.1016/j.precamres.2008.10.012>
- 1266 Ossa Ossa, F., Eickmann, B., Hofmann, A., Planavsky, N.J., Asael, D., Pambo, F., Bekker, A.,
 1267 2018. Two-step deoxygenation at the end of the Paleoproterozoic Lomagundi Event.
 1268 *Earth and Planetary Science Letters* 486, 70–83.
 1269 <https://doi.org/10.1016/j.epsl.2018.01.009>
- 1270 Ovchinnikova, G.V., Kuznetsov, A.B., Melezhik, V.A., Gorokhov, I.M., Vasil'eva, I.M.,
 1271 Gorokhovskii, B.M., 2007. Pb-Pb age of Jatulian carbonate rocks: The Tulomozero

1272 Formation of southeast Karelia. *Stratigr. Geol. Correl.* 15, 359–372.
1273 <https://doi.org/10.1134/S0869593807040028>

1274 Paiste, K., Lepland, A., Zerkle, A.L., Kirsimäe, K., Izon, G., Patel, N.K., McLean, F.,
1275 Kreitsmann, T., Mänd, K., Bui, T.H., Romashkin, A.E., Rychanchik, D.V., Prave, A.R.,
1276 2018. Multiple sulphur isotope records tracking basinal and global processes in the 1.98
1277 Ga Zaonega Formation, NW Russia. *Chemical Geology* 499, 151–164.
1278 <https://doi.org/10.1016/j.chemgeo.2018.09.025>

1279 Paiste, K., Pellerin, A., Zerkle, A.L., Kirsimäe, K., Prave, A.R., Romashkin, A.E., Lepland, A.,
1280 2020. The pyrite multiple sulfur isotope record of the 1.98 Ga Zaonega Formation:
1281 Evidence for biogeochemical sulfur cycling in a semi-restricted basin. *Earth and*
1282 *Planetary Science Letters* 534, 116092. <https://doi.org/10.1016/j.epsl.2020.116092>

1283 Papineau, D., 2010. Global biogeochemical changes at both ends of the proterozoic: insights
1284 from phosphorites. *Astrobiology* 10, 165–181. <https://doi.org/10.1089/ast.2009.0360>

1285 Partin, C.A., Bekker, A., Planavsky, N.J., Scott, C.T., Gill, B.C., Li, C., Podkovyrov, V.,
1286 Maslov, A., Konhauser, K.O., Lalonde, S.V., Love, G.D., Poulton, S.W., Lyons, T.W.,
1287 2013. Large-scale fluctuations in Precambrian atmospheric and oceanic oxygen levels
1288 from the record of U in shales. *Earth Planet Sc Lett* 369, 284–293.
1289 <https://doi.org/10.1016/j.epsl.2013.03.031>

1290 Pasquier, V., Sansjofre, P., Rabineau, M., Revillon, S., Houghton, J., Fike, D.A., 2017. Pyrite
1291 sulfur isotopes reveal glacial–interglacial environmental changes. *Proceedings of the*
1292 *National Academy of Sciences* 114, 5941–5945.
1293 <https://doi.org/10.1073/pnas.1618245114>

1294 Perttunen, V., Vaasjoki, M., 2001. U–Pb geochronology of the Peräpohja Schist Belt,
1295 northwestern Finland. *Geological Survey of Finland, Special Paper* 33 45–84.

1296 Picard, A., Gartman, A., Clarke, D.R., Girguis, P.R., 2018. Sulfate-reducing bacteria influence
1297 the nucleation and growth of mackinawite and greigite. *Geochimica et Cosmochimica*
1298 *Acta* 220, 367–384. <https://doi.org/10.1016/j.gca.2017.10.006>

1299 Planavsky, N.J., Bekker, A., Hofmann, A., Owens, J.D., Lyons, T.W., 2012. Sulfur record of
1300 rising and falling marine oxygen and sulfate levels during the Lomagundi event. *P Natl*
1301 *Acad Sci USA* 109, 18300–18305. <https://doi.org/10.1073/pnas.1120387109>

1302 Planavsky, N.J., Slack, J.F., Cannon, W.F., O’Connell, B., Isson, T.T., Asael, D., Jackson, J.C.,
1303 Hardisty, D.S., Lyons, T.W., Bekker, A., 2018. Evidence for episodic oxygenation in a
1304 weakly redox-buffered deep mid-Proterozoic ocean. *Chemical Geology* 483, 581–594.
1305 <https://doi.org/10.1016/j.chemgeo.2018.03.028>

1306 Poulton, S.W., Canfield, D.E., 2005. Development of a sequential extraction procedure for iron:
1307 implications for iron partitioning in continentally derived particulates. *Chemical*
1308 *Geology* 214, 209–221. <https://doi.org/10.1016/j.chemgeo.2004.09.003>

1309 Poulton, S.W., Canfield, D.E., 2011. Ferruginous Conditions: A Dominant Feature of the Ocean
1310 through Earth’s History. *Elements* 7, 107–112.
1311 <https://doi.org/10.2113/gselements.7.2.107>

1312 Poulton, S.W., Fralick, P.W., Canfield, D.E., 2010. Spatial variability in oceanic redox structure
1313 1.8 billion years ago. *Nature Geoscience* 3, 486–490. <https://doi.org/10.1038/ngeo889>

1314 Poulton, S.W., Fralick, P.W., Canfield, D.E., 2004. The transition to a sulphidic ocean [similar]
1315 1.84 billion years ago [WWW Document]. *Nature*. URL
1316 <http://link.galegroup.com/apps/doc/A186294134/AONE?sid=googlescholar> (accessed
1317 9.8.18).

1318 Preuß, A., Schauder, R., Fuchs, G., 1989. Carbon Isotope Fractionation by Autotrophic Bacteria
1319 with Three Different CO₂ Fixation Pathways. *Zeitschrift für Naturforschung* 44c, 397–
1320 402.

- 1321 Priyatkina, N., Khudoley, A.K., Ustinov, V.N., Kullerud, K., 2014. 1.92 Ga kimberlitic rocks
 1322 from Kimozero, NW Russia: Their geochemistry, tectonic setting and unusual field
 1323 occurrence. *Precambrian Res* 249, 162–179.
 1324 <https://doi.org/10.1016/j.precamres.2014.05.009>
- 1325 Puchtel, I.S., Arndt, N.T., Hofmann, A.W., Haase, K.M., Kroner, A., Kulikov, V.S., Kulikova,
 1326 V.V., Garbe-Schonberg, C.D., Nemchin, A.A., 1998. Petrology of mafic lavas within
 1327 the Onega plateau, central Karelia: evidence for 2.0 Ga plume-related continental crustal
 1328 growth in the Baltic Shield. *Contributions to Mineralogy and Petrology* 130, 134–153.
 1329 <https://doi.org/10.1007/s004100050355>
- 1330 Puchtel, I.S., Brugmann, G.E., Hofmann, A.W., 1999. Precise Re-Os mineral isochron and Pb-
 1331 Nd-Os isotope systematics of a mafic-ultramafic sill in the 2.0 Ga Onega plateau (Baltic
 1332 Shield). *Earth Planet Sc Lett* 170, 447–461. [https://doi.org/10.1016/s0012-821x\(99\)00118-1](https://doi.org/10.1016/s0012-821x(99)00118-1)
- 1334 Pufahl, P.K., Hiatt, E.E., Kyser, T.K., 2010. Does the Paleoproterozoic Animikie Basin record
 1335 the sulfidic ocean transition? *Geology* 38, 659–662. <https://doi.org/10.1130/G30747.1>
- 1336 Qu, Y., Crne, A.E., Lepland, A., Van Zuilen, M.A., 2012. Methanotrophy in a Paleoproterozoic
 1337 oil field ecosystem, Zaonega Formation, Karelia, Russia. *Geobiology* 10, 467–478.
 1338 <https://doi.org/Doi.10.1111/Gbi.12007>
- 1339 Qu, Y., Lepland, A., van Zuilen, M.A., Whitehouse, M., Črne, A.E., Fallick, A.E., 2018.
 1340 Sample-scale carbon isotopic variability and diverse biomass in the Paleoproterozoic
 1341 Zaonega Formation, Russia. *Precambrian Research* 315, 222–231.
 1342 <https://doi.org/10.1016/j.precamres.2018.07.008>
- 1343 Raiswell, R., Canfield, D.E., 2012. The Iron Biogeochemical Cycle Past and Present. *Geochem*
 1344 *Perspect* 1, 1–220. <https://doi.org/10.7185/geochempersp.1.1>
- 1345 Raiswell, R., Canfield, D.E., Berner, R.A., 1994. A Comparison of Iron Extraction Methods for
 1346 the Determination of Degree of Pyritisation and the Recognition of Iron-Limited Pyrite
 1347 Formation. *Chemical Geology* 111, 101–110. [https://doi.org/Doi.10.1016/0009-2541\(94\)90084-1](https://doi.org/Doi.10.1016/0009-2541(94)90084-1)
- 1349 Rau, G., 1978. Carbon-13 Depletion in a Subalpine Lake: Carbon Flow Implications. *Science*
 1350 201, 901–902. <https://doi.org/10.1126/science.201.4359.901>
- 1351 Raven, M.R., Fike, D.A., Gomes, M.L., Webb, S.M., Bradley, A.S., McClelland, H.-L.O.,
 1352 2018. Organic carbon burial during OAE2 driven by changes in the locus of organic
 1353 matter sulfurization. *Nature Communications* 9, 3409. <https://doi.org/10.1038/s41467-018-05943-6>
- 1355 Raven, M.R., Sessions, A.L., Adkins, J.F., Thunell, R.C., 2016. Rapid organic matter
 1356 sulfurization in sinking particles from the Cariaco Basin water column. *Geochimica et*
 1357 *Cosmochimica Acta* 190, 175–190. <https://doi.org/10.1016/j.gca.2016.06.030>
- 1358 Reddy, S.M., Evans, D.A.D., 2009. Palaeoproterozoic supercontinents and global evolution:
 1359 correlations from core to atmosphere: Fig. 1. Geological Society, London, Special
 1360 Publications 323, 1–26. <https://doi.org/10.1144/SP323.1>
- 1361 Rees, C.E., Jenkins, W.J., Monster, J., 1978. The sulphur isotope geochemistry of ocean water
 1362 sulphate. *Geochimica et Cosmochimica Acta* 42, 377–381.
 1363 [https://doi.org/10.1016/0016-7037\(78\)90268-5](https://doi.org/10.1016/0016-7037(78)90268-5)
- 1364 Reuschel, M., Melezhik, V.A., Strauss, H., 2012a. Sulfur isotopic trends and iron speciation
 1365 from the c. 2.0 Ga Pilgijärvi Sedimentary Formation, NW Russia. *Precambrian*
 1366 *Research* 196–197, 193–203. <https://doi.org/10.1016/j.precamres.2011.12.009>
- 1367 Reuschel, M., Melezhik, V.A., Whitehouse, M.J., Lepland, A., Fallick, A.E., Strauss, H.,
 1368 2012b. Isotopic evidence for a sizeable seawater sulfate reservoir at 2.1 Ga. *Precambrian*
 1369 *Res* 192–95, 78–88. <https://doi.org/DOI.10.1016/j.precamres.2011.10.013>

- 1370 Rico, K.I., Sheldon, N.D., 2019. Nutrient and iron cycling in a modern analogue for the
1371 redoxcline of a Proterozoic ocean shelf. *Chemical Geology*.
1372 <https://doi.org/10.1016/j.chemgeo.2019.02.032>
- 1373 Ries, J.B., Fike, D.A., Pratt, L.M., Lyons, T.W., Grotzinger, J.P., 2009. Superheavy pyrite (
1374 $^{34}\text{S}_{\text{pyr}} > ^{34}\text{S}_{\text{SCAS}}$) in the terminal Proterozoic Nama Group, southern Namibia: A
1375 consequence of low seawater sulfate at the dawn of animal life. *Geology* 37, 743–746.
1376 <https://doi.org/10.1130/G25775A.1>
- 1377 Sackett, W.M., 1978. Carbon and hydrogen isotope effects during thermo-catalytic production
1378 of hydrocarbons in laboratory simulation experiments. *Geochimica et Cosmochimica*
1379 *Acta* 42, 571–580.
- 1380 Sawaki, Y., Moussavou, M., Sato, T., Suzuki, K., Ligna, C., Asanuma, H., Sakata, S., Obayashi,
1381 H., Hirata, T., Edou-Minko, A., 2017. Chronological constraints on the
1382 Paleoproterozoic Francevillian Group in Gabon. *Geoscience Frontiers* 8, 397–407.
1383 <https://doi.org/10.1016/j.gsf.2016.10.001>
- 1384 Schidlowski, M., 2001. Carbon isotopes as biogeochemical recorders of life over 3.8 Ga of
1385 Earth history: evolution of a concept. *Precambrian Research* 106, 117–134.
1386 [https://doi.org/10.1016/S0301-9268\(00\)00128-5](https://doi.org/10.1016/S0301-9268(00)00128-5)
- 1387 Schoell, M., 1983. Genetic-characterization of natural gases. *American Association of*
1388 *Petroleum Geologists* 67, 546.
- 1389 Scholz, F., 2018. Identifying oxygen minimum zone-type biogeochemical cycling in Earth
1390 history using inorganic geochemical proxies. *Earth-Science Reviews* 184, 29–45.
1391 <https://doi.org/10.1016/j.earscirev.2018.08.002>
- 1392 Scholz, F., Schmidt, M., Hensen, C., Eroglu, S., Geilert, S., Gutjahr, M., Liebetrau, V., 2019.
1393 Shelf-to-basin iron shuttle in the Guaymas Basin, Gulf of California. *Geochimica et*
1394 *Cosmochimica Acta* 261, 76–92. <https://doi.org/10.1016/j.gca.2019.07.006>
- 1395 Schröder, S., Bekker, A., Beukes, N.J., Strauss, H., Niekerk, H.S.V., 2008. Rise in seawater
1396 sulphate concentration associated with the Paleoproterozoic positive carbon isotope
1397 excursion: evidence from sulphate evaporites in the ~2.2–2.1 Gyr shallow-marine
1398 Lucknow Formation, South Africa. *Terra Nova* 20, 108–117.
1399 <https://doi.org/10.1111/j.1365-3121.2008.00795.x>
- 1400 Scott, C., Wing, B.A., Bekker, A., Planavsky, N.J., Medvedev, P., Bates, S.M., Yun, M., Lyons,
1401 T.W., 2014. Pyrite multiple-sulfur isotope evidence for rapid expansion and contraction
1402 of the early Paleoproterozoic seawater sulfate reservoir. *Earth Planet Sc Lett* 389, 95–
1403 104. <https://doi.org/DOI 10.1016/j.epsl.2013.12.010>
- 1404 Seal, R.R., 2006. Sulfur isotope geochemistry of sulfide minerals. *Rev Mineral Geochem* 61,
1405 633–677. <https://doi.org/DOI 10.2138/rmg.2006.61.12>
- 1406 Sheen, A.I., Kendall, B., Reinhard, C.T., Creaser, R.A., Lyons, T.W., Bekker, A., Poulton,
1407 S.W., Anbar, A.D., 2018. A model for the oceanic mass balance of rhenium and
1408 implications for the extent of Proterozoic ocean anoxia. *Geochimica et Cosmochimica*
1409 *Acta* 227, 75–95. <https://doi.org/10.1016/j.gca.2018.01.036>
- 1410 Sim, M.S., Bosak, T., Ono, S., 2011. Large Sulfur Isotope Fractionation Does Not Require
1411 Disproportionation. *Science* 333, 74–77. <https://doi.org/10.1126/science.1205103>
- 1412 Slotznick, S.P., Eiler, J.M., Fischer, W.W., 2018. The effects of metamorphism on iron
1413 mineralogy and the iron speciation redox proxy. *Geochimica et Cosmochimica Acta*
1414 224, 96–115. <https://doi.org/10.1016/j.gca.2017.12.003>
- 1415 Sperling, E.A., Rooney, A.D., Hays, L., Sergeev, V.N., Vorob'eva, N.G., Sergeeva, N.D.,
1416 Selby, D., Johnston, D.T., Knoll, A.H., 2014. Redox heterogeneity of subsurface waters
1417 in the Mesoproterozoic ocean. *Geobiology* 12, 373–386.
1418 <https://doi.org/10.1111/gbi.12091>

- 1419 Sperling, E.A., Wolock, C.J., Morgan, A.S., Gill, B.C., Kunzmann, M., Halverson, G.P.,
 1420 Macdonald, F.A., Knoll, A.H., Johnston, D.T., 2015. Statistical analysis of iron
 1421 geochemical data suggests limited late Proterozoic oxygenation. *Nature* 523, 451–454.
 1422 <https://doi.org/10.1038/nature14589>
- 1423 Stepanova, A.V., Samsonov, A.V., Larionov, A.N., 2014. The final episode of middle
 1424 Proterozoic magmatism in the Onega structure: data on trans-Onega dolerites.
 1425 *Proceedings of the Karelian Research Centre of the Russian Academy of Sciences* 1,
 1426 3–16.
- 1427 Strauss, H., Melezhik, V.A., Lepland, A., Fallick, A.E., Hanski, E.J., Filippov, M.M., Deines,
 1428 Y.E., Illing, C.J., Črne, A.E., Brasier, A.T., 2013. Enhanced Accumulation of Organic
 1429 Matter: The Shunga Event, in: Melezhik, V.A., Prave, A.R., Hanski, E.J., Fallick, A.E.,
 1430 Lepland, A., Kump, L.R., Strauss, H. (Eds.), *Reading the Archive of Earth's*
 1431 *Oxygenation*. Springer Berlin Heidelberg, Berlin, Heidelberg, pp. 1195–1273.
 1432 https://doi.org/10.1007/978-3-642-29670-3_6
- 1433 van Breugel, Y., Schouten, S., Paetzel, M., Nordeide, R., Sinninghe Damsté, J.S., 2005. The
 1434 impact of recycling of organic carbon on the stable carbon isotopic composition of
 1435 dissolved inorganic carbon in a stratified marine system (Kyllaren fjord, Norway).
 1436 *Organic Geochemistry* 36, 1163–1173.
 1437 <https://doi.org/10.1016/j.orggeochem.2005.03.003>
- 1438 Wagner, T., Boyce, A.J., 2006. Pyrite metamorphism in the Devonian Hunsrück Slate of
 1439 Germany: Insights from laser microprobe sulfur isotope analysis and thermodynamic
 1440 modeling. *Am J Sci* 306, 525–552. <https://doi.org/10.2475/07.2006.02>
- 1441 Waldron, H.N., Probyn, T.A., 1991. Short-term variability during an anchor station study in the
 1442 southern Benguela upwelling system: Nitrogen supply to the euphotic zone during a
 1443 quiescent phase in the upwelling cycle. *Progress in Oceanography* 28, 153–166.
 1444 [https://doi.org/10.1016/0079-6611\(91\)90024-G](https://doi.org/10.1016/0079-6611(91)90024-G)
- 1445 Weber, F., Gauthier-Lafaye, F., 2013. No proof from carbon isotopes in the Francevillian
 1446 (Gabon) and Onega (Fennoscandian shield) basins of a global oxidation event at 1980–
 1447 2090 Ma following the Great Oxidation Event (GOE). *Comptes Rendus Geoscience*
 1448 345, 28–35. <https://doi.org/10.1016/j.crte.2012.12.003>
- 1449 Weber, F., Gauthier-Lafaye, F., Whitechurch, H., Ulrich, M., El Albani, A., 2016. The 2-Ga
 1450 Eburnean Orogeny in Gabon and the opening of the Francevillian intracratonic basins:
 1451 A review. *Comptes Rendus Geoscience* 348, 572–586.
 1452 <https://doi.org/10.1016/j.crte.2016.07.003>
- 1453 Werne, J.P., Lyons, T.W., Hollander, D.J., Formolo, M.J., Sinninghe Damsté, J.S., 2003.
 1454 Reduced sulfur in euxinic sediments of the Cariaco Basin: sulfur isotope constraints on
 1455 organic sulfur formation. *Chemical Geology, Isotopic records of microbially mediated*
 1456 *processes* 195, 159–179. [https://doi.org/10.1016/S0009-2541\(02\)00393-5](https://doi.org/10.1016/S0009-2541(02)00393-5)
- 1457 Whiticar, M.J., 1999. Carbon and hydrogen isotope systematics of bacterial formation and
 1458 oxidation of methane. *Chemical Geology* 161, 291–314. [https://doi.org/10.1016/S0009-2541\(99\)00092-3](https://doi.org/10.1016/S0009-2541(99)00092-3)
- 1460 Wing, B.A., Halevy, I., 2014. Intracellular metabolite levels shape sulfur isotope fractionation
 1461 during microbial sulfate respiration. *Proceedings of the National Academy of Sciences*
 1462 111, 18116–18125. <https://doi.org/10.1073/pnas.1407502111>
- 1463 Woodcock, N.H., 2004. Life span and fate of basins. *Geology* 32, 685–688.
 1464 <https://doi.org/10.1130/G20598.1>
- 1465 Xiao, S., Schiffbauer, J.D., McFadden, K.A., Hunter, J., 2010. Petrographic and SIMS pyrite
 1466 sulfur isotope analyses of Ediacaran chert nodules: Implications for microbial processes
 1467 in pyrite rim formation, silicification, and exceptional fossil preservation. *Earth and*
 1468 *Planetary Science Letters* 297, 481–495. <https://doi.org/10.1016/j.epsl.2010.07.001>

- 1469 Young, G.M., Brunn, V. von, Gold, D.J.C., Minter, W.E.L., 1998. Earth'S Oldest Reported
 1470 Glaciation: Physical and Chemical Evidence From the Archean Mozaan Group (~2.9
 1471 Ga) of South Africa. *The Journal of Geology* 106, 523–538.
 1472 <https://doi.org/10.1086/516039>
- 1473 Yudovich, Y., Makarikhin, V., Medvedev, P., Sukhanov, N., 1991. Carbon isotope anomalies
 1474 in carbonates of the Karelian Complex. *Carbon isotope anomalies in carbonates of the*
 1475 *Karelian Complex* 28, 56–62.
- 1476 Zaback, D.A., Pratt, L.M., Hayes, J.M., 1993. Transport and reduction of sulfate and
 1477 immobilization of sulfide in marine black shales. *Geology* 21, 141–144.
 1478 [https://doi.org/10.1130/0091-7613\(1993\)021<0141:TAROSA>2.3.CO;2](https://doi.org/10.1130/0091-7613(1993)021<0141:TAROSA>2.3.CO;2)
- 1479 Zerkle, A.L., Farquhar, J., Johnston, D.T., Cox, R.P., Canfield, D.E., 2009. Fractionation of
 1480 multiple sulfur isotopes during phototrophic oxidation of sulfide and elemental sulfur
 1481 by a green sulfur bacterium. *Geochimica et Cosmochimica Acta* 73, 291–306.
 1482 <https://doi.org/10.1016/j.gca.2008.10.027>
- 1483 Zerkle, A.L., House, C.H., Brantley, S.L., 2005. Biogeochemical signatures through time as
 1484 inferred from whole microbial genomes. *Am J Sci* 305, 467–502. <https://doi.org/DOI>
 1485 10.2475/ajs.305.6-8.467
- 1486 Zerkle, A.L., Jones, D.S., Farquhar, J., Macalady, J.L., 2016. Sulfur isotope values in the
 1487 sulfidic Frasassi cave system, central Italy: A case study of a chemolithotrophic S-based
 1488 ecosystem. *Geochimica et Cosmochimica Acta* 173, 373–386.
 1489 <https://doi.org/10.1016/j.gca.2015.10.028>
- 1490 Zerkle, A.L., Kamyshny, A., Kump, L.R., Farquhar, J., Oduro, H., Arthur, M.A., 2010. Sulfur
 1491 cycling in a stratified euxinic lake with moderately high sulfate: Constraints from
 1492 quadruple S isotopes. *Geochimica et Cosmochimica Acta* 74, 4953–4970.
 1493 <https://doi.org/10.1016/j.gca.2010.06.015>
- 1494 Zerkle, A.L., Poulton, S.W., Newton, R.J., Mettam, C., Claire, M.W., Bekker, A., Junium, C.K.,
 1495 2017. Onset of the aerobic nitrogen cycle during the Great Oxidation Event. *Nature* 542,
 1496 465–467. <https://doi.org/10.1038/nature20826>

1497 **Figure captions**

1498 **Figure 1.** Key geobiological events and trends in Earth history; evolution of atmospheric
 1499 oxygen modified from Lyons et al. (2014) and $\delta^{13}\text{C}_{\text{carb}}$ trend from Farquhar et al. (2014).

1500 **Figure 2.** A. Simplified geological map of the Onega basin, NW Russia. Circles show locations
 1501 of cores discussed in the text. B. World map showing the present distribution of
 1502 Paleoproterozoic rocks, highlighting the locations of the Animikie Basin, Francevillian Basin,
 1503 Onega Basin and Pechenga Greenstone Belt.

1504 **Figure 3.** Lithological profiles for the dolostone marker unit and underlying P-rich mudstone
 1505 used for correlating the OPH, OnZap 1 and 3, and FAR-DEEP 12AB cores across the Onega
 1506 Basin, NW Russia. Photos show the dolostone-mudstone contact and the occurrence of mm- to
 1507 cm-thick discontinuous layers of apatite at the base of the massive grey dolomite bed. Phosphate
 1508 (expressed as P_2O_5) concentration logs displayed to the right of lithological profiles represent
 1509 semi-quantitative determinations undertaken by XRF logging directly on cores (OnZap 1, 3 and

1510 FAR-DEEP 12AB cores) whereas data for the OPH core are from bulk XRF quantitative
1511 determinations on extracted subsamples.

1512 **Figure 4.** Lithostratigraphic profile and new $\delta^{13}\text{C}_{\text{org}}$ data from the OPH core and previously
1513 published sulfur isotope, total organic carbon (TOC) and total sulfur (TS) data for the OPH core
1514 (Paiste et al., 2020). These data are used to divide the OPH core into Members A, B and C (see
1515 text for details). Black bars at the side of the OPH lithological profile represent intervals of core
1516 recovery; interpolation of rock types in zones of no recovery are based on interpretations of
1517 downhole geophysical data (after Morozov et al., 2010).

1518 **Figure 5.** Compilation of $\delta^{13}\text{C}_{\text{org}}$, $\delta^{34}\text{S}$, TS and TOC profiles for six cores from the Zaonega
1519 Formation. The three Members defined in the OPH core (Members A, B and C; see Fig. 4) can
1520 be identified in all the cores (data from Lepland et al., 2014; Melezhik et al., 2015; Qu et al.,
1521 2012; Scott et al., 2014; Strauss et al., 2013). Sedimentary and magmatic intervals are shown
1522 for the OPH, FAR-DEEP (FD) and OnZap core profiles but cores C-5190 and C-175 are left
1523 blank because, although C-S isotope data have been utilised from those cores (e.g. Scott et al.,
1524 2014), there are no lithological descriptions to accompany those data hence the exact
1525 distribution of sedimentary and magmatic bodies is not known.

1526 **Figure 6.** Compilation of sulfide (grey) and sulfate (light blue) $\delta^{34}\text{S}$ data through time (modified
1527 after (Crockford et al., 2019; Fike et al., 2015; Havig et al., 2017) with expanded trends shown
1528 for stratigraphic pyrite $\delta^{34}\text{S}$ profiles from the Francevillian (Canfield et al., 2013; Ossa Ossa et
1529 al., 2018), Onega (Paiste et al., 2020, 2018; Scott et al., 2014), Pechenga (Reuschel et al., 2012a)
1530 and Animikie (Planavsky et al., 2018; Poulton et al., 2004) successions. Note that two different
1531 stratigraphic schemes have been proposed for the Francevillian reference column (Formations
1532 FB, FC and FD), one by Canfield et al., (2013) and the other by Ossa-Ossa et al., (2018), who
1533 use additional subdivisions of Formation FB (units 1a through 2b); both versions are shown.

1534 **Supplementary material**

1535 **Supplementary Table 1.** Organic carbon isotope data for the Onega Parametric drillhole
1536 (OPH).